



XIV School on Synchrotron Radiation:  
*Fundamentals, Methods and Applications*  
Muggia, Italy / 18-29 September 2017



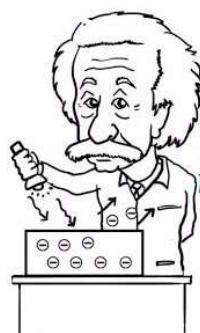
# Photoemission Spectroscopy: Fundamental Aspects

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CNISM Unita' di Ricerca di Roma 3



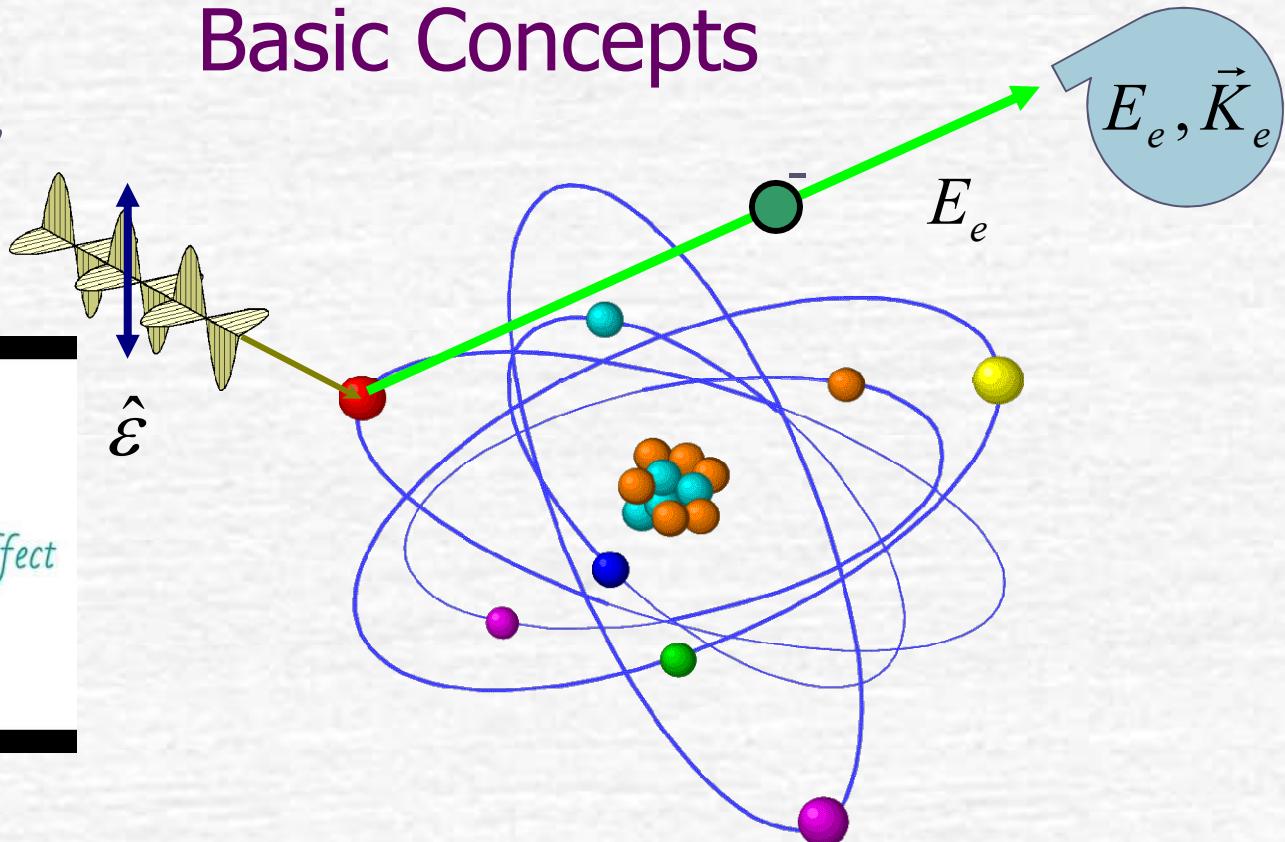
# Basic Concepts



*It's the  
PhotoElectric Effect*

$$E_e^{MAX} = h\nu - \Phi$$

$h\nu$



$$E_e = h\nu - \Phi - \Delta energy$$

# Outline

- 1. Photoelectron energy and bound state energy**
- 2. Satellite structures and multiplet splitting**
- 3. Chemical shift**
- 4. Molecular photoelectron spectra**
- 5. Photoelectron angular distributions**
- 6. Hole state relaxation**
- 7. Photoemission in solids**
- 8. EDC and core ionization**
- 9. Angular resolved PES**
- 10. High energy photoemission HAXPES**

# Photoemission Schematics:

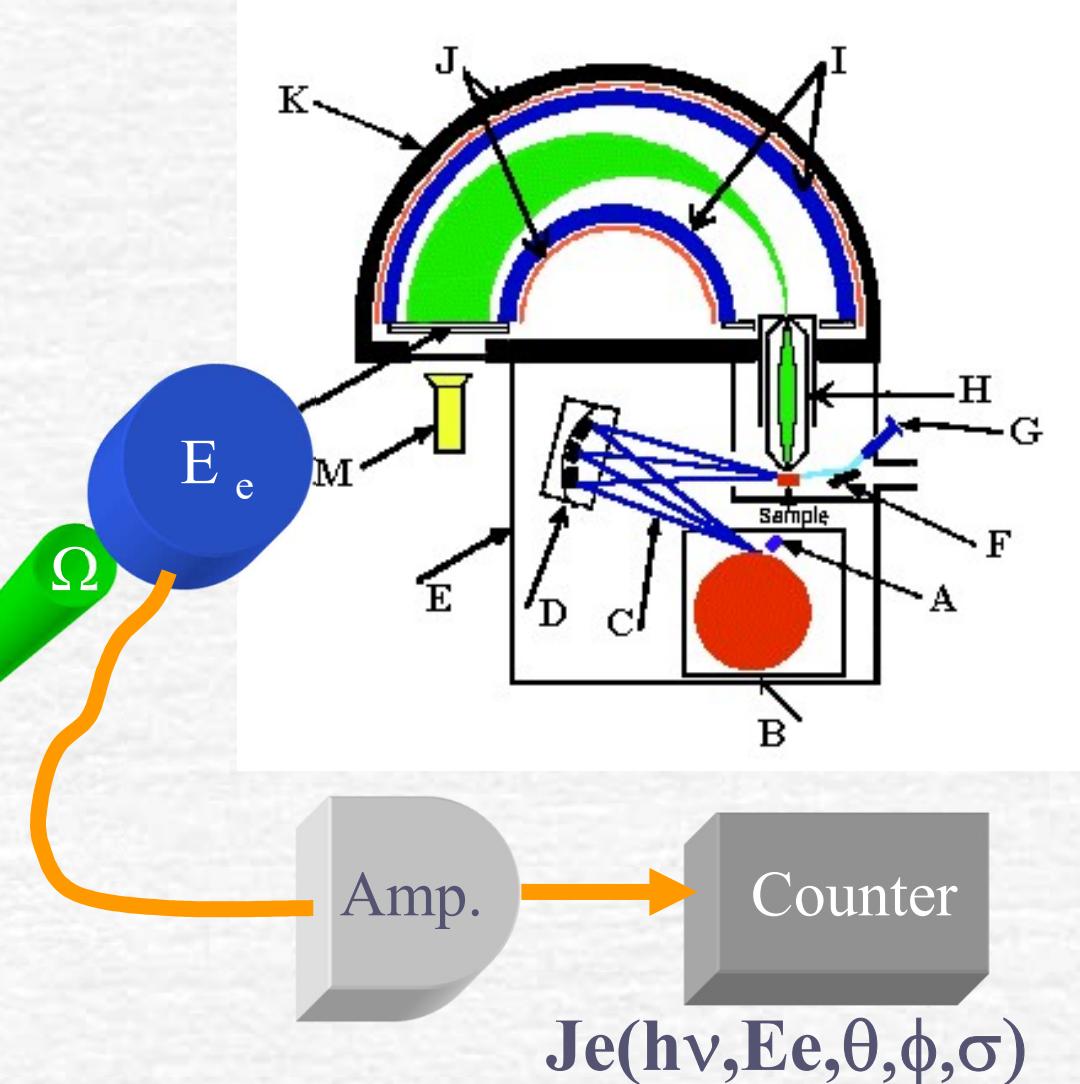
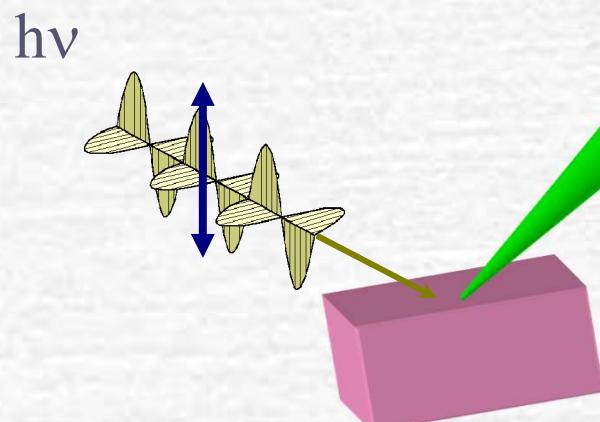
$\text{He I}\alpha=21.23\text{eV}$

$\text{He II}\alpha=40.82\text{eV}$

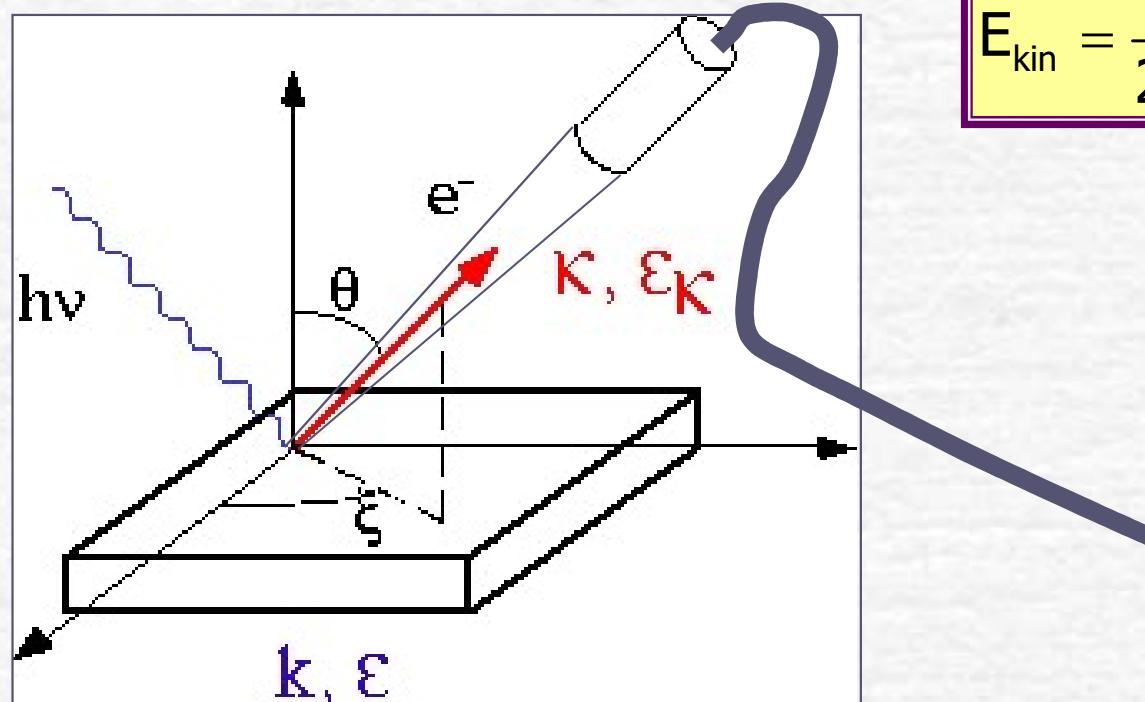
$\text{Mg K}\alpha_{1,2} = 1253,6 \text{ eV}$

$\text{Al K}\alpha_{1,2}=1486,6\text{eV}$

## Synchrotron Radiation



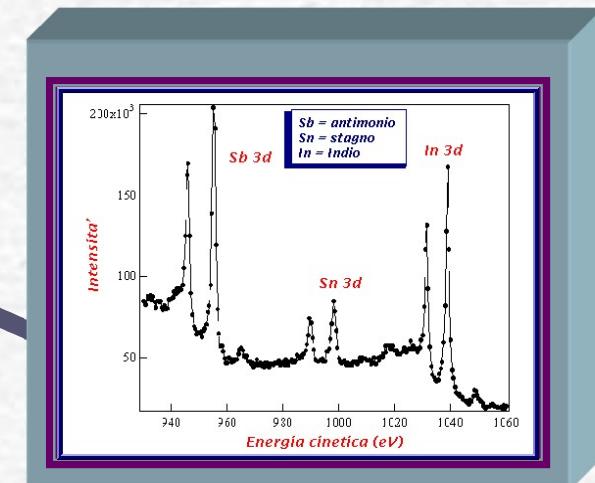
# ENERGY CONSERVATION, BINDING ENERGY AND PHOTOELECTRON ENERGY



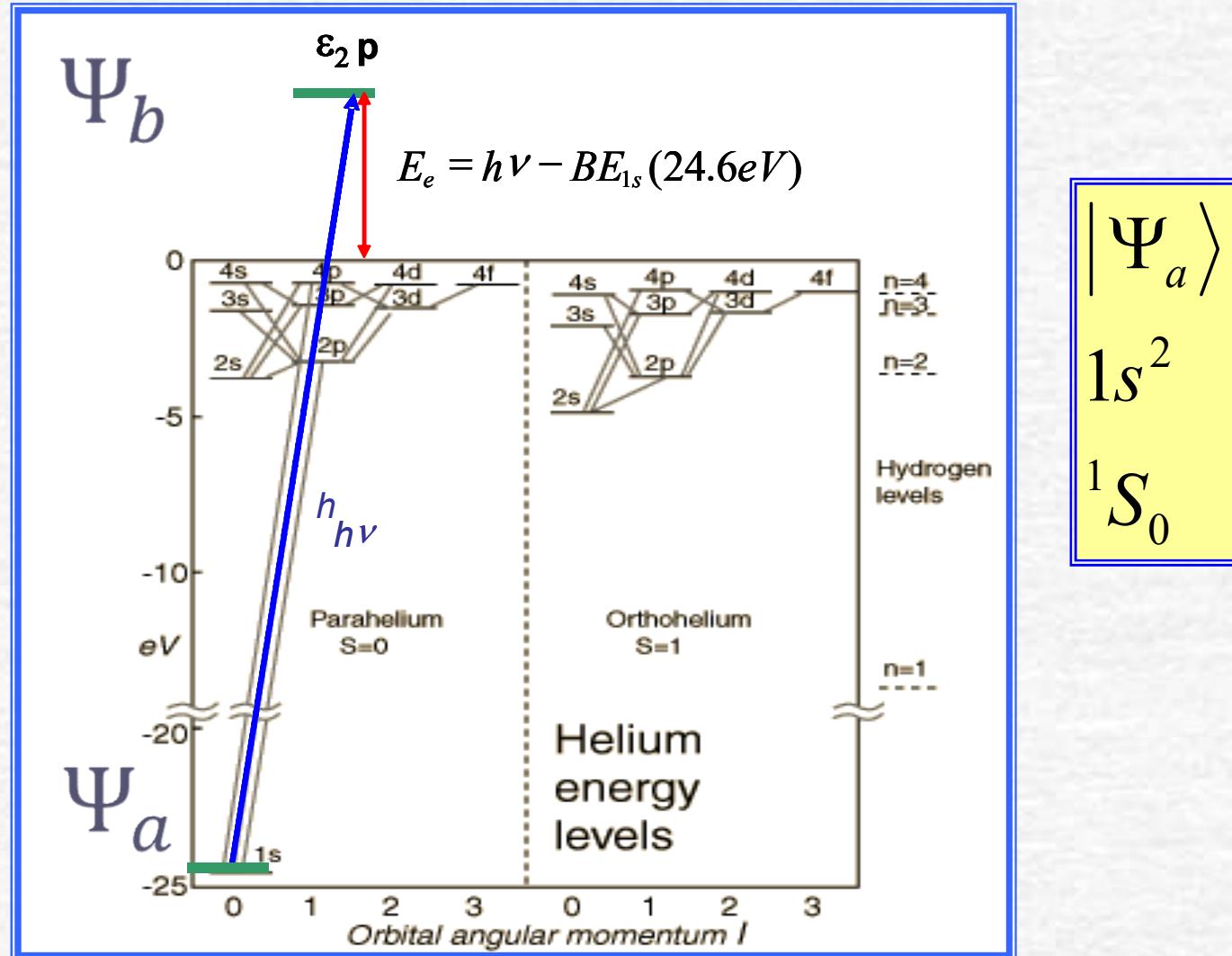
$$E_{\text{kin}} = \hbar\omega - \phi - |E_b|$$

$$E_{\text{kin}} = \frac{p^2}{2m}$$

$$p = \sqrt{2mE_{\text{kin}}}$$



# The photoemission process



# Interaction radiation matter and transition probability

$$\frac{d\sigma}{dh\nu} = 4\pi^2 \alpha h\nu \sum_B \left| \hat{\varepsilon} \bullet \left\langle \Psi_B \left| \sum_i \vec{r}_i \right| \Psi_A \right\rangle \right|^2 \delta(E_B - E_A - h\nu)$$

Bertoni's lectures at this school

Initial state A = Neutral ground(excited) state  
Final state B = Residual ion + free electron(s)

$$\frac{d\sigma}{dh\nu} = \int \int_E \int_\Omega \frac{d\sigma}{d\Omega dE} d\Omega dE$$

# X-section vs. Photoemission current

$$J_e(h\nu, \vartheta, \phi) = J_{h\nu}(\rho l) \int \int \frac{d\sigma}{d\Omega dE} F_{an}(E, \Omega) \eta_{\text{det}}(E) d\Omega dE$$

Photoemission peak lineshape

- |   |           |
|---|-----------|
| 1. Photon monochromaticity                      | Gaussian  |
| 2. Electron analyzer resolution                 | Gaussian  |
| 3. Final state lifetime (uncertainty principle) | Lorentian |

Lineshape = Convolution (1,2,3)

# Energy balance for 2e atom

$$E_B = E_A + h\nu$$

$$\Psi_A = \hat{A}\phi_1\phi_2$$

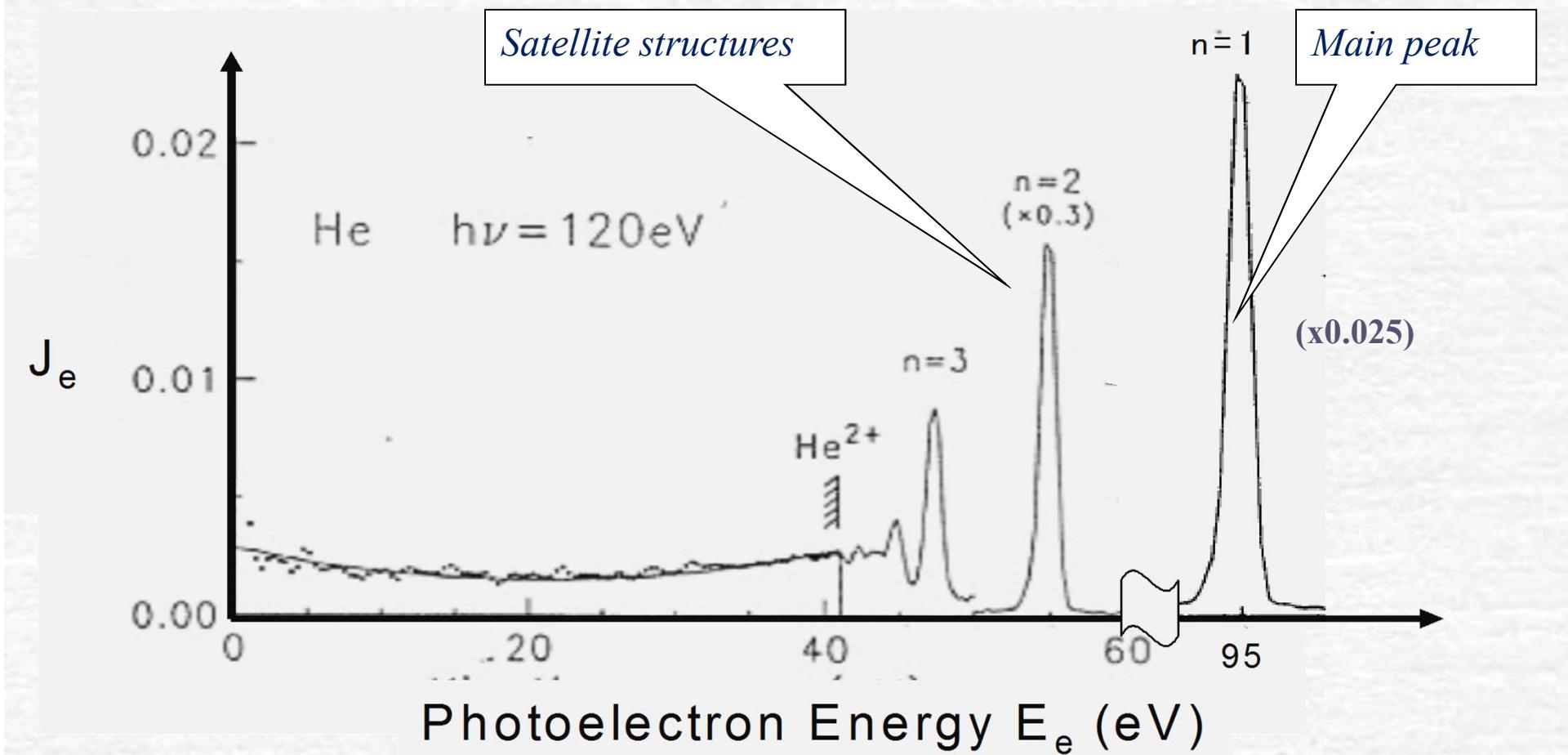
$$\Psi_B = \hat{A}\phi_1\epsilon_2$$

$$\cancel{E_{1s}} + E_e = \cancel{E_{1s}} + E_{1s} + h\nu$$

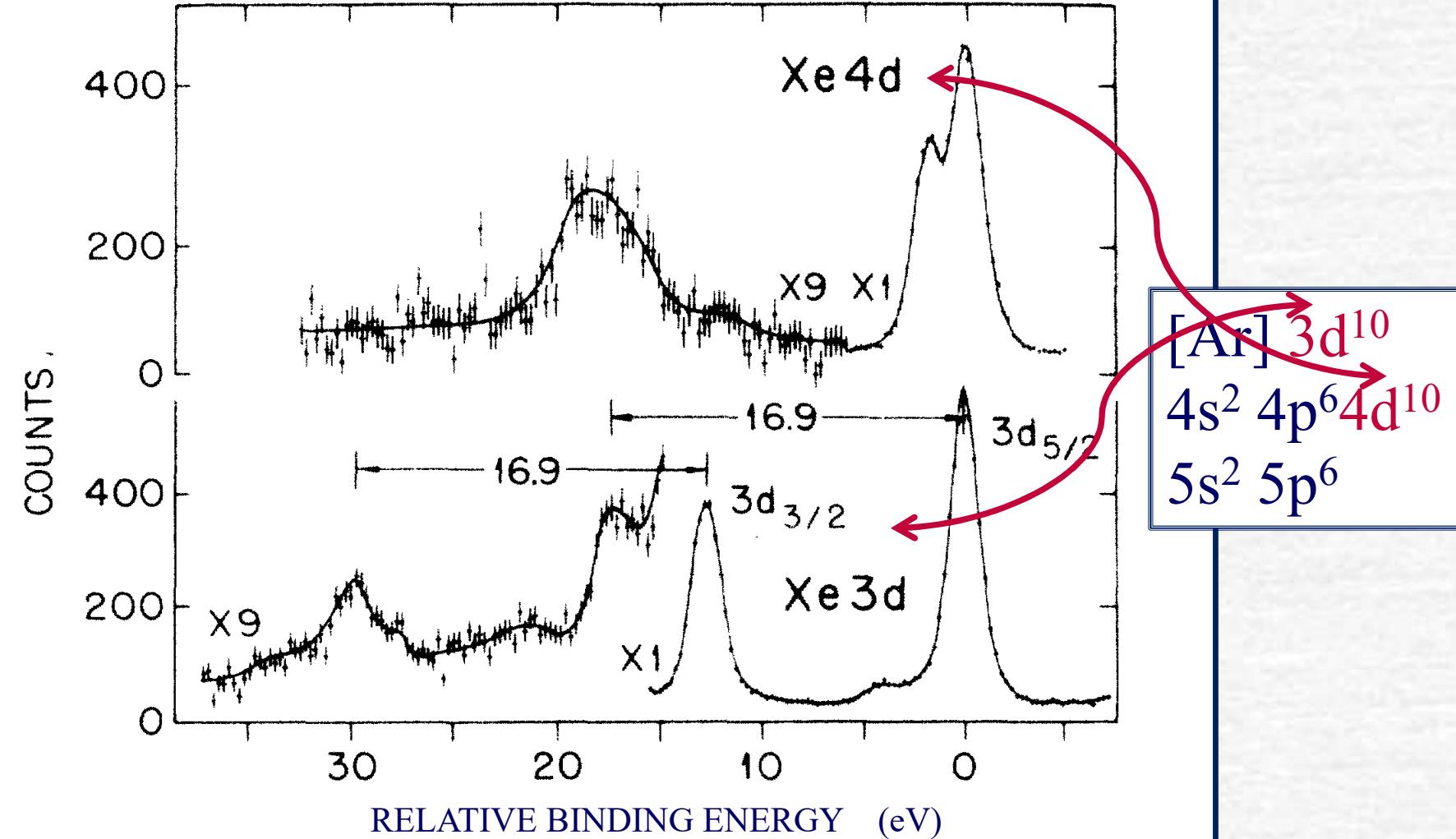
$$E_e = h\nu - BE_{1s} (24.6eV)$$

One single photoemission peak is expected  
Energy and momentum are conserved

# Complexity of the photoemission spectrum



# How complex can it be?

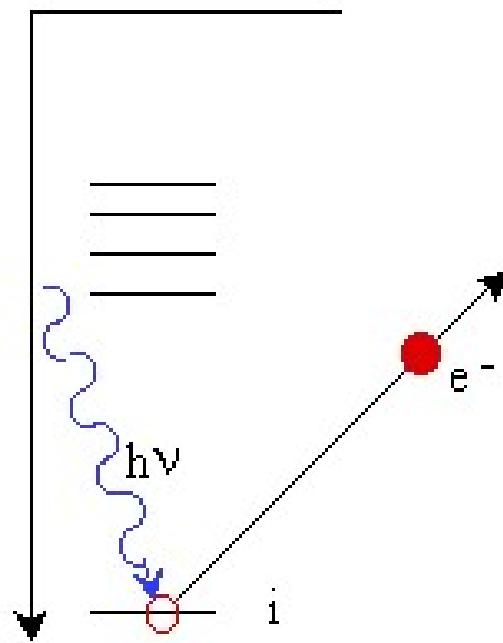


PHYSICAL REVIEW VOLUME 9, NUMBER 4 APRIL 1974 1603

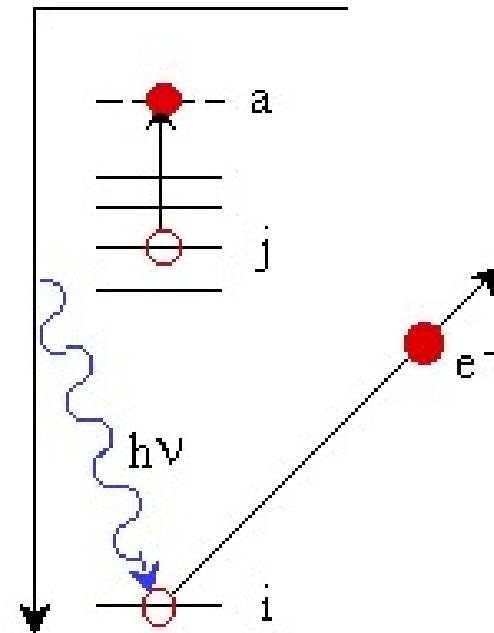
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# Primary photoionization processes

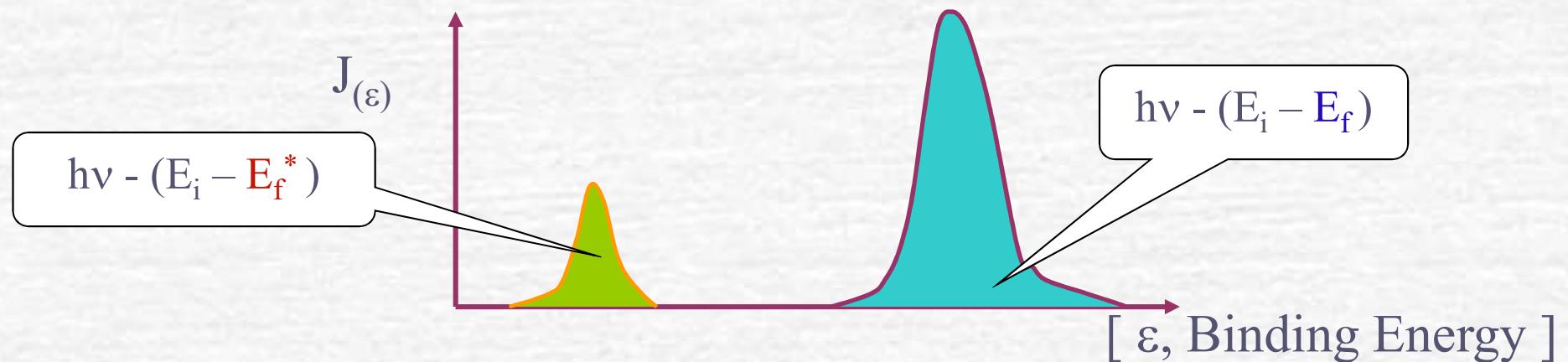
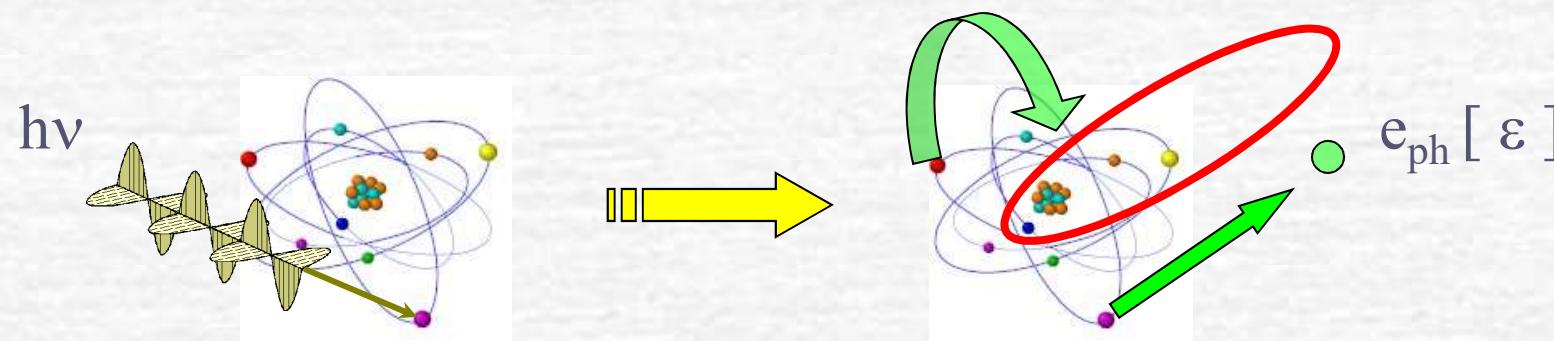
MAIN PROCESS



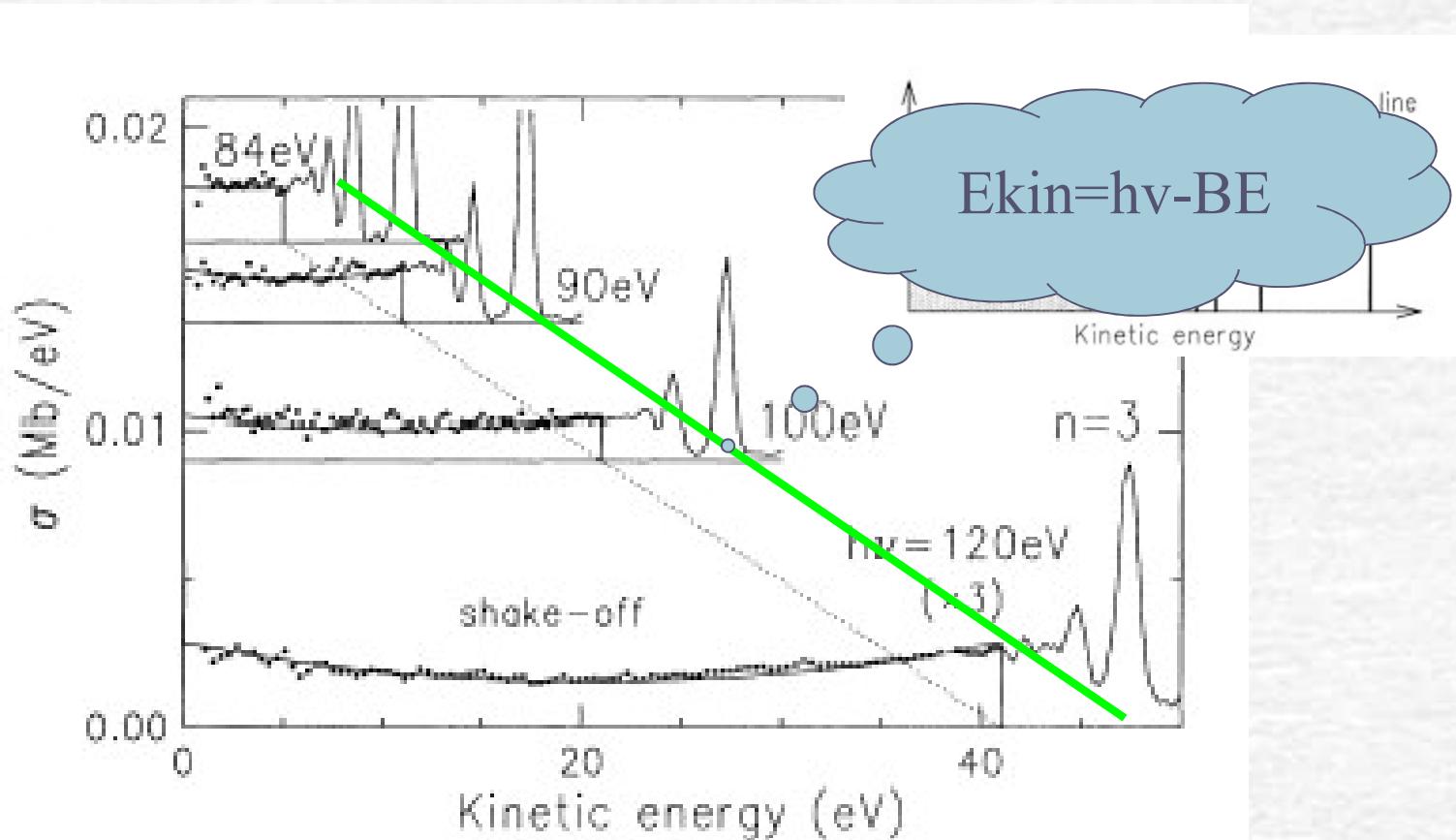
SATELLITE PROCESS



- Photon = single particle operator
- 2 or more particles involved in final state = e-e correlation
- Relaxation & e-e correlation in photoemission = satellite



# The He satellite structure



# A many electron atom

$$H_0 \left| \Psi_A^{(N)} \right\rangle = E_A^{(N)} \left| \Psi_A^{(N)} \right\rangle$$

$$\begin{aligned} H_0 &= H_0(kin) + H_0(e-n) + H_0(e-e) + H_0(s-o) = \\ &= \sum_1^N \frac{p_i^2}{2m} + \sum_1^N -\frac{Ze^2}{r_i} + \sum_{i>j}^N \frac{e^2}{r_{ij}} + \sum_1^N \zeta(r_j) \vec{l}_i \bullet \vec{s}_i \end{aligned}$$

$$\left| \Psi_A^{(N)} \right\rangle = \hat{A}(\phi_j(\vec{r}_i, \sigma_i); \Psi_R^{(N-1)})$$

Single  
particle  
orbital

$$H'_0 \left| \Psi_B^{(N)} \right\rangle = E_B^{(N)} \left| \Psi_B^{(N)} \right\rangle$$

# sudden approximation

Neglect  
nucleus  
motion

$$\left| \Psi_B^{(N)} \right\rangle = \hat{A}(\varepsilon_l; \left| \Psi_B^{(N-1)} \right\rangle)$$

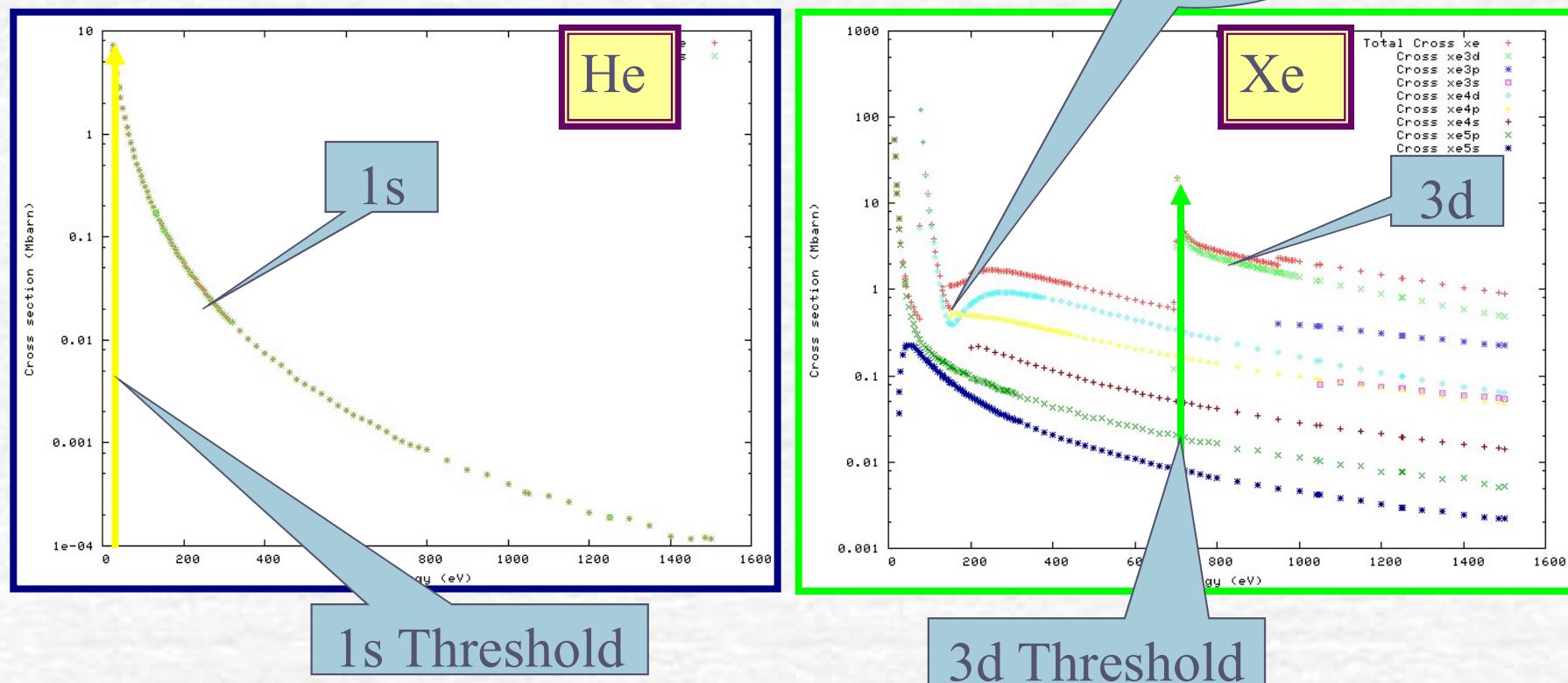
$$\frac{d\sigma}{d\Omega dE_e} \propto \frac{1}{h\nu} \sum_{A,B} \left| \hat{\varepsilon} \bullet \left\langle \varepsilon_l \left| \phi_j(\vec{r}_j, \sigma_j) \right\rangle \right\rangle \left\langle \Psi_B^{(N-1)} \left| \Psi_R^{(N-1)} \right\rangle \right|^2 \delta(E_e + E_B^{(N-1)} - E_A - h\nu) \right.$$

# frozen core approximation

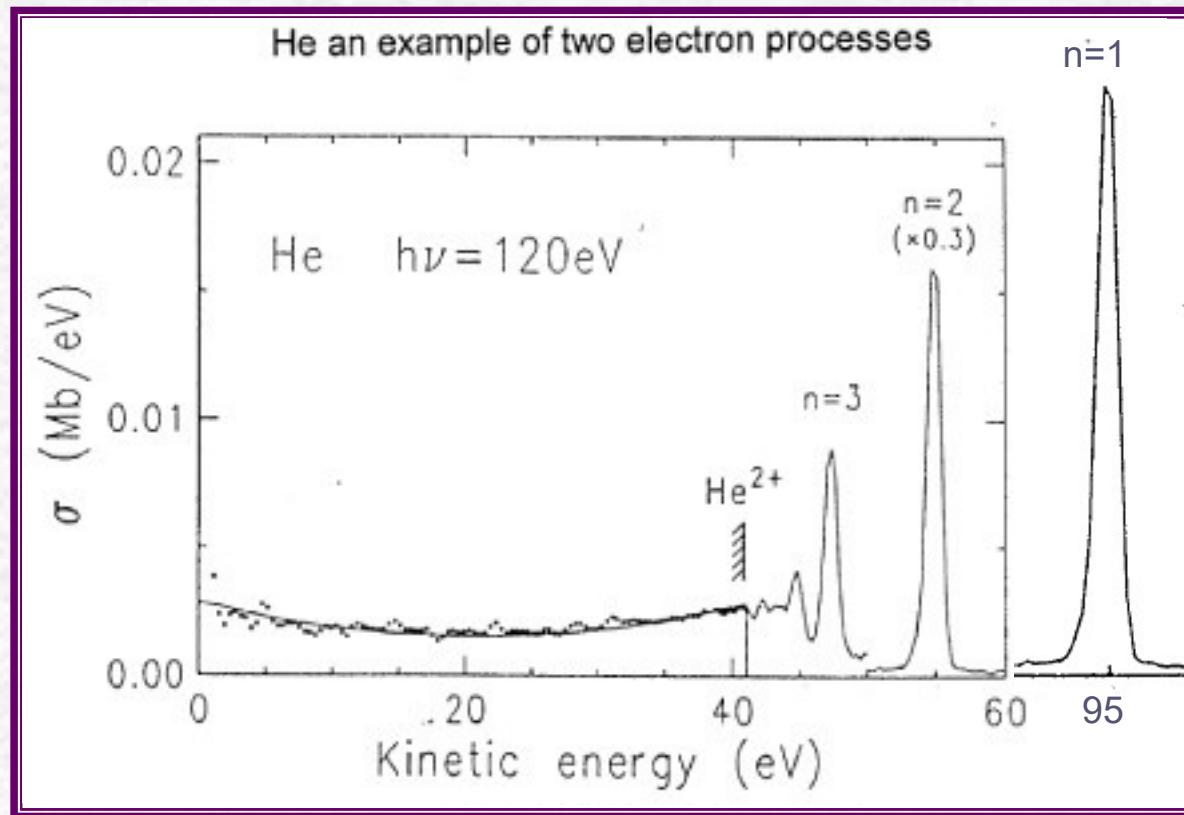
$$H_0' = H_0$$

$$\frac{d\sigma}{d\Omega dE_e} \propto \frac{1}{h\nu} \sum_{A,B} \left| \hat{\varepsilon} \bullet \left\langle \varepsilon_l \left| \vec{r}_j \left| \phi_j(\vec{r}_j, \sigma_j) \right\rangle \right\rangle \right|^2 \delta(E_e + \varepsilon_j - h\nu) \right.$$

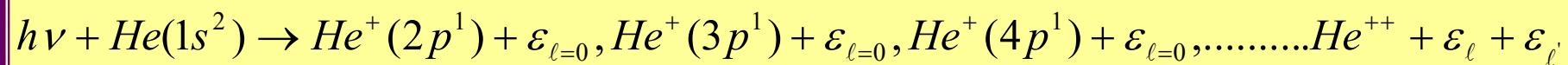
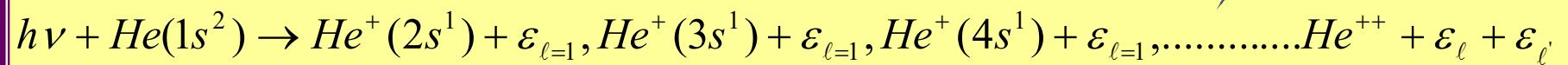
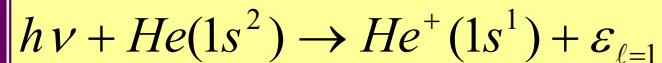
# Total photoemission cross section



# Photoelectron current vs. photoelectron energy

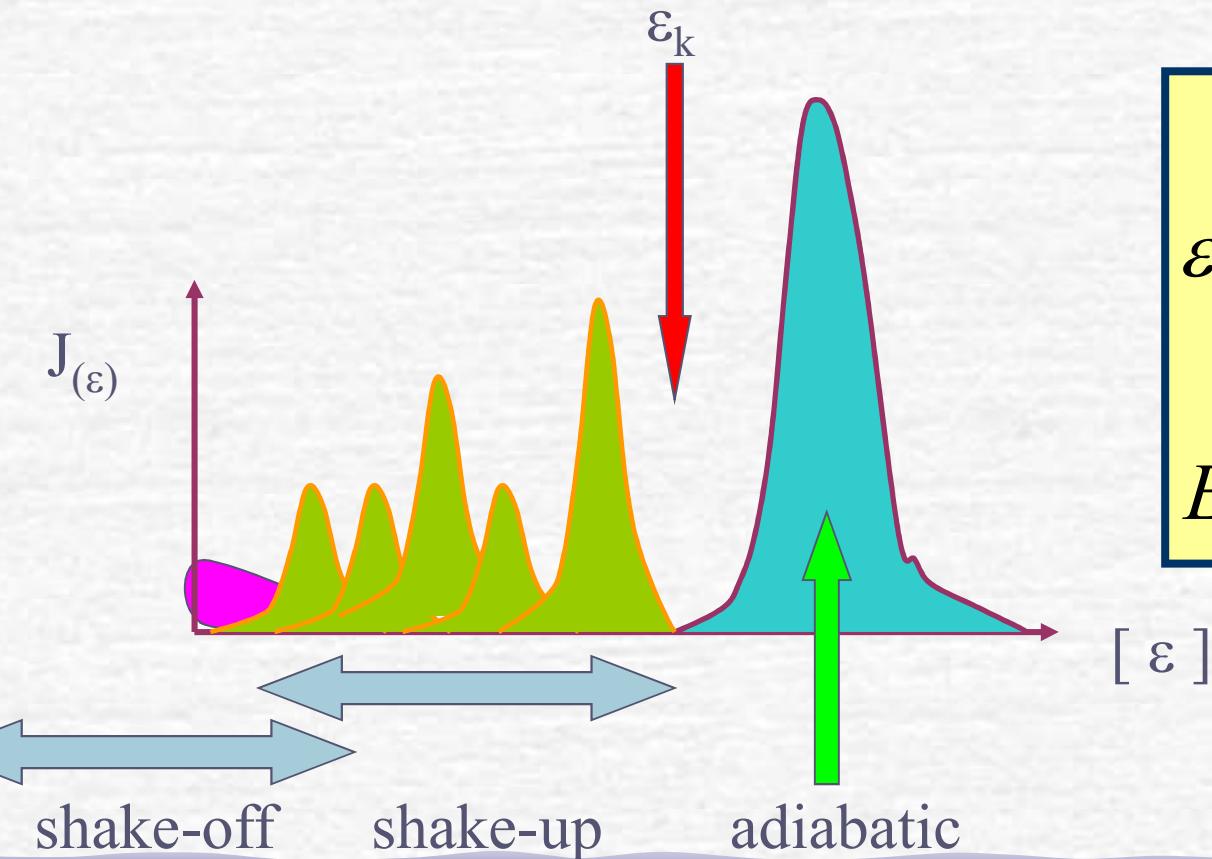


Monopole  
and dipole  
selection  
rules



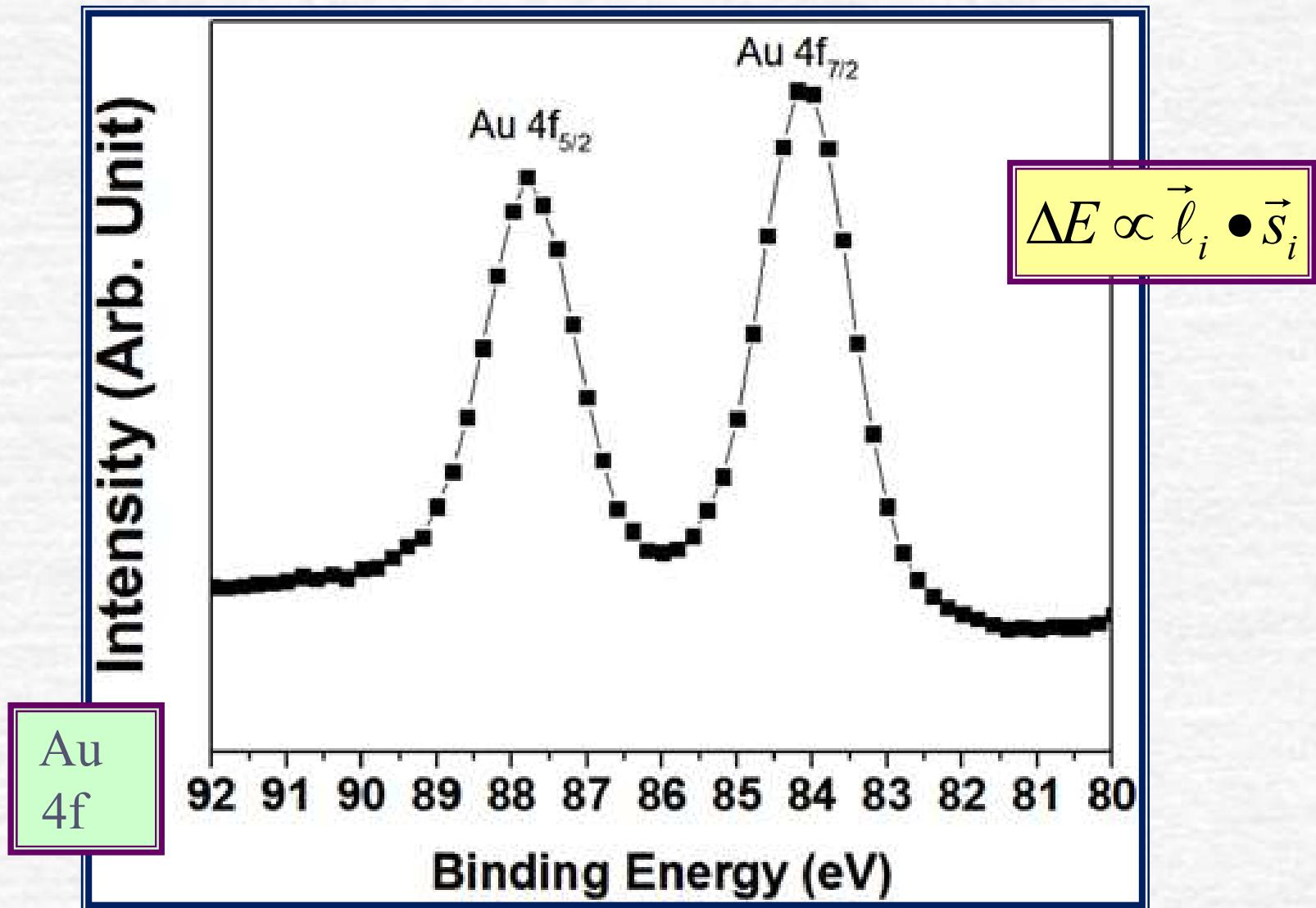
# Koopmans energy vs. photoemission peaks

$$\frac{d\sigma}{d\Omega dE_e} \propto \frac{1}{h\nu} \sum_{A,B} \left| \hat{\varepsilon} \bullet \left\langle \varepsilon_l \left| \vec{r}_j \right| \phi_j(\vec{r}_j, \sigma_j) \right\rangle \left\langle \Psi_B^{(N-1)} \left| \Psi_R^{(N-1)} \right. \right\rangle \right|^2 \delta(E_e + E_B^{(N-1)} - E_A - h\nu)$$



$$\varepsilon_k = \frac{\sum_i \varepsilon_i I_i}{\sum_i I_i}$$
$$E_{relax} = \varepsilon_{adiab} - \varepsilon_k$$

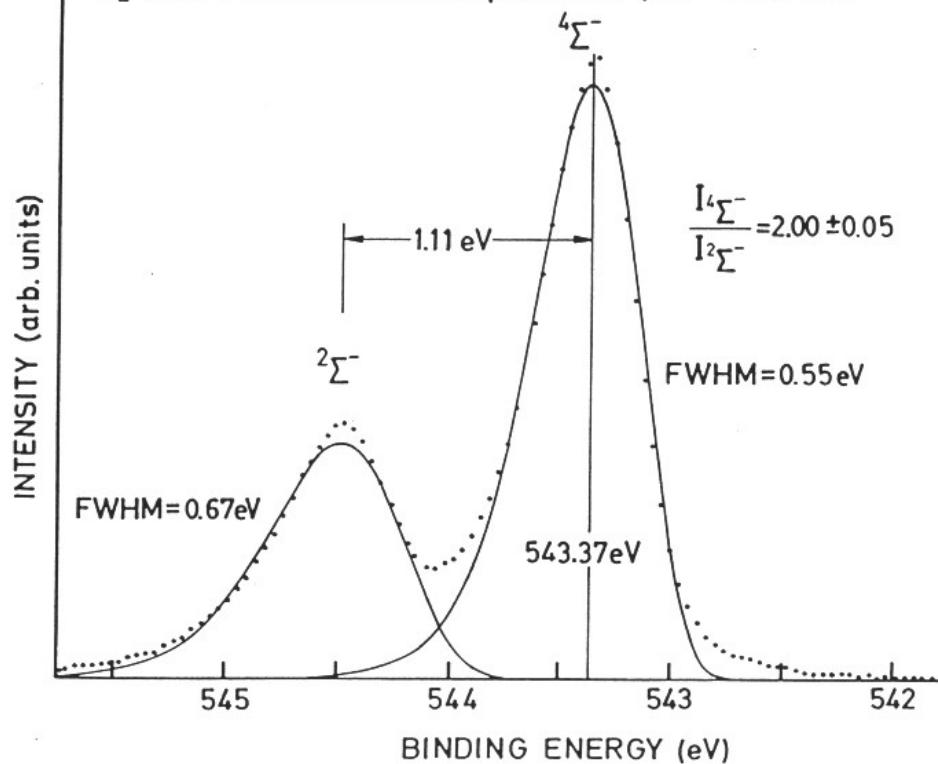
# Spin orbit splitting



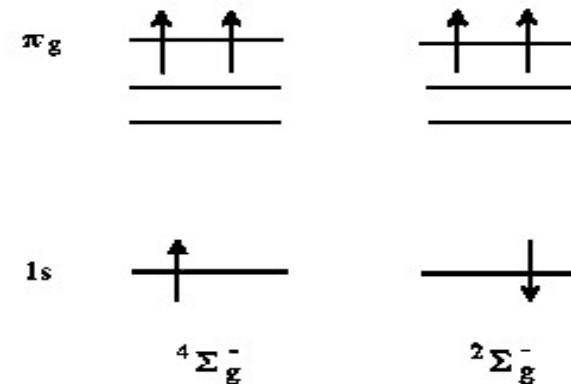
# Molecular multiplet splitting O<sub>2</sub>



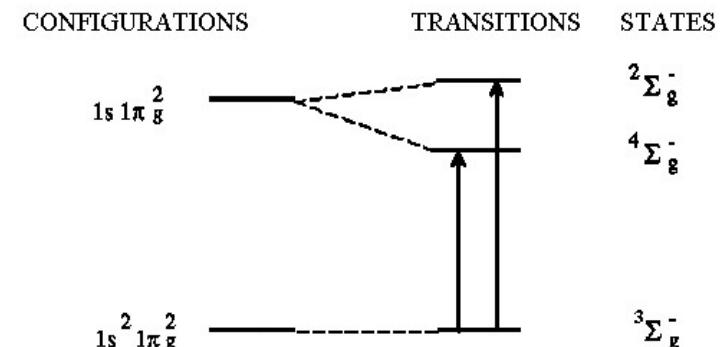
O<sub>2</sub> 01s Photoelectron spectrum ( $h\nu = 1487 \text{ eV}$ )



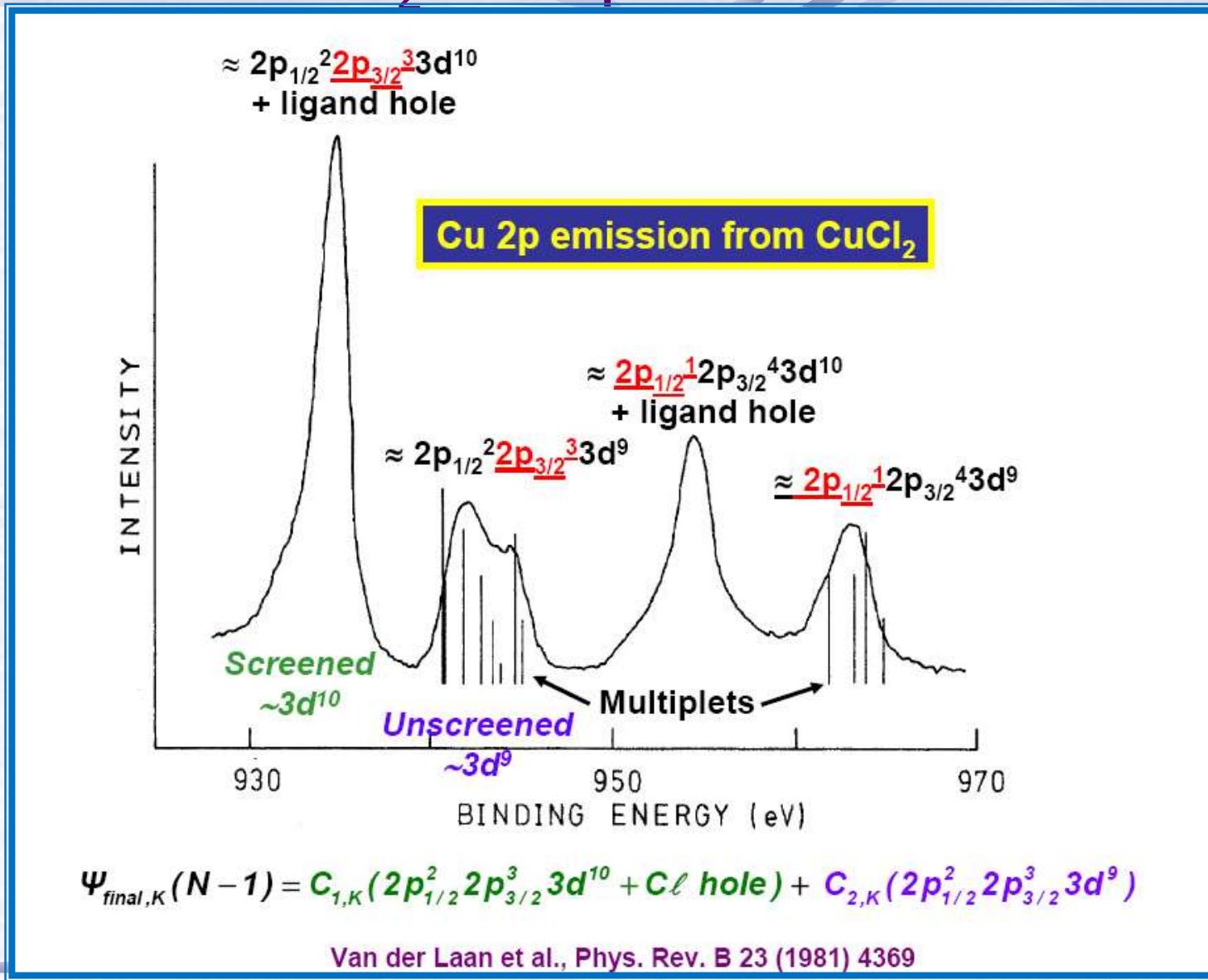
CORE HOLE MULTIPLET STATES OF O<sub>2</sub><sup>+</sup>



MULTIPLET TRANSITIONS IN O<sub>2</sub>



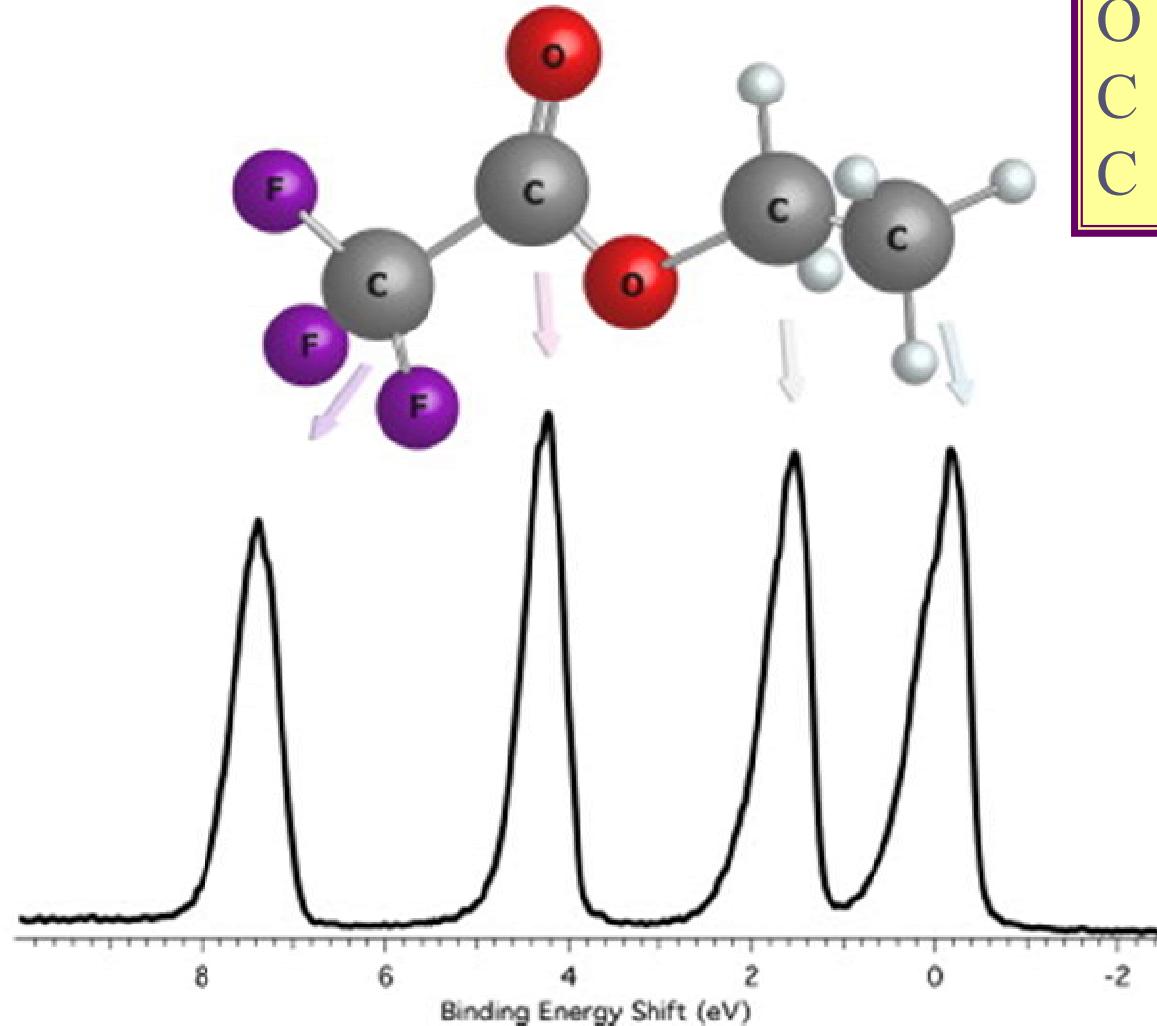
# CuCl<sub>2</sub> multiplet & satellite



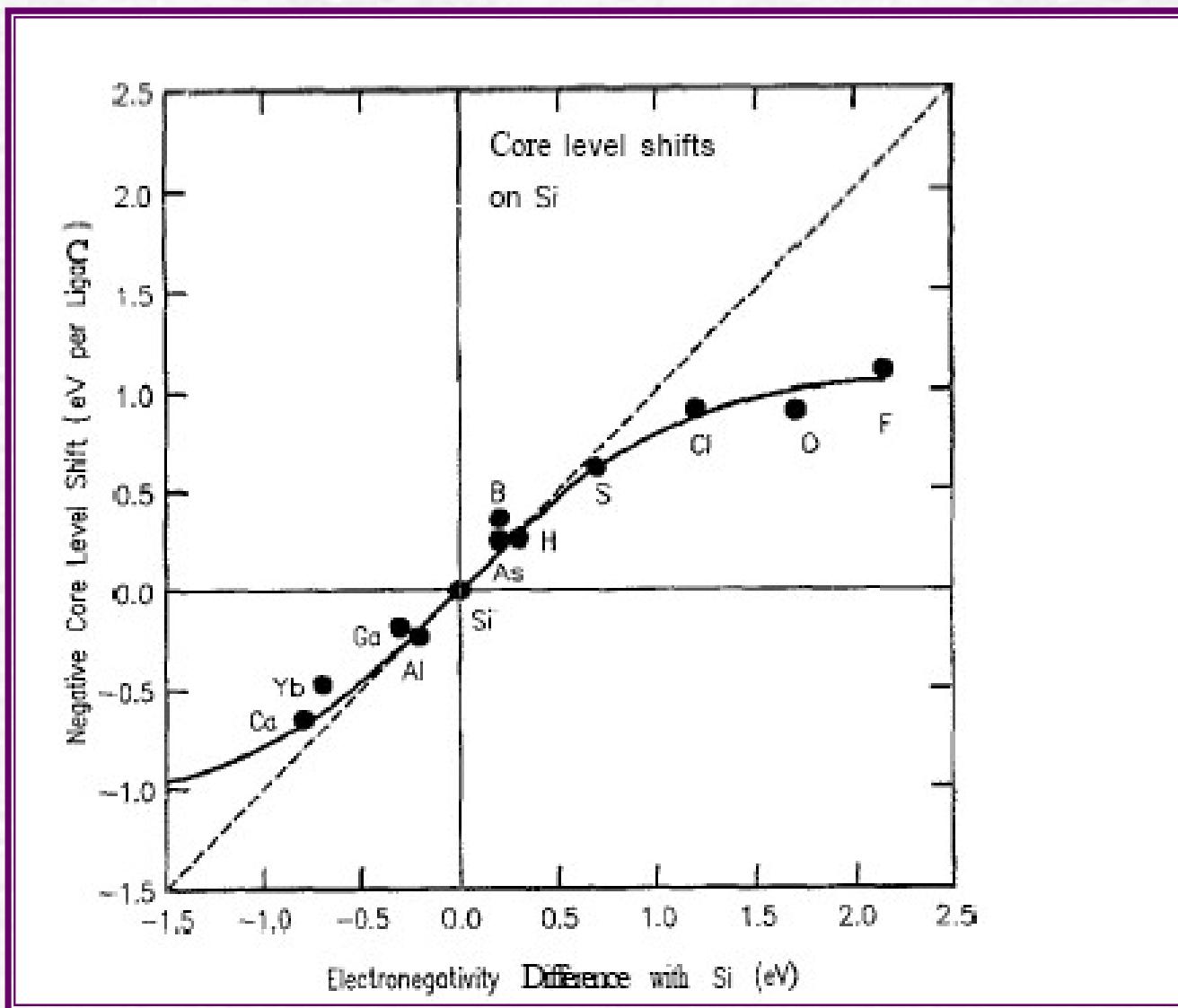
# Chemical shift

Journal of Electron Spectroscopy and Related Phenomena  
Volume 185, Issues 8–9, September 2012, Pages 191–197

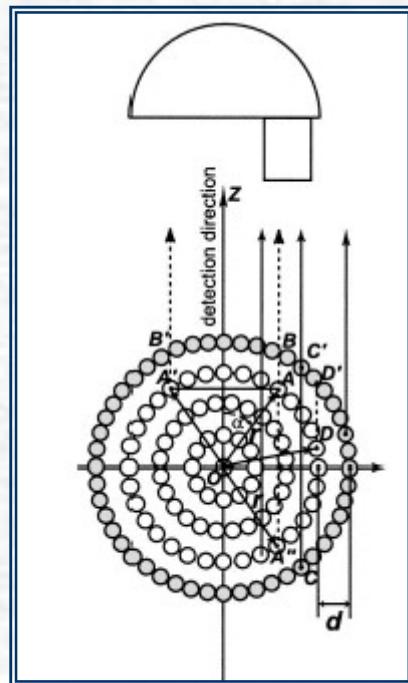
|      |                      |
|------|----------------------|
| C 1s | 285-300 eV           |
| O 1s | 530-540 eV           |
| C 1s | $\text{CO}_2$ 298 eV |
| C 1s | $\text{CH}_4$ 291 eV |



# Chemical shift vs. electronegativity



# Sensitivity to the local environment in free clusters



J. Chem. Phys.,  
Vol. 120, No. 1, 1  
January 2004

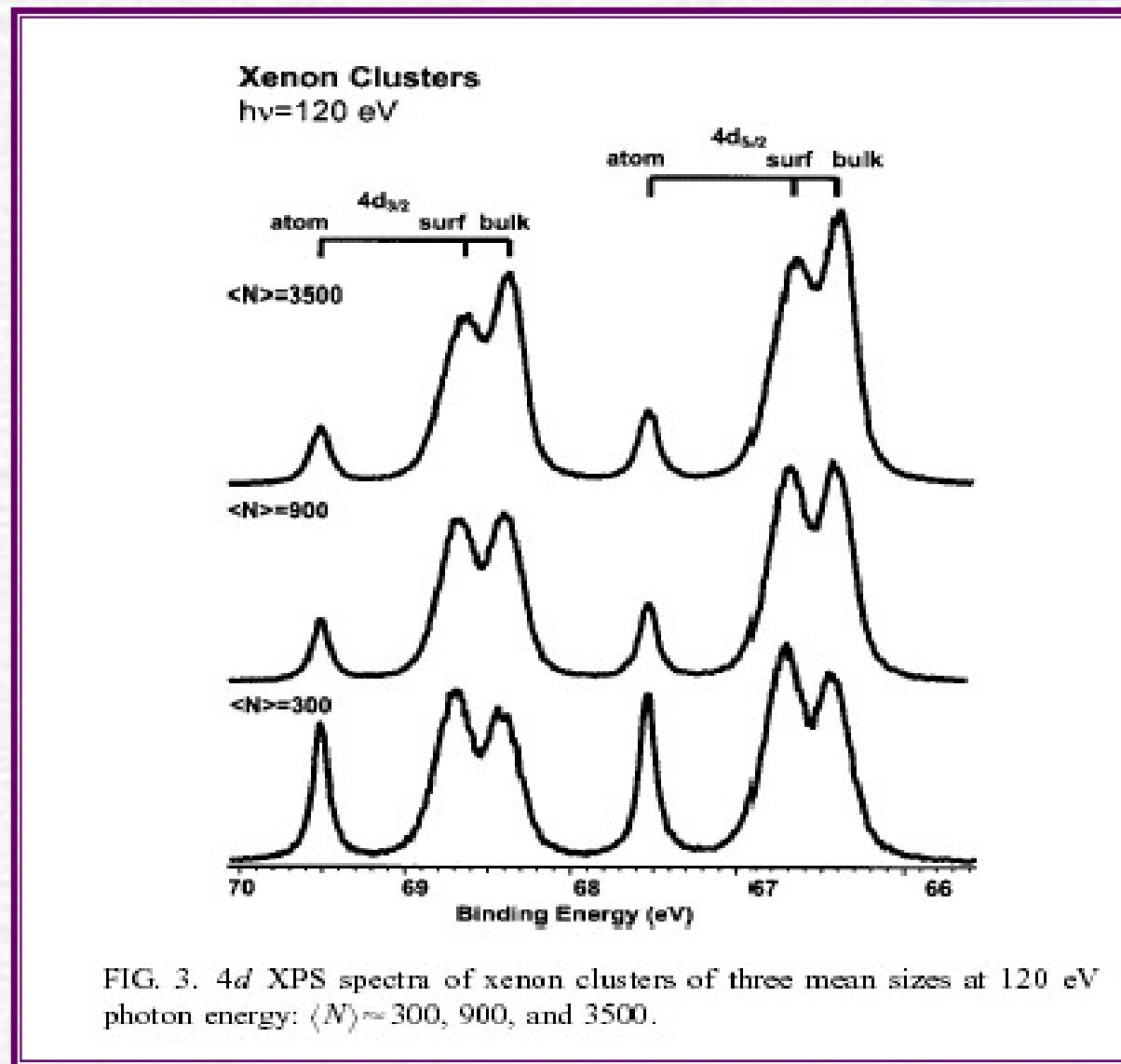
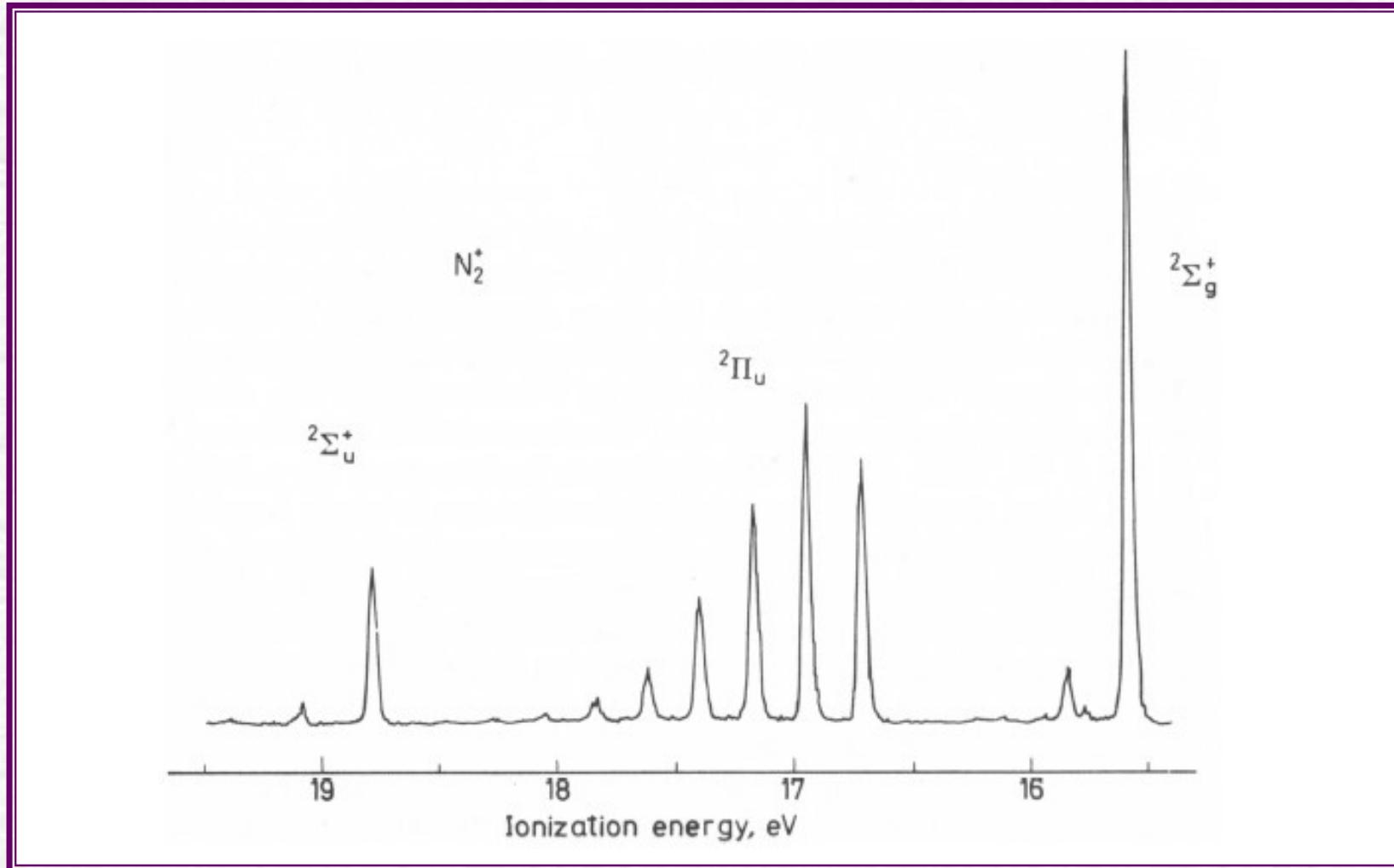
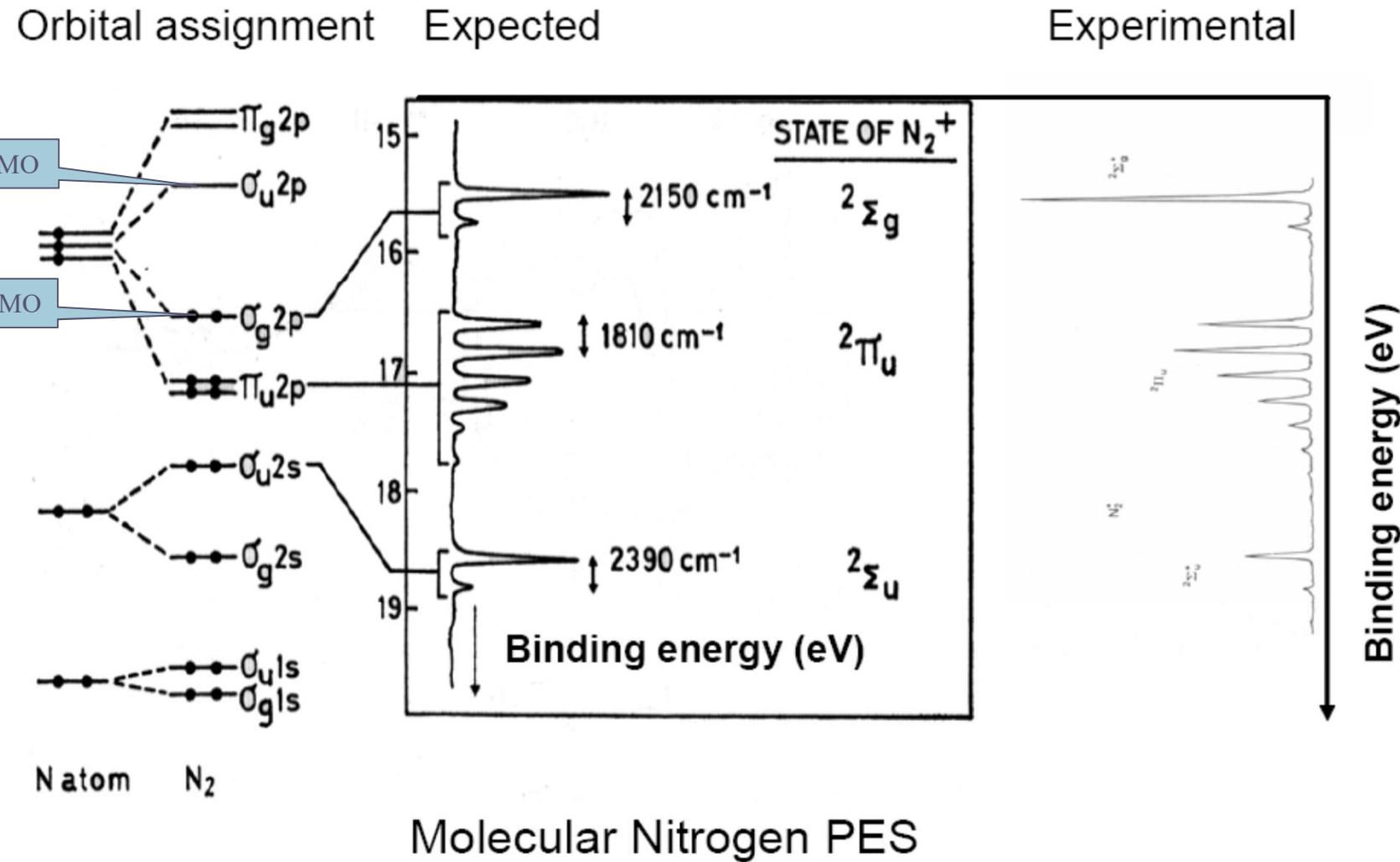


FIG. 3. 4d XPS spectra of xenon clusters of three mean sizes at 120 eV photon energy:  $\langle N \rangle \approx 300$ , 900, and 3500.

# PES spectrum of N<sub>2</sub>



# Experimental PE spectrum of N<sub>2</sub>, and MOs



# Molecular PE x-section

$$\begin{aligned}
 H_0 &= H_0(kin) + H_0(e-n) + H_0(e-e) + H_0(s-o) + H_0(n-n) = \\
 &= \sum_1^N \frac{p_i^2}{2m} + \sum_1^N -\frac{Ze^2}{r_i} + \sum_{i>j}^N \frac{e^2}{r_{ij}} + \sum_1^N \zeta(r_j) \vec{l}_i \bullet \vec{s}_i + \sum_{i>j}^M \frac{e^2 Z_i Z_j}{r_{ij}}
 \end{aligned}$$

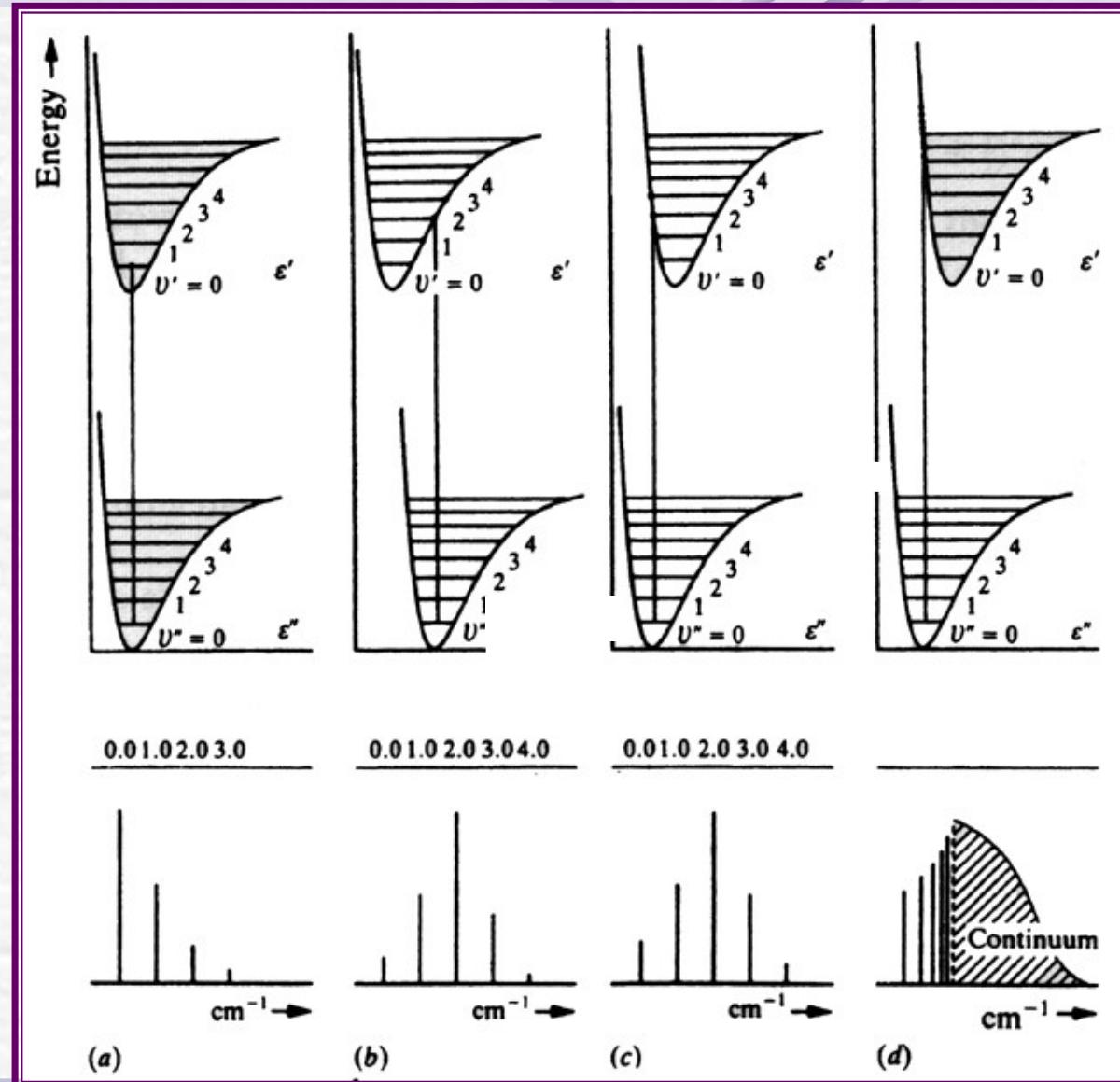
Born Oppenheimer

$$\left| \Psi_{A,B}^{(N)} \right\rangle = \left| \Psi_{A,B}^{(N)} \right\rangle \left| \Psi_{A,B}^{vib} \right\rangle$$

$$\frac{d\sigma}{d\Omega dE_e} \propto \frac{1}{h\nu} \sum_{A,B} \left| \hat{\varepsilon} \bullet \left\langle \varepsilon_l \left| \vec{r}_j \left| \sum_{A\lambda} C_{A\lambda} \phi_{A\lambda} \right\rangle \right\rangle \left\langle \Psi_B^{(N-1)} \left| \Psi_R^{(N-1)} \right\rangle \right| \right|^2$$

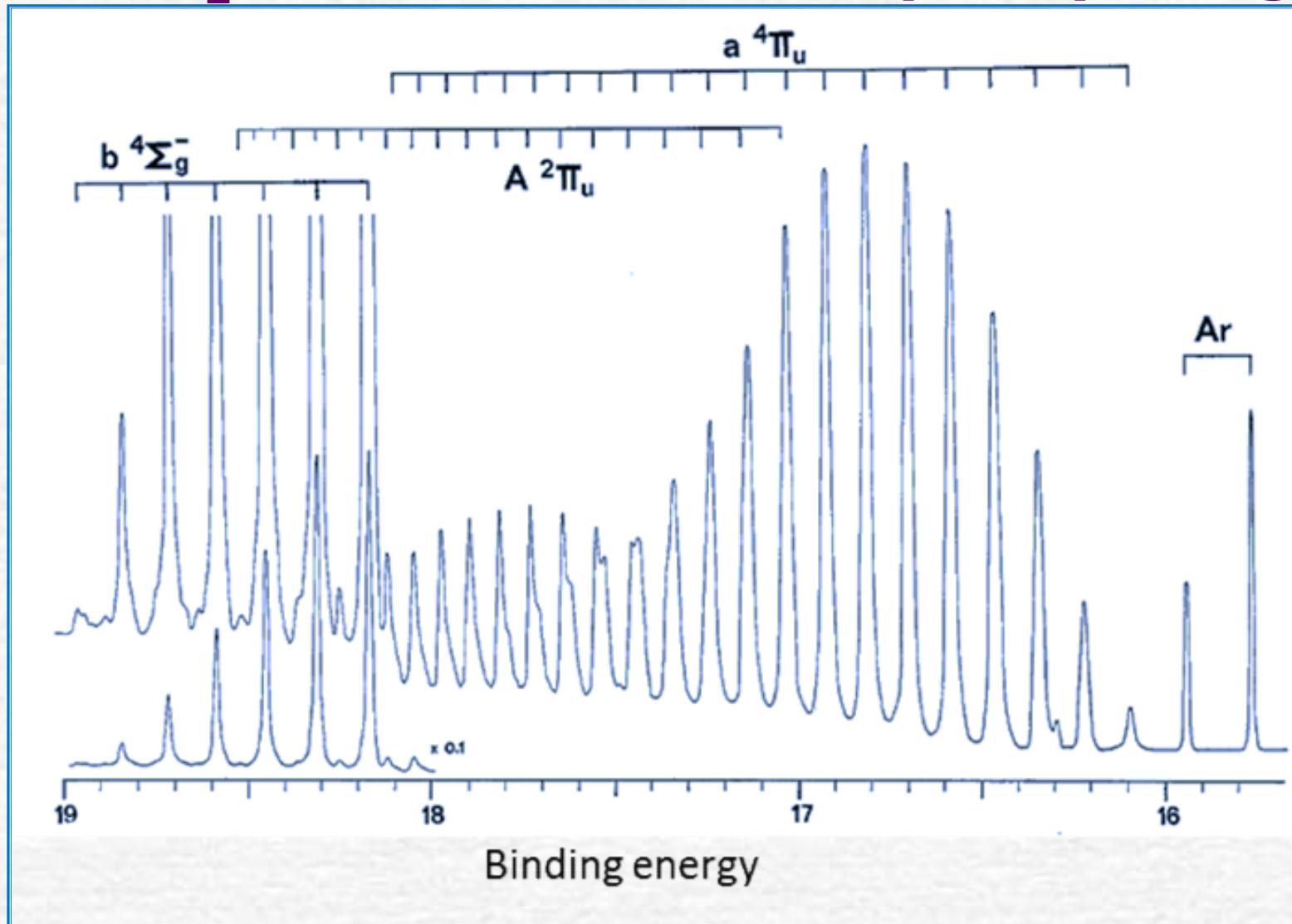
$$\left| \left\langle \Psi_B^{vib} \left| \Psi_A^{vib} \right\rangle \right|^2 \delta(E_e + E_B^{(N-1)} - E_A - h\nu) \right.$$

# Frank Condon Factors

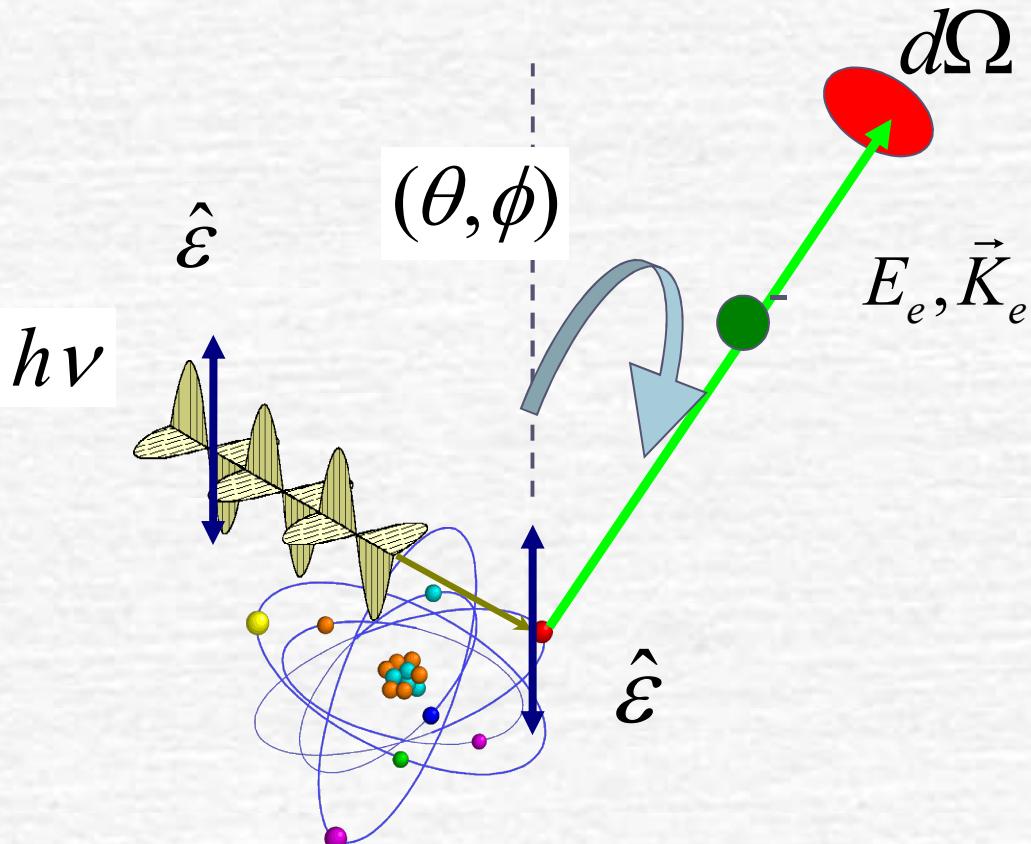


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# PES O<sub>2</sub> vibrational and multiplet splitting

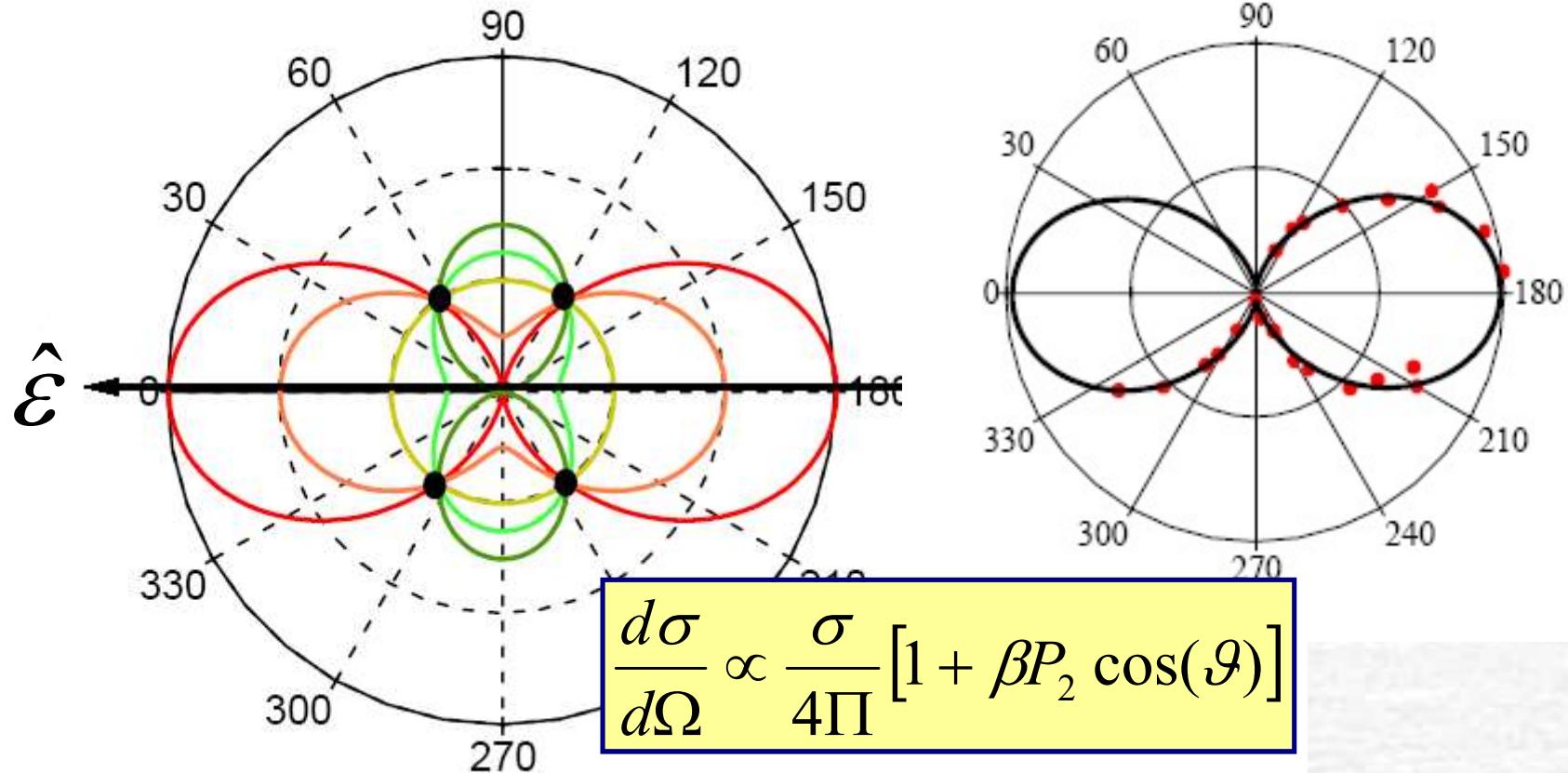


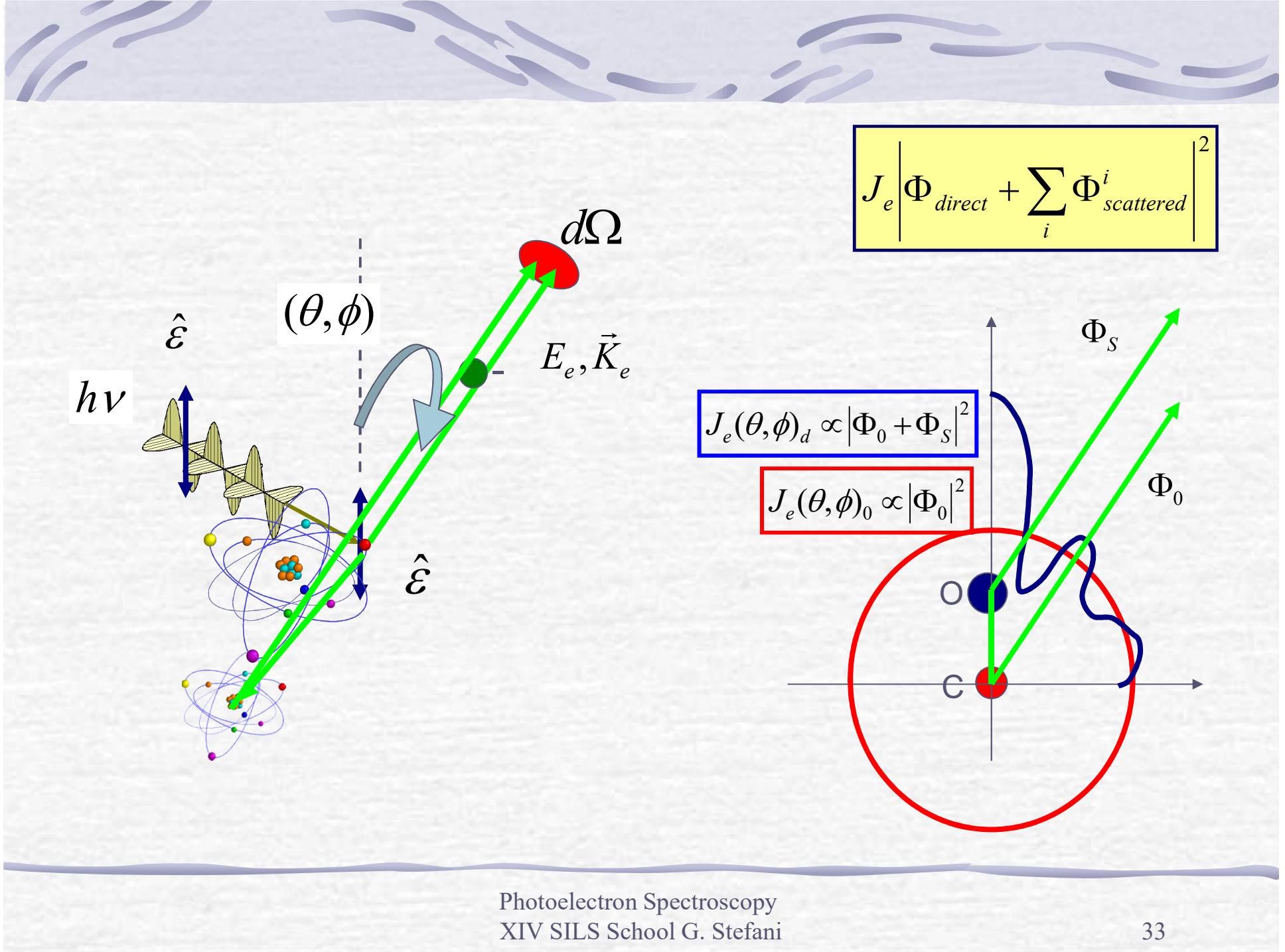
# PE angular distribution



$$\frac{d\sigma}{d\Omega} \propto \frac{1}{h\nu} \sum_{A,B} \left| \hat{\varepsilon} \bullet \left\langle \varepsilon_l | \vec{r}_j | \phi_j(\vec{r}_j, \sigma_j) \right\rangle \left\langle \Psi_B^{(N-1)} | \Psi_R^{(N-1)} \right\rangle \right|^2$$

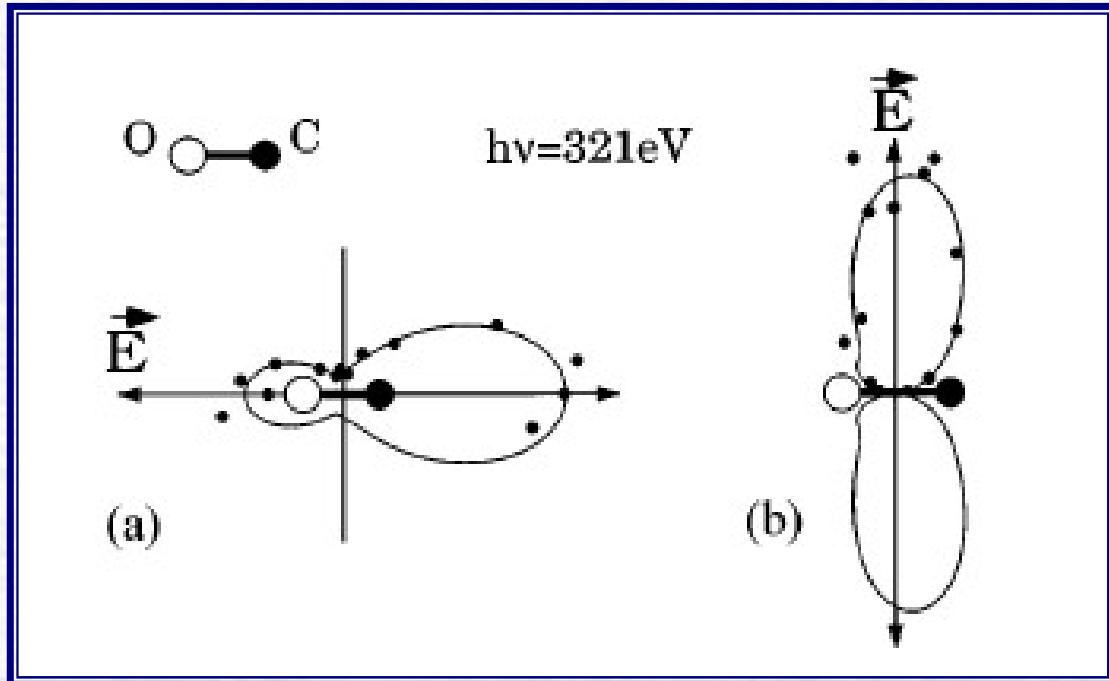
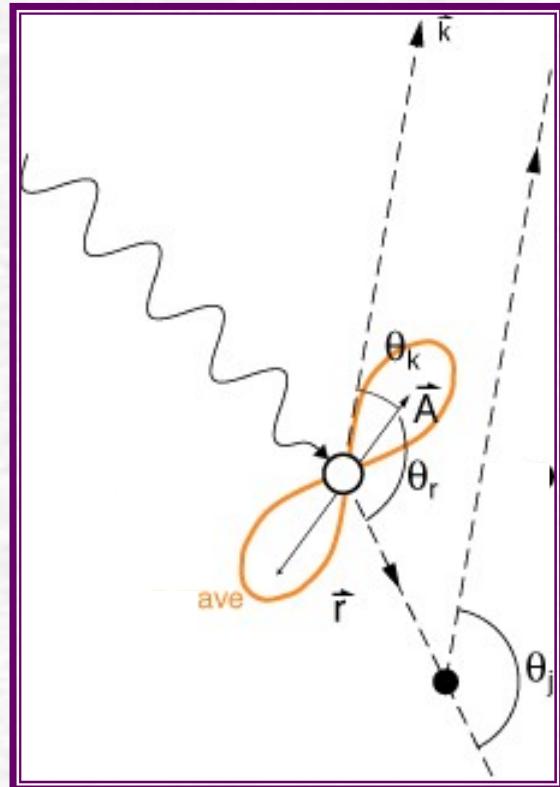
# Angular distributions





# Fixed in space molecules

## CO C 1s



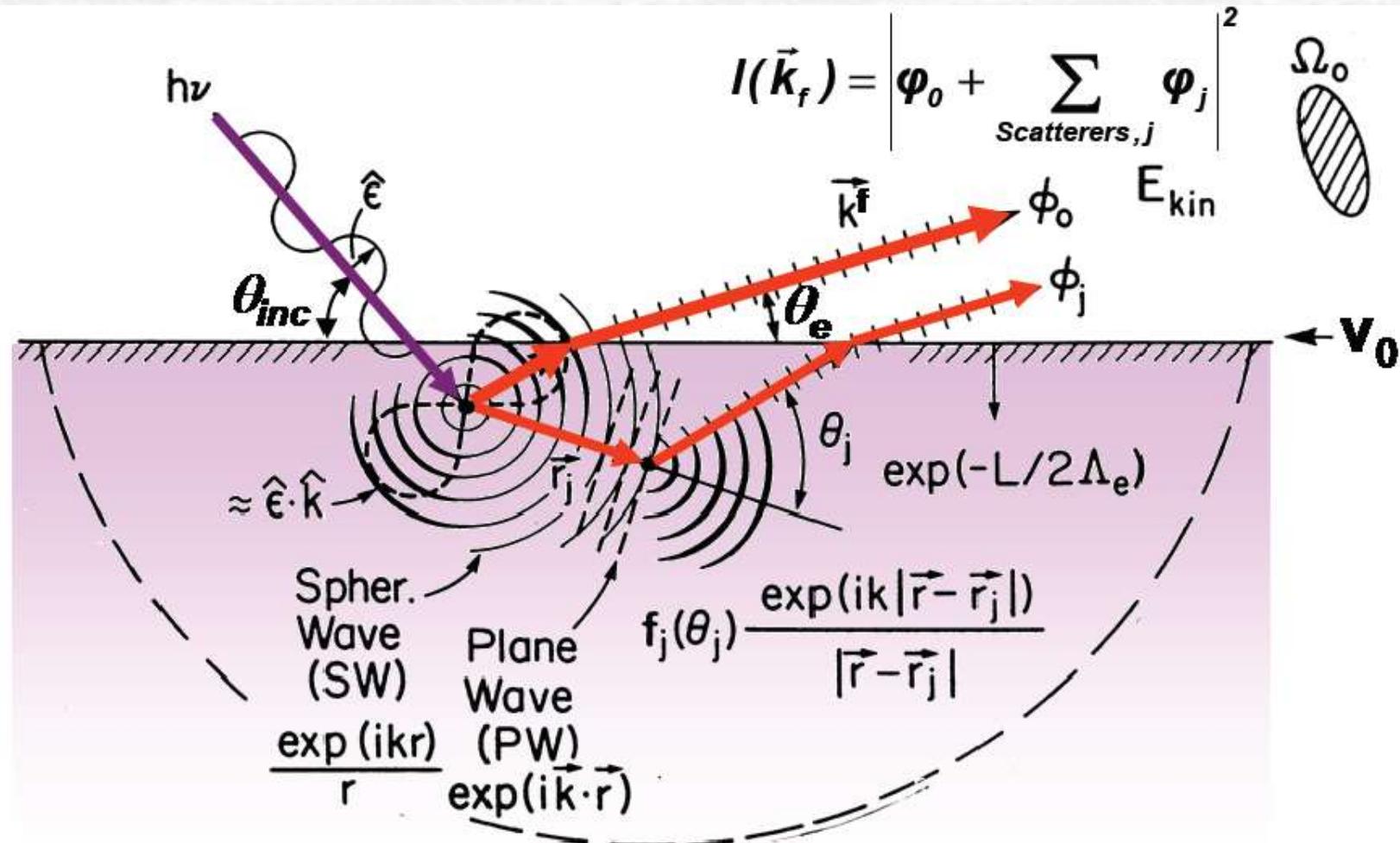
F. Heiser et al.

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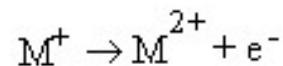
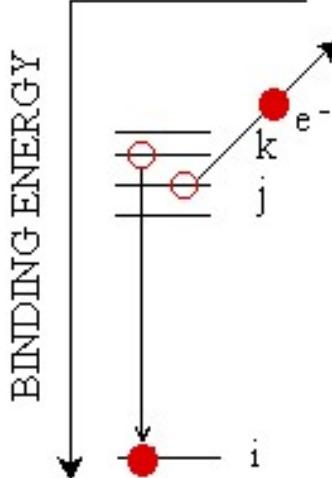
# Application to surfaces



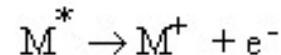
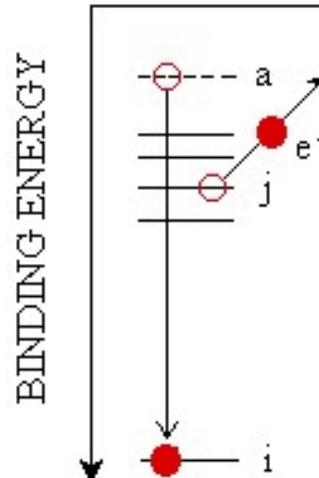
# Core hole relaxation

## SECONDARY PROCESSES

### AUGER ELECTRON DECAY



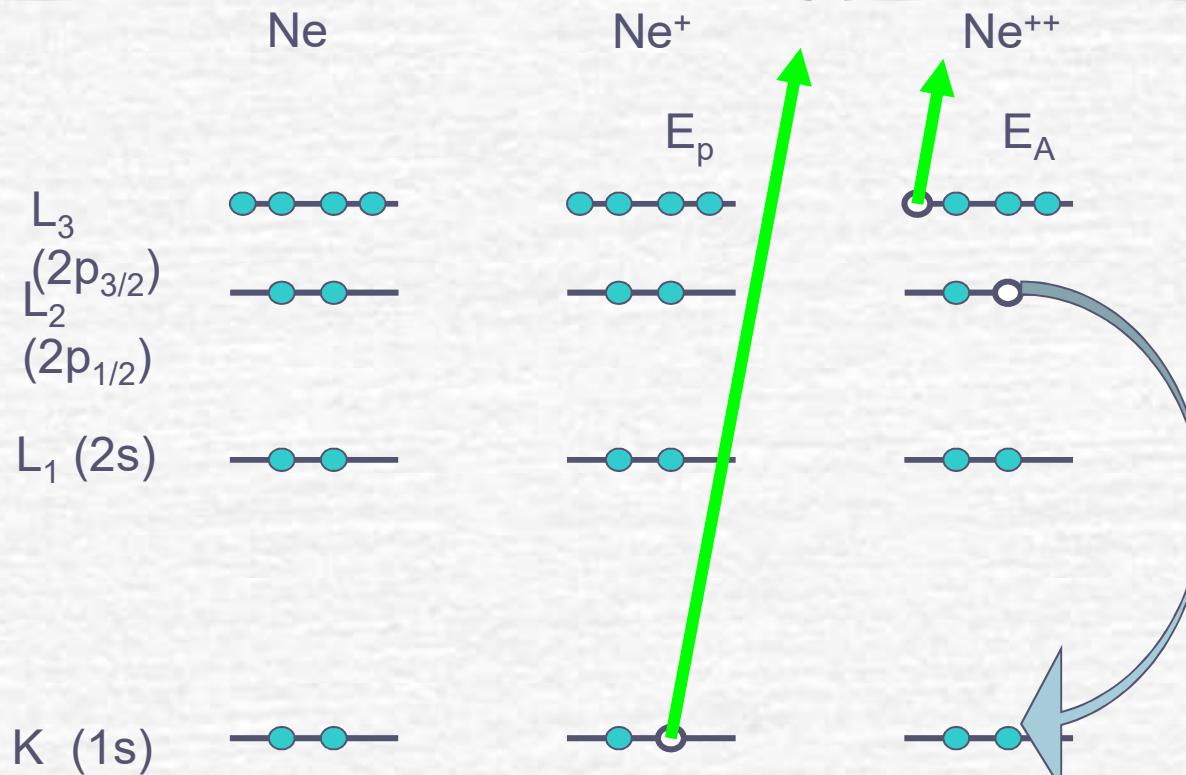
### AUTOIONIZATION



Global energy, angular  
Momentum and parity

Energy, angular momentum,  
Dipole selections at each step

# Auger decay

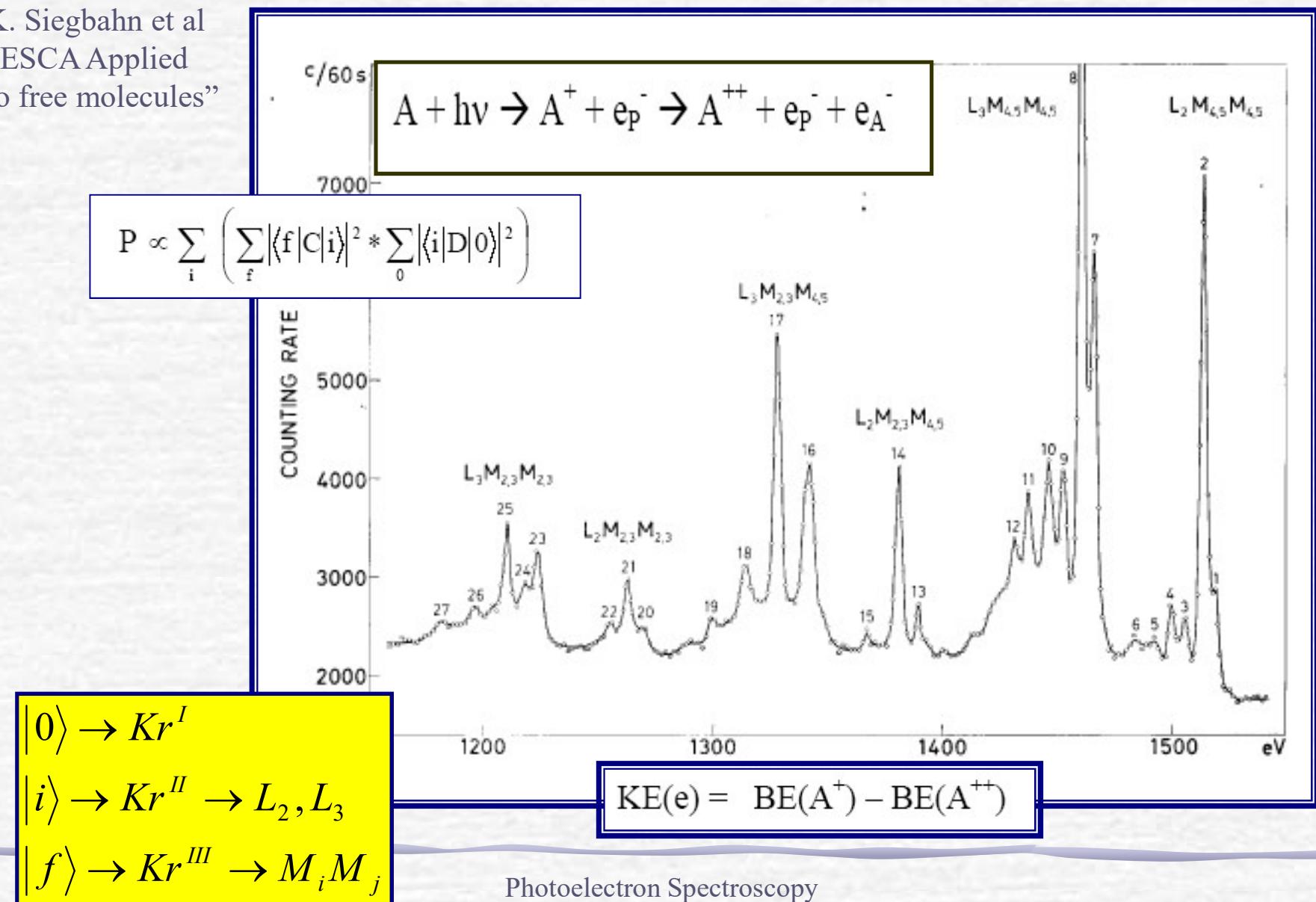


| Auger Transition                   | Double ion valence configuration | Multiplet Terms   |
|------------------------------------|----------------------------------|---|
| KL <sub>1</sub> L <sub>1</sub>     | 2s <sup>0</sup> 2p <sup>6</sup>  | <sup>1</sup> S <sub>0</sub>   |
| KL <sub>1</sub> L <sub>2,3</sub>   | 2s <sup>1</sup> 2p <sup>5</sup>  | <sup>1</sup> P <sub>1</sub> , <sup>3</sup> P <sub>0</sub> , <sup>3</sup> P <sub>2</sub> , <sup>3</sup> P <sub>1</sub> |
| KL <sub>2,3</sub> L <sub>2,3</sub> | 2s <sup>2</sup> 2p <sup>4</sup>  | <sup>1</sup> S <sub>0</sub> , <sup>3</sup> P <sub>0</sub> , <sup>3</sup> P <sub>2</sub> , <sup>1</sup> D <sub>2</sub> |

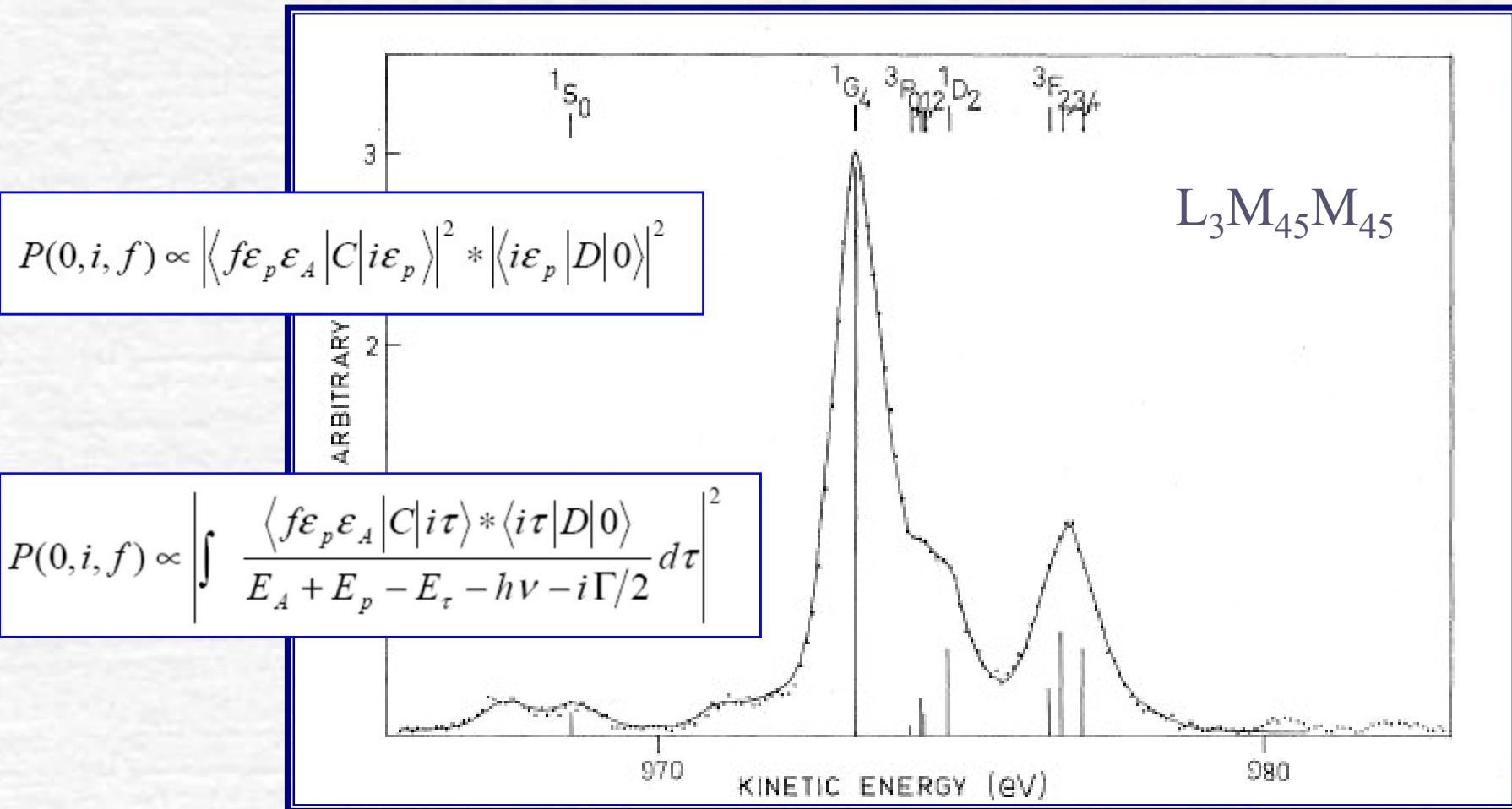
$$EA(KL_1L_2) = E(K) - E(L_1) - E(L_2, L_1)$$

# Kr Auger spectrum

K. Siegbahn et al  
“ESCA Applied  
to free molecules”

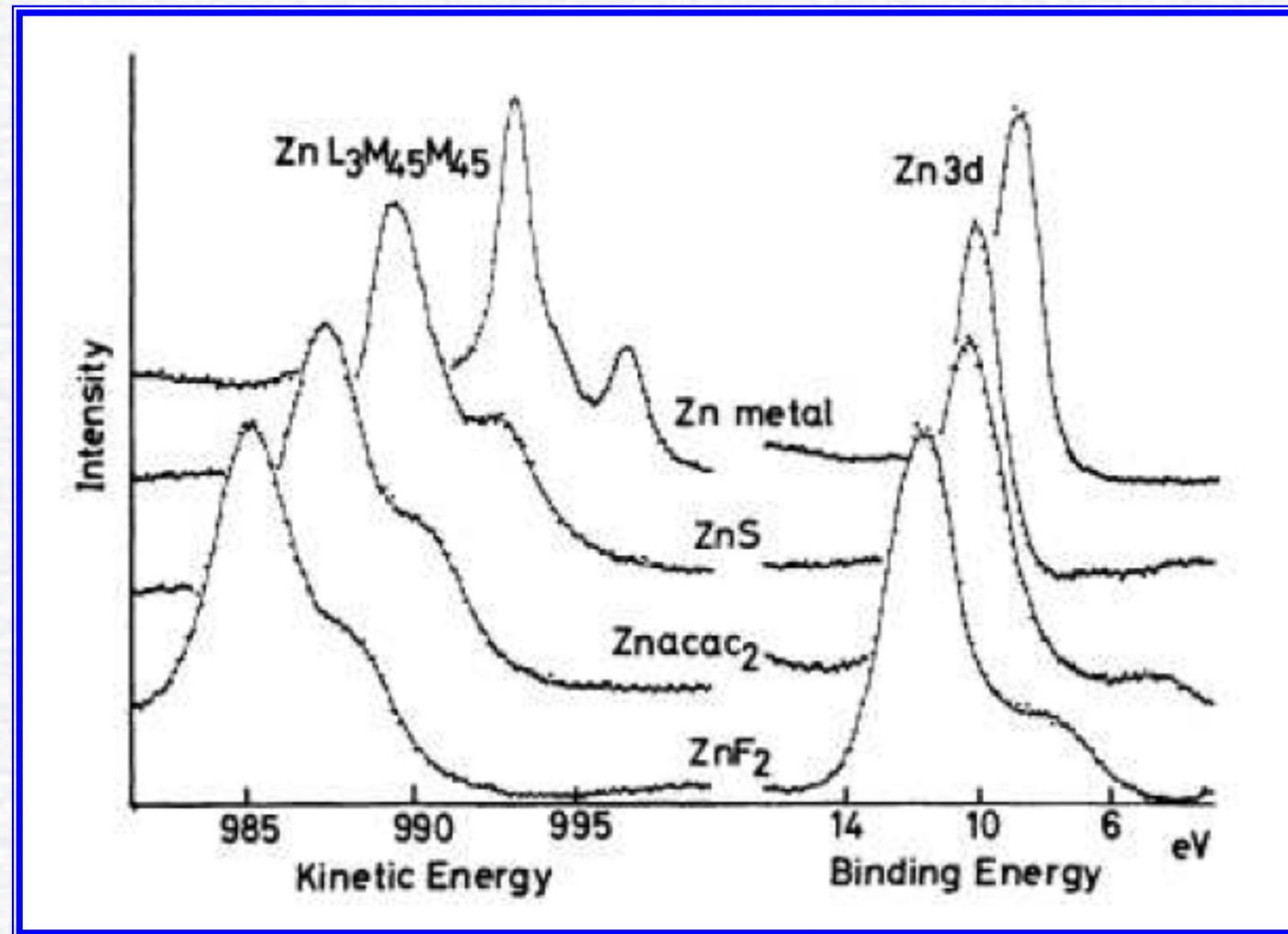


# Zn Auger multiplet splitting

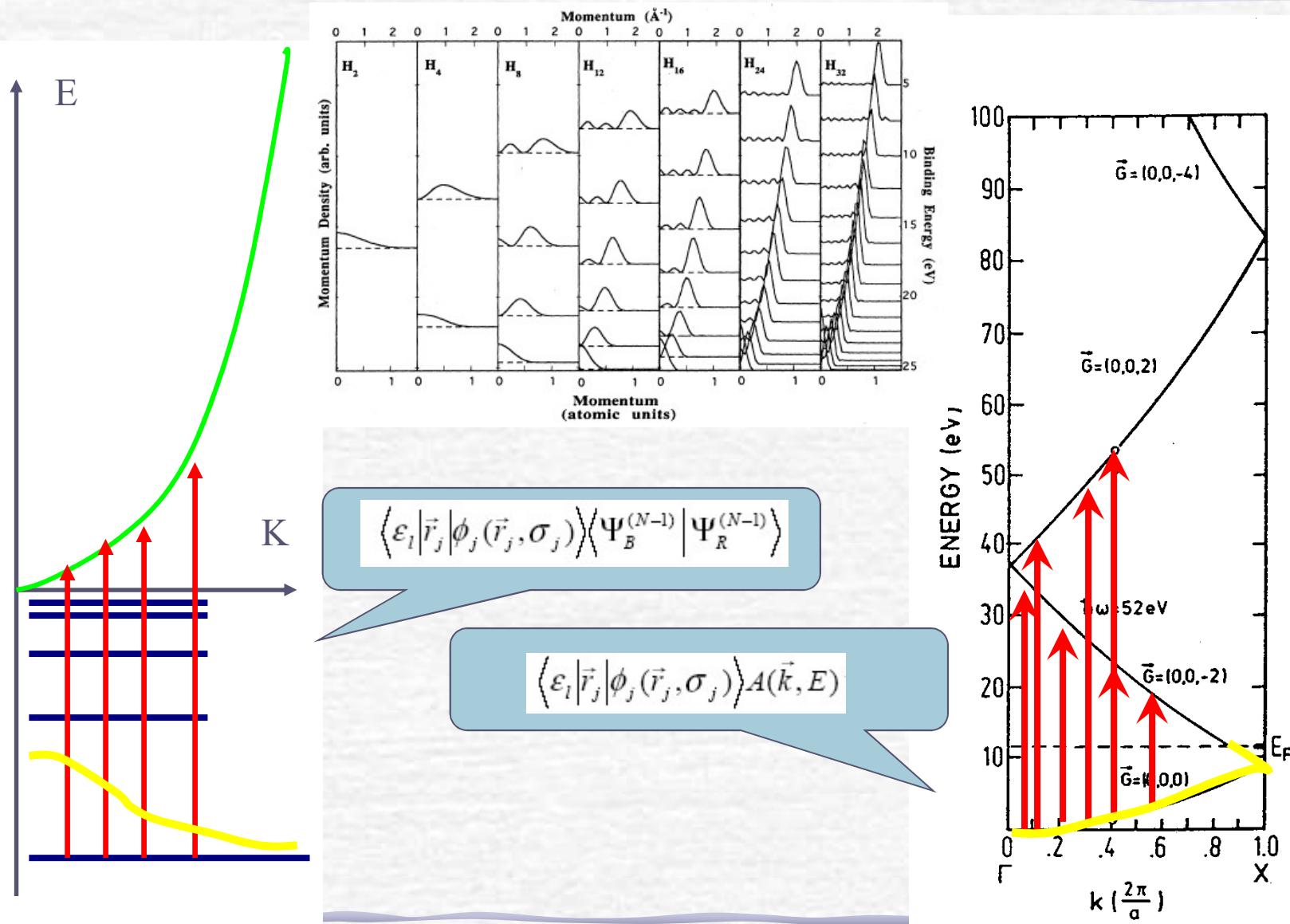


Aksela et al. PRL 33,999 (1974)

# Auger chemical shift

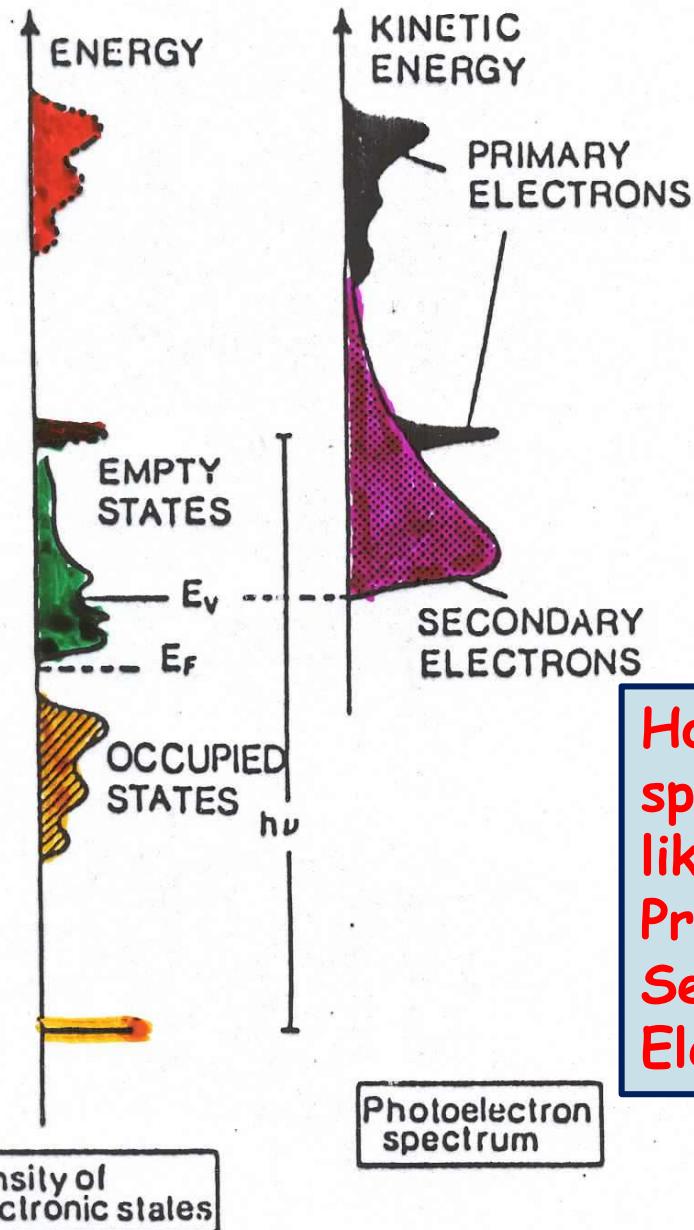
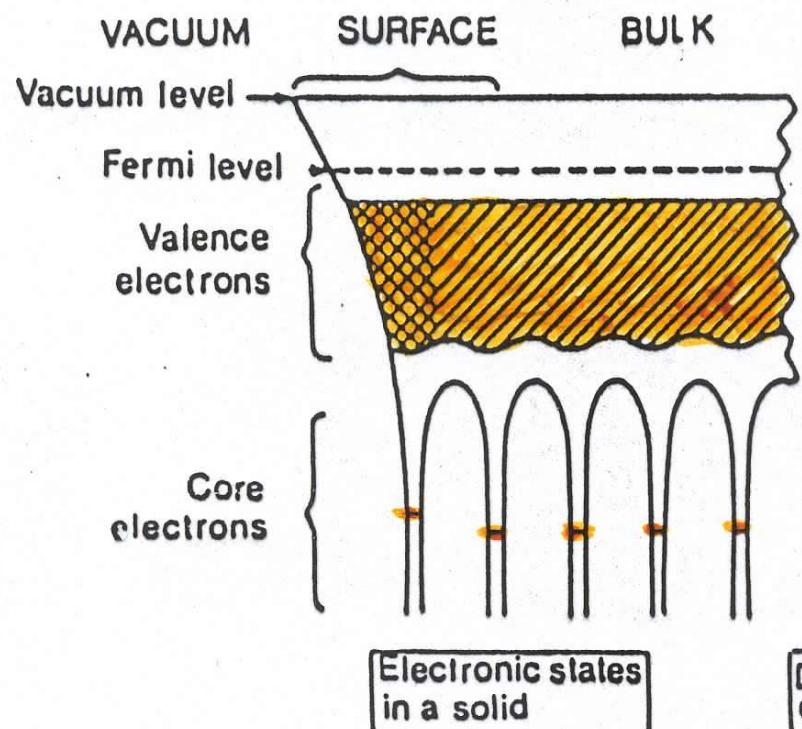


# From central to periodic potential



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# Photoemission Spectroscopy In Solids



How real spectra look like:  
Primary and Secondary Electrons

# Spectral Function in Interacting Solids

$$\Psi_{Nf} = A \Psi^{N-1} f \Phi_{kf}$$

$$J_e \propto \sum_{if} |M_{if}^2| \sum_m |m_{im}|^2 \times \delta(E_i^N + h\nu - E_{m-kin}^{N-1})$$

$$A(k, \varepsilon) = \sum_m (| \langle \Psi_{m-kin}^{N-1} | c_k | \Psi_i^N \rangle |^2 \times \delta(\varepsilon + E_{m-kin}^{N-1} - E_i^N))$$

$$J_e(k, \omega) \propto \sum_{if} |M_{if}|^2 A(k, E_{kin} - h\nu) f(E_{kin} - h\nu)$$

For non interacting particles

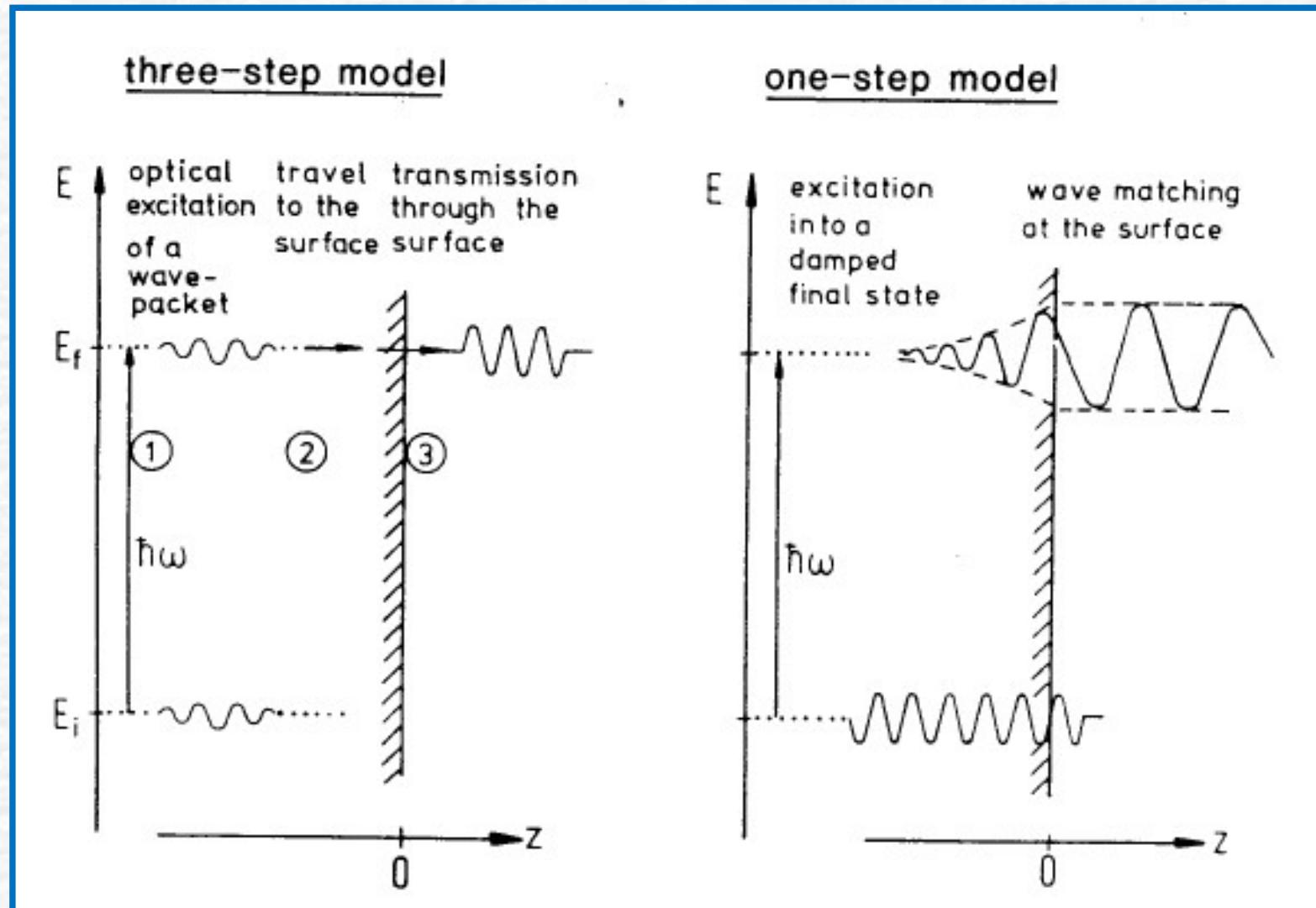
$$A(\varepsilon, k) = \delta(\varepsilon - E_k)$$

where

$$E_k = E_i^{N-1} - E_i^N$$

$$A(\varepsilon, k) = 1/\pi |\Sigma''(k, \varepsilon)| / [|\varepsilon - E_k - \Sigma'(k, \varepsilon)|^2 + |\Sigma''(k, \varepsilon)|^2]$$

# The Three-Step model 1



# The three-step model 2

## 1. Dipole transition

$$J_e = \sum_{if} f(E_i) [1-f(E_f)] M_{if}^2 \times \delta[E_{kin} - (E_f - \Phi)] \delta(E_f - E_i - h\nu) \delta(k_i + G - k_f)$$

## 2. Elastic transport

$$d(E_f, k) = \alpha\lambda/(1+\alpha\lambda)$$

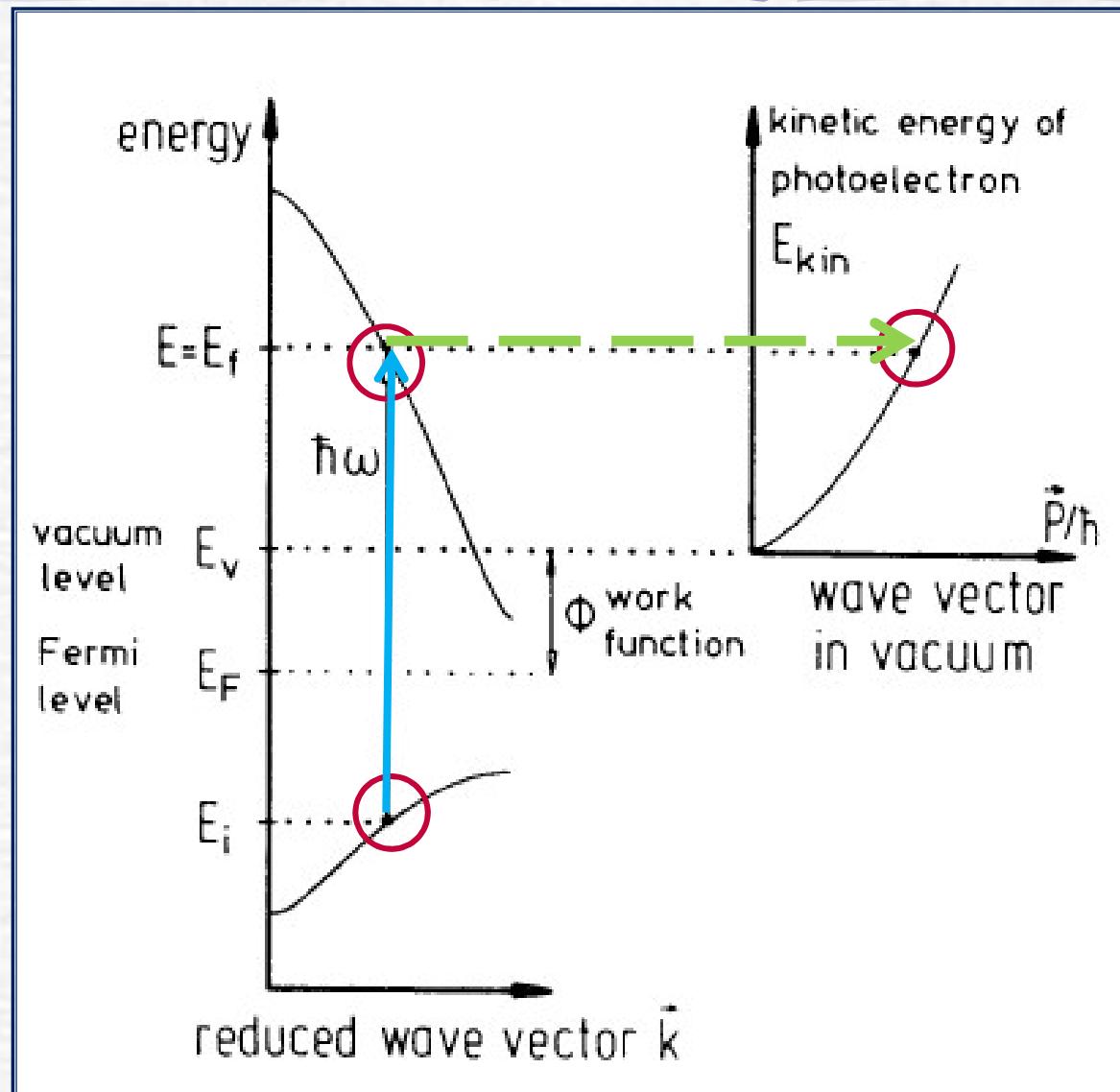
## 3. Exit to vacuum

$$T(E_f, k_{ext}) = \begin{cases} 0 & \text{if } E_f < E_F + \Phi \\ 1/2 \sqrt{[1 - (E_F + \Phi)/E_f]} & \text{if } E_f > E_F + \Phi \end{cases}$$

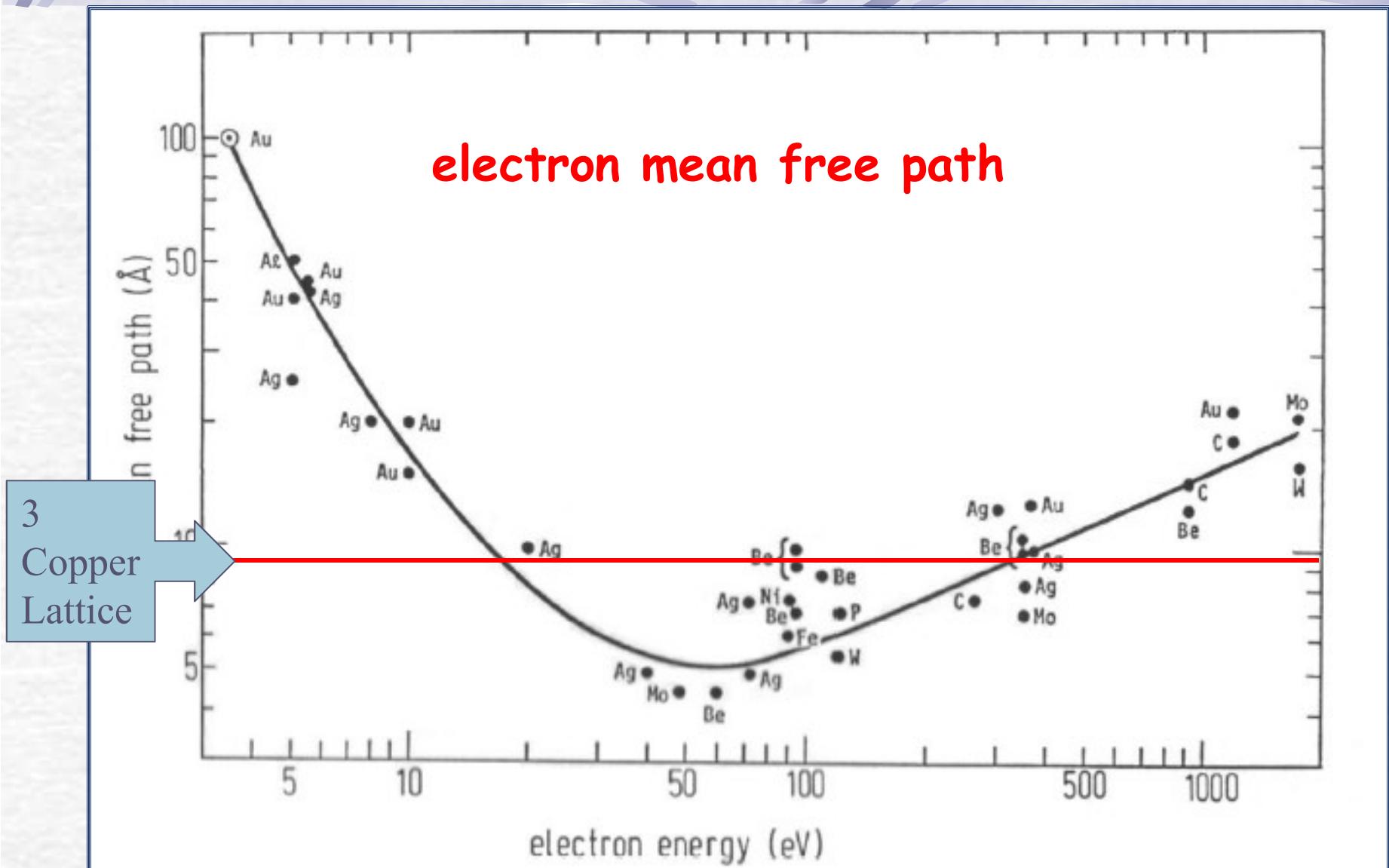
## Total current

$$J_e \propto \sum_{if} f(E_i) [1-f(E_f)] M_{if}^2 \times T(E_f, k_{ext}) \times d(E_f, k) \times \delta[E_{kin} - (E_f - \Phi)] \delta(E_f - E_i - h\nu) \times \delta(k_i + G - k_f) \times \delta(k''_i + G'' - k''_{ext})$$

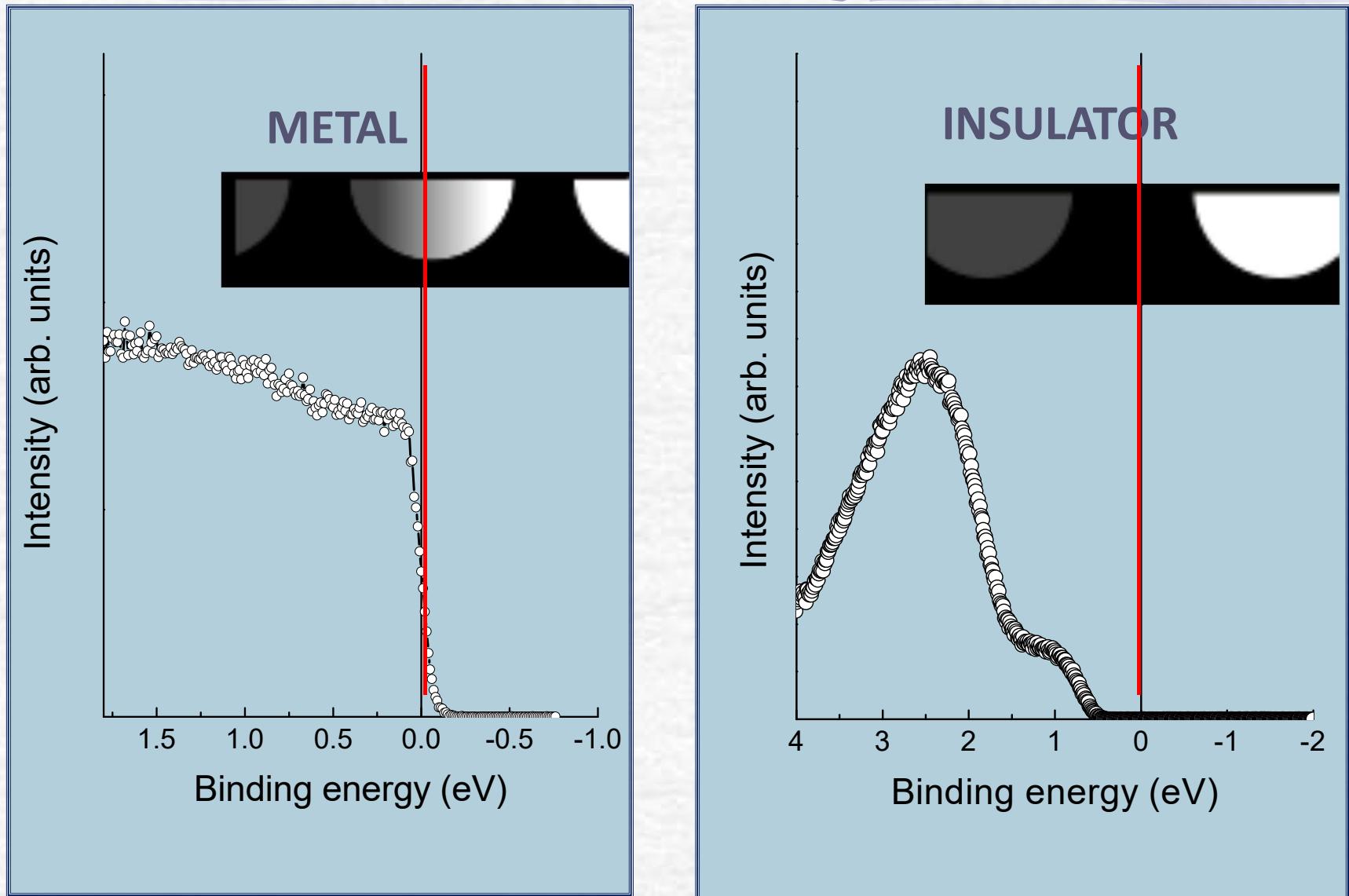
# How to reconstruct the initial state in 3 step model



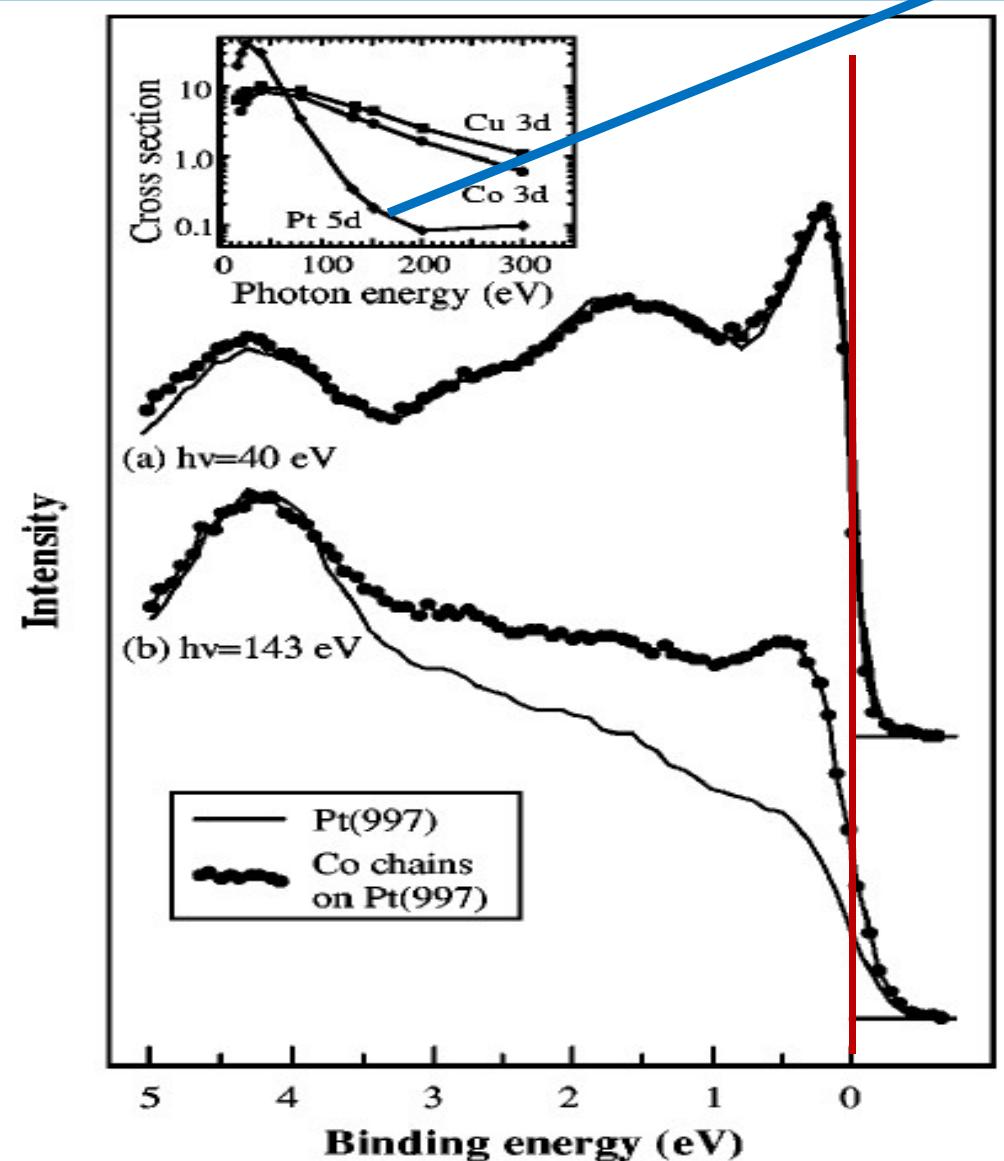
# Surface vs. Bulk Sensitivity in Photoemission



# Photoemission spectrum at Fermi Edge

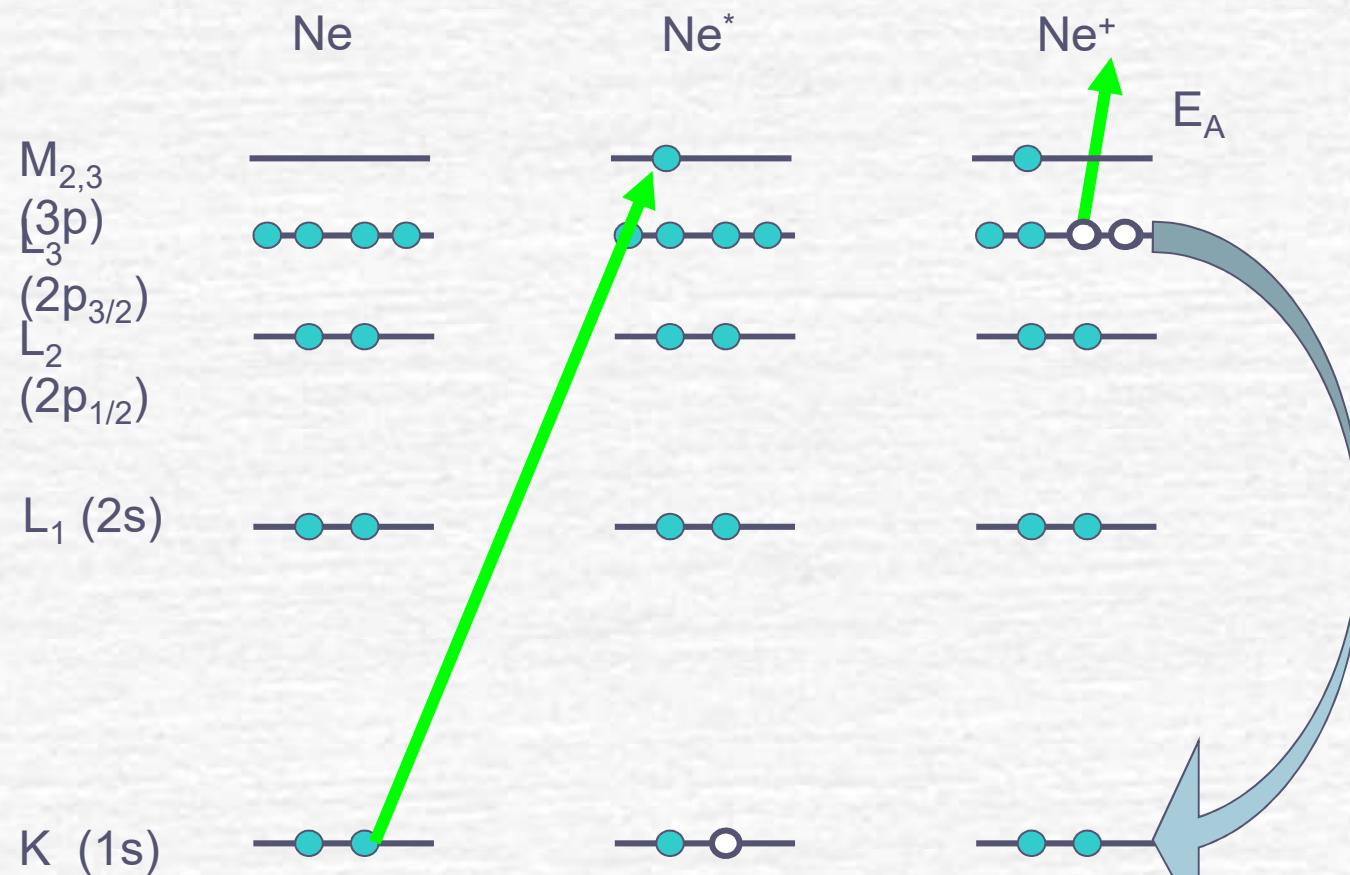


# Valence band Energy Distribution Curves and Cooper minimum

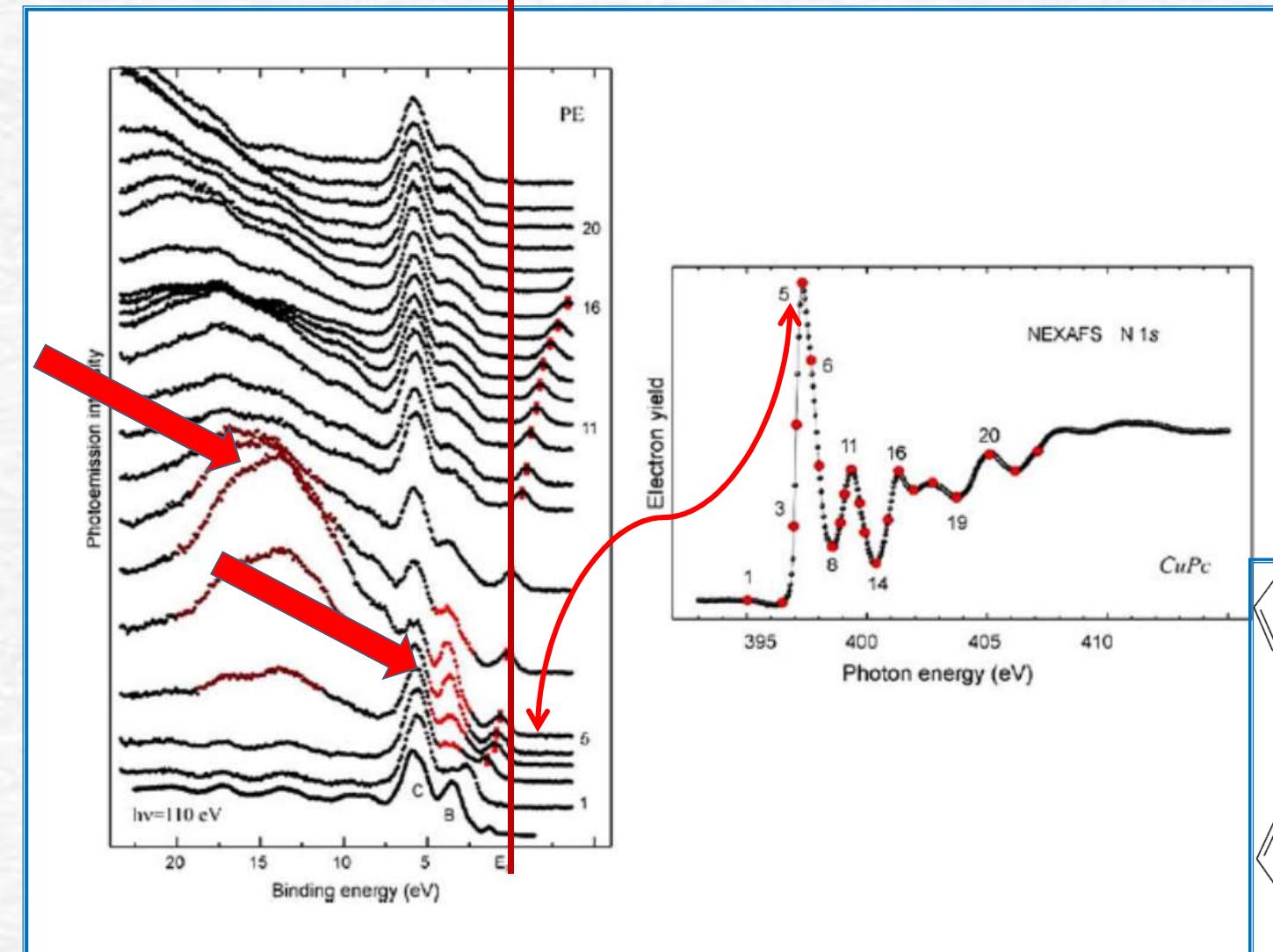


Valence Band EDCs of the clean Pt(997) surface (thin lines) and of Co-nanowires grown on Pt(997) (dots and thick lines), taken at different photon energies

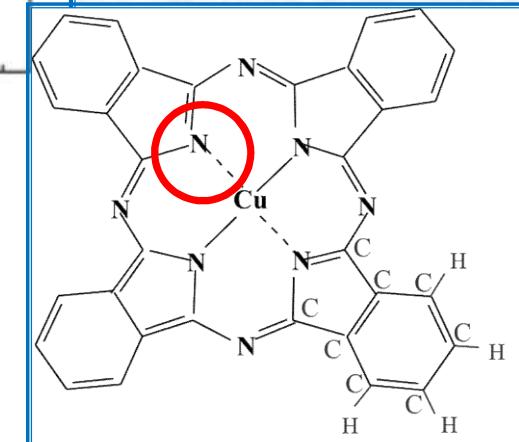
# Autoionizing decay



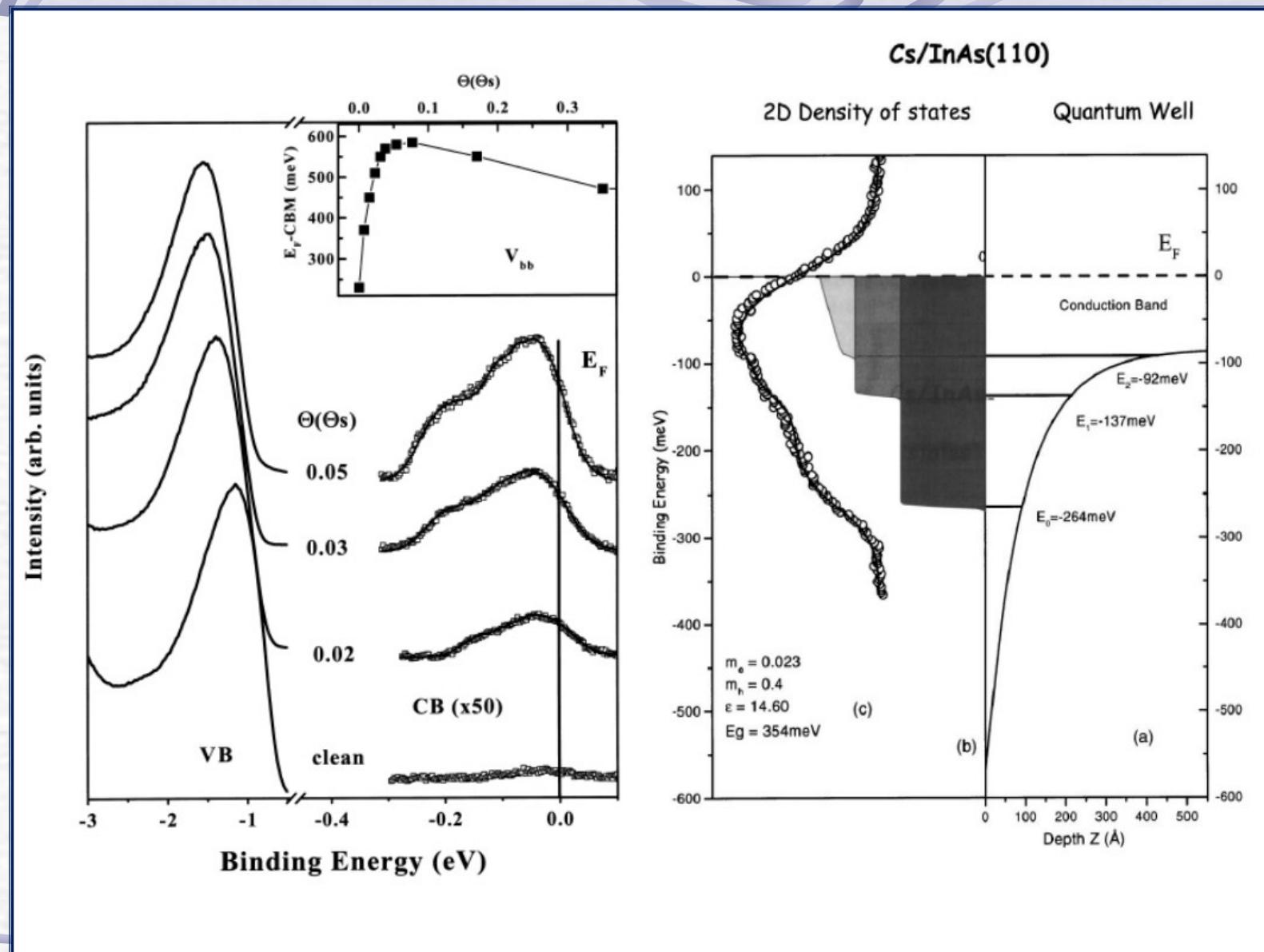
# Resonant photoemission EDC



Valence Band EDCs of a CuPc thin-film taken at different photon energies (left panel) and X-ray Absorption Spectroscopy (XAS) from the same CuPc across the N K-edge (right panel).



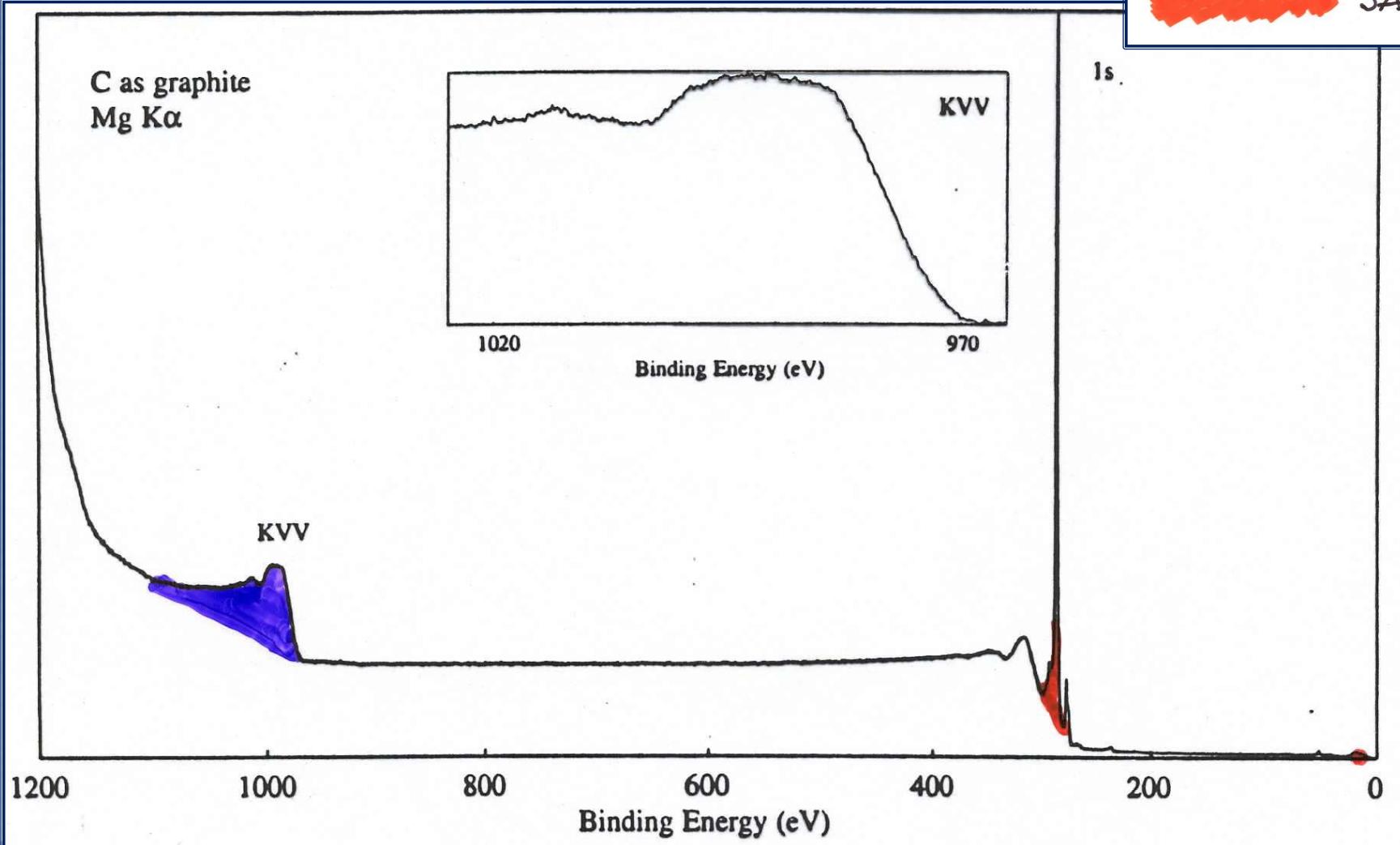
# 2D electron gas spatially confined Cs/InAs(110)



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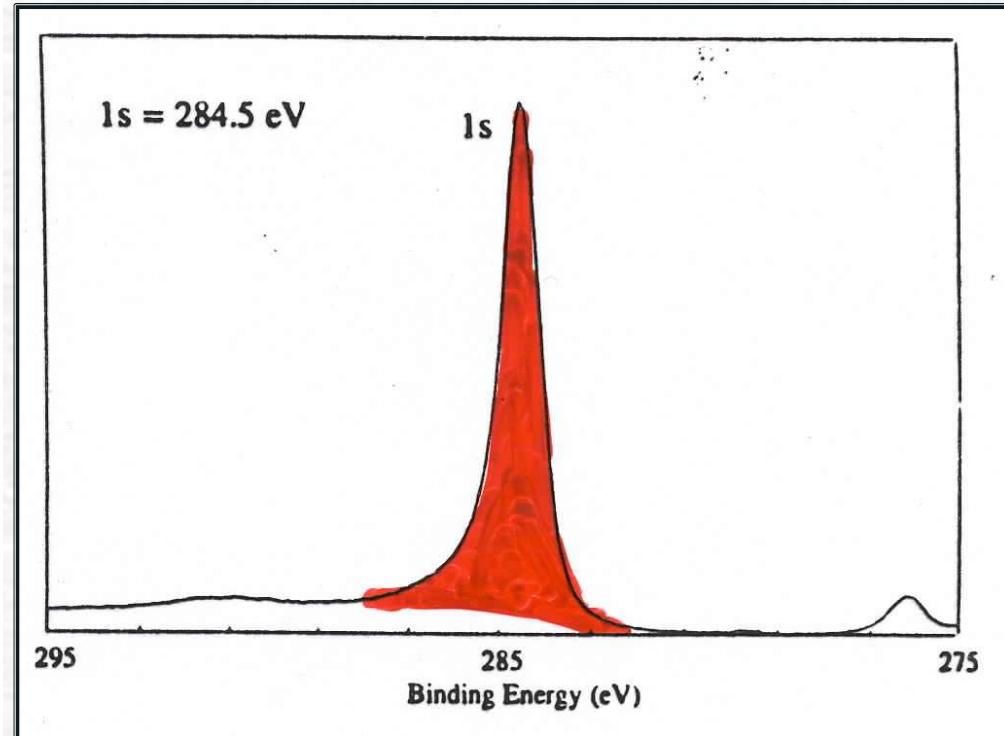
## Photoemission - Core Level Spectroscopy

### Wide XPS spectrum of graphite (C)



# Photoemission Spectroscopy

Core level XPS spectrum  
of graphite (C)



The singlet **C 1s** line is characterized by:

- 1) A specific binding energy which reflects the specific atomic species (C) in a specific chemical environment
- 2) A finite width reflecting the instrumental resolution, lifetime broadening and other many-body effects

## Core Levels, chemical shift

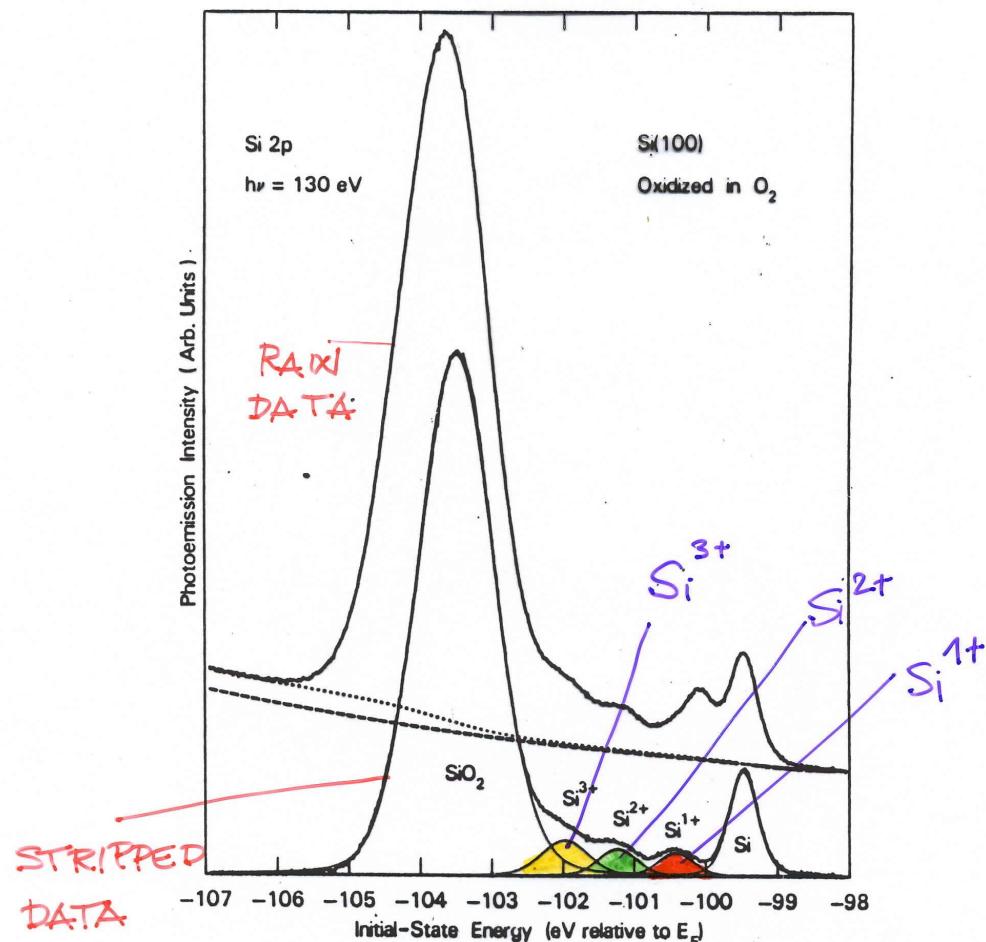
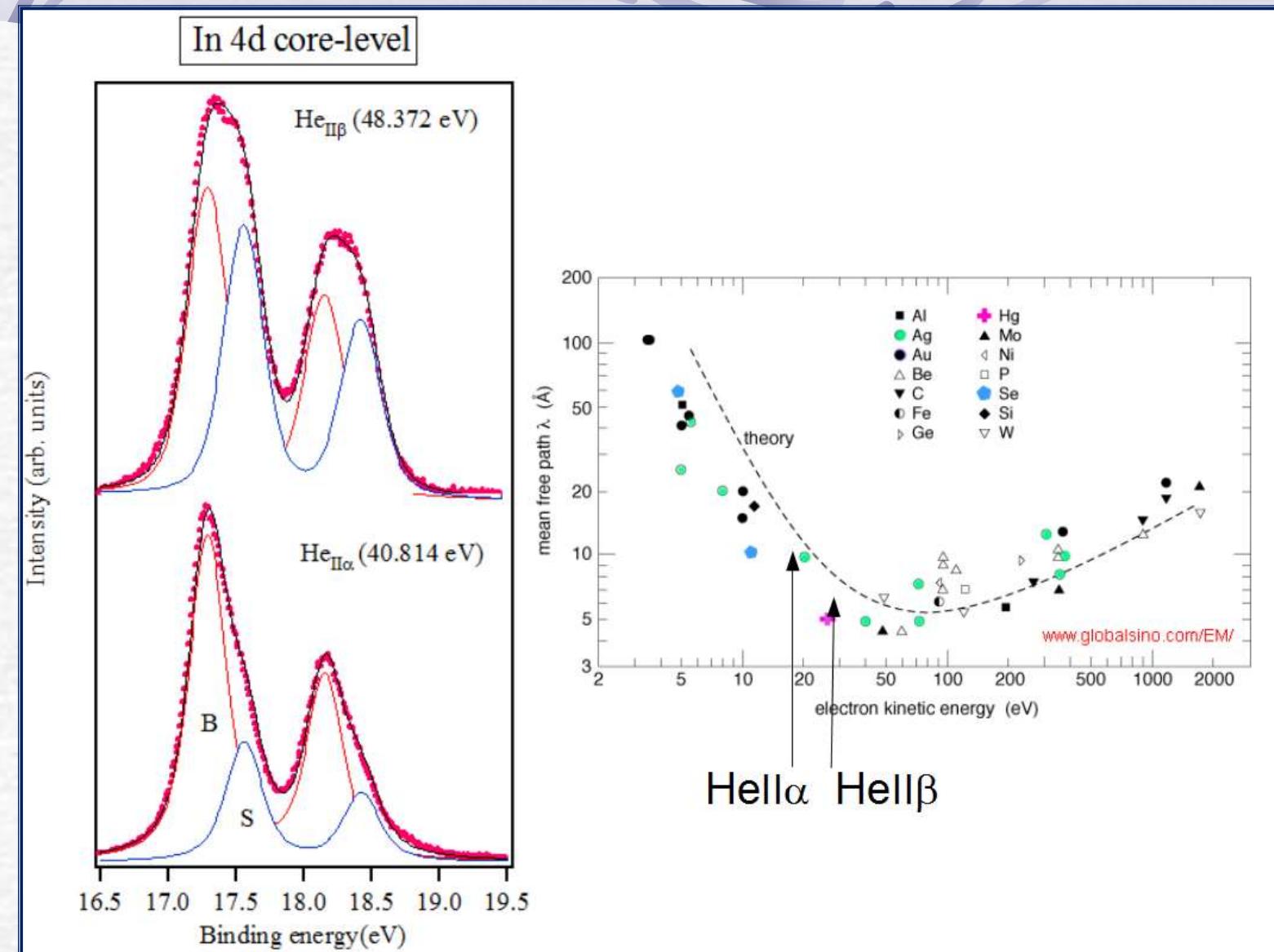


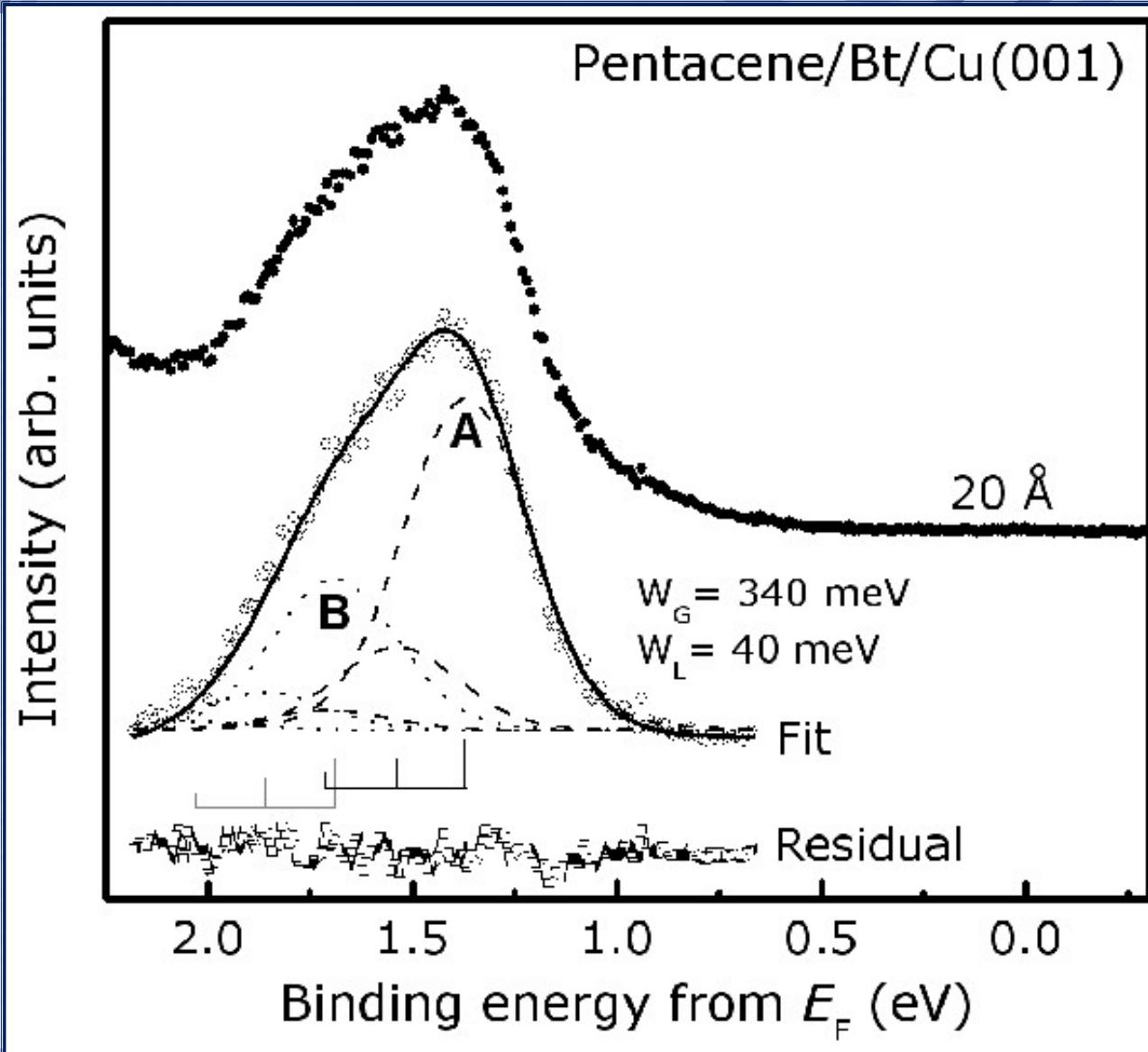
FIG. 1. Intermediate-oxidation states at the  $\text{SiO}_2/\text{Si}(100)$  interface, identified by their  $\text{Si} 2p$  core-level shifts. The top curve represents the raw photoemission data for the  $\text{Si} 2p_{1/2,3/2}$  core levels. The bottom curve has the  $\text{Si} 2p_{1/2}$  line and the secondary electron background subtracted. All three intermediate-oxidation states are seen. For a truncated bulk structure only  $\text{Si}^{2+}$  would be present since the  $\text{Si}(100)$  surface has two broken bonds per atom.

The  $\text{Si} 2p$  line is characterized by the occurrence of 5 chemically distinct components which reflect different chemical states of the  $\text{Si}$  atoms at the interface

# Surface core level shift vs. mean free path In 4d



# Valence PE vibrational spectrum

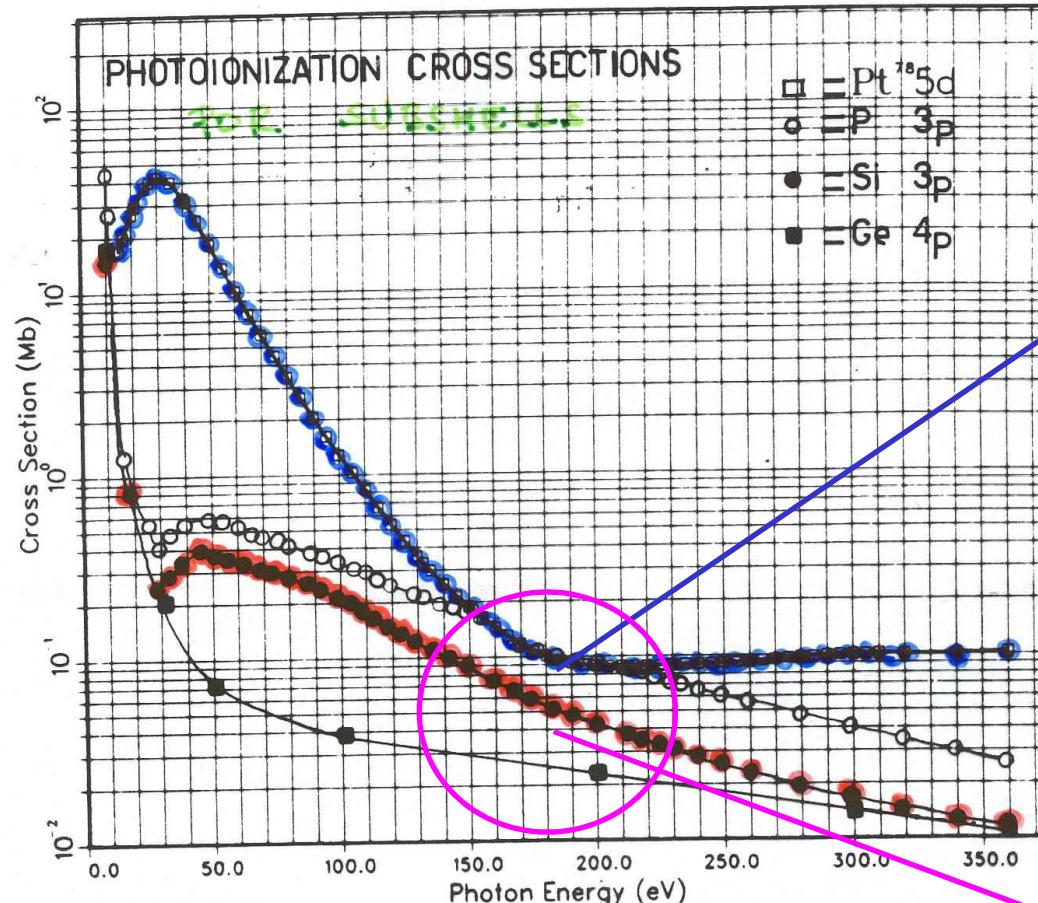


benzene-thiol:  
 $C_6H_5-SH$

Betti, Kanjilal and Mariani, J. Phys. Chem. A 111, 12454 (2007)

# Cooper Minimum Photoemission

It is possible when one of the valence band orbital shows a Cooper minimum in the photoionization cross section



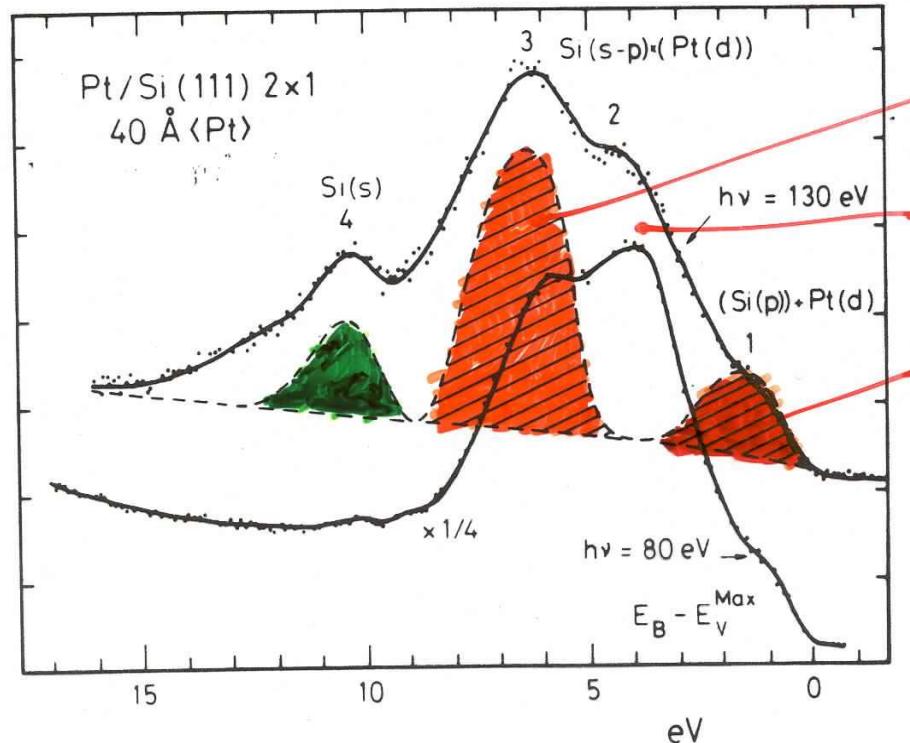
Photoionization cross section for 4d and 5d subshells in the energy range 0–200 eV compared to the cross sections for the 3ps and 4sp valence states of the semiconductors [21].

Cooper minimum  
in the Pt 5d  
cross section

A Cooper minimum exists when the radial part of the orbital wave function exhibits a node

The Pt 5d and Si 3p cross sections are comparable

# Cooper Minimum Photoemission

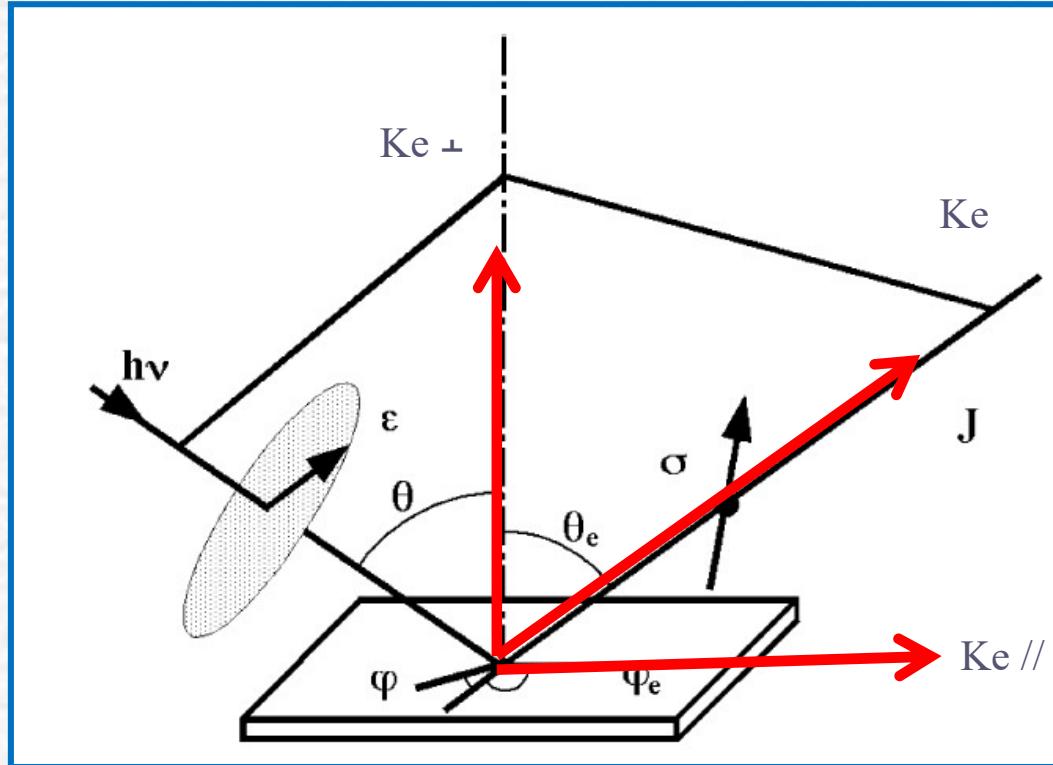


Analysis of the Si sp partial DOS at the Pt-Si(111) reacted interface (40 Å Pt-Si(111) at room temperature). The top panel displays the CM and the  $h\nu = 80$  eV photoemission data, and a three-peak partial DOS that accounts for the Si hybridized 3sp charge at the interface; a gap is present in correspondence to the localized Pt5d states. The same three-peak partial DOS is then self-convoluted and compared to the integrated Si L<sub>2,3</sub>VV lineshape. The correspondence of all peaks and relative intensities (a part of the known reduction of the Si 3s contribution) confirms the CM derivation of the Si sp partial DOS [159].

Pt 5d anti-bonding  
Pt 5d non-bonding  
Pt 5d bonding

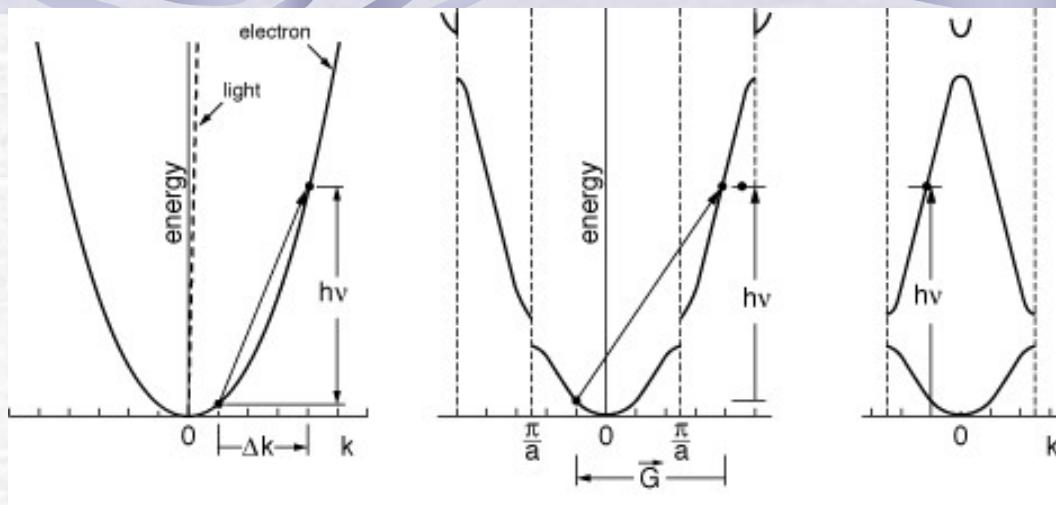
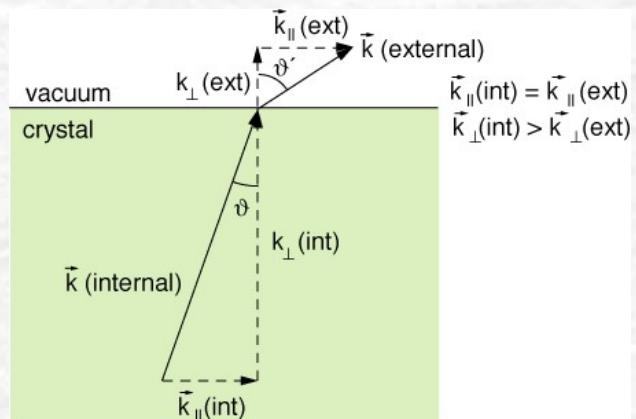
A joint analysis of VB photoemission spectra taken at and off the Cooper minimum enables one to disentangle the differing site- and orbital-specific contributions

# Angular resolved photoemission

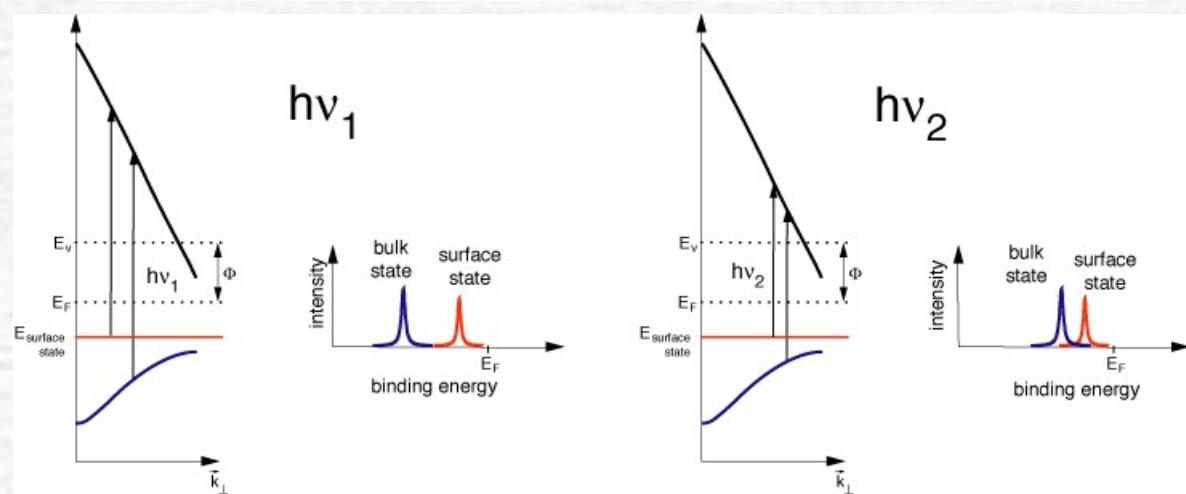


# angular resolved photoemission

How to photo-excite electrons from a crystal in a periodic potential, by photoemission



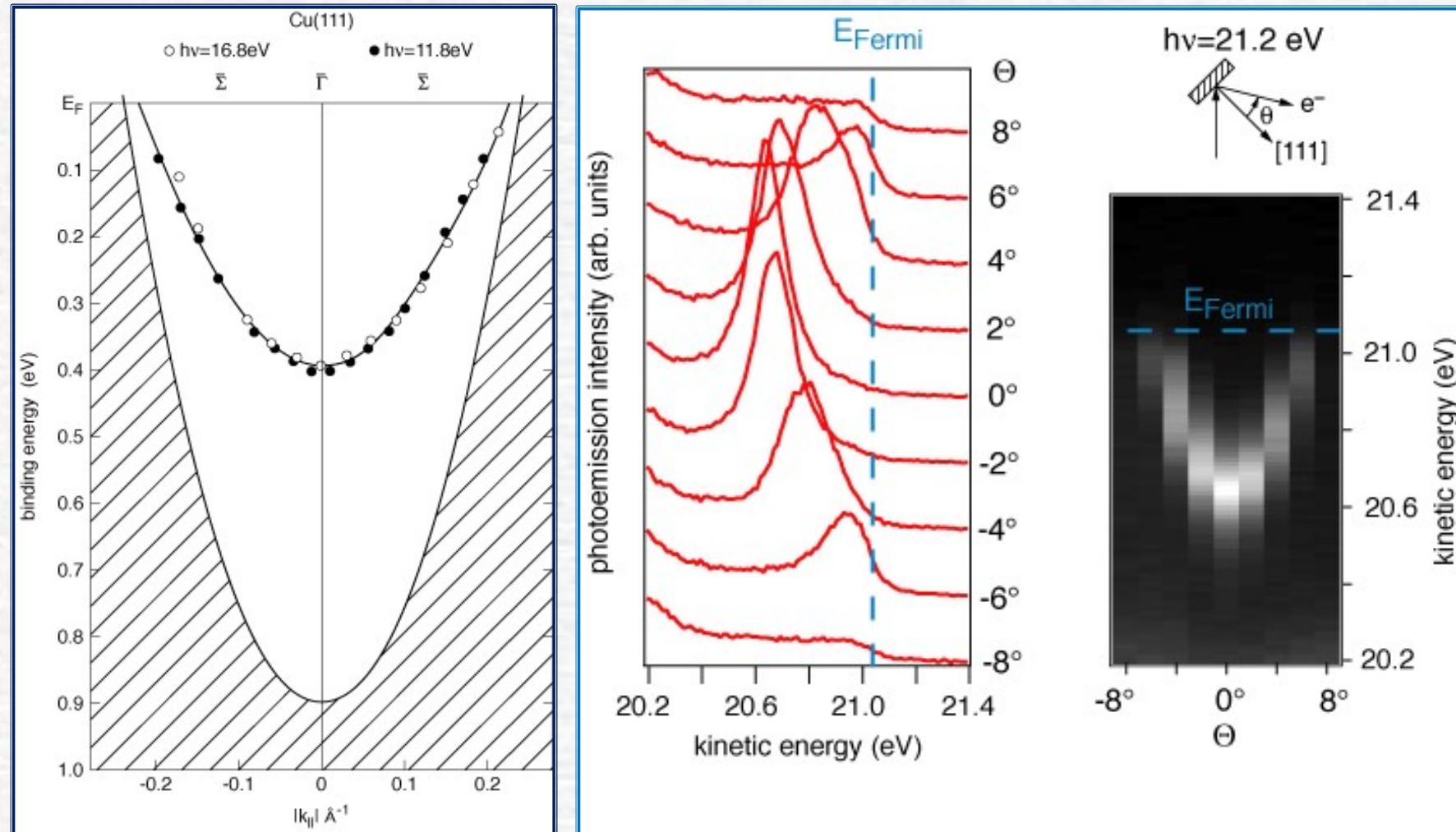
graphs from Ph. Hoffmann



$$\vec{k}_{||i} = \vec{k}_{||f} = \sin(\Theta) \sqrt{\frac{2m}{\hbar^2}} \sqrt{E_{kin}} = \sin(\Theta) \sqrt{\frac{2m}{\hbar^2}} \sqrt{h\nu - E_{bin} - e\Phi}$$

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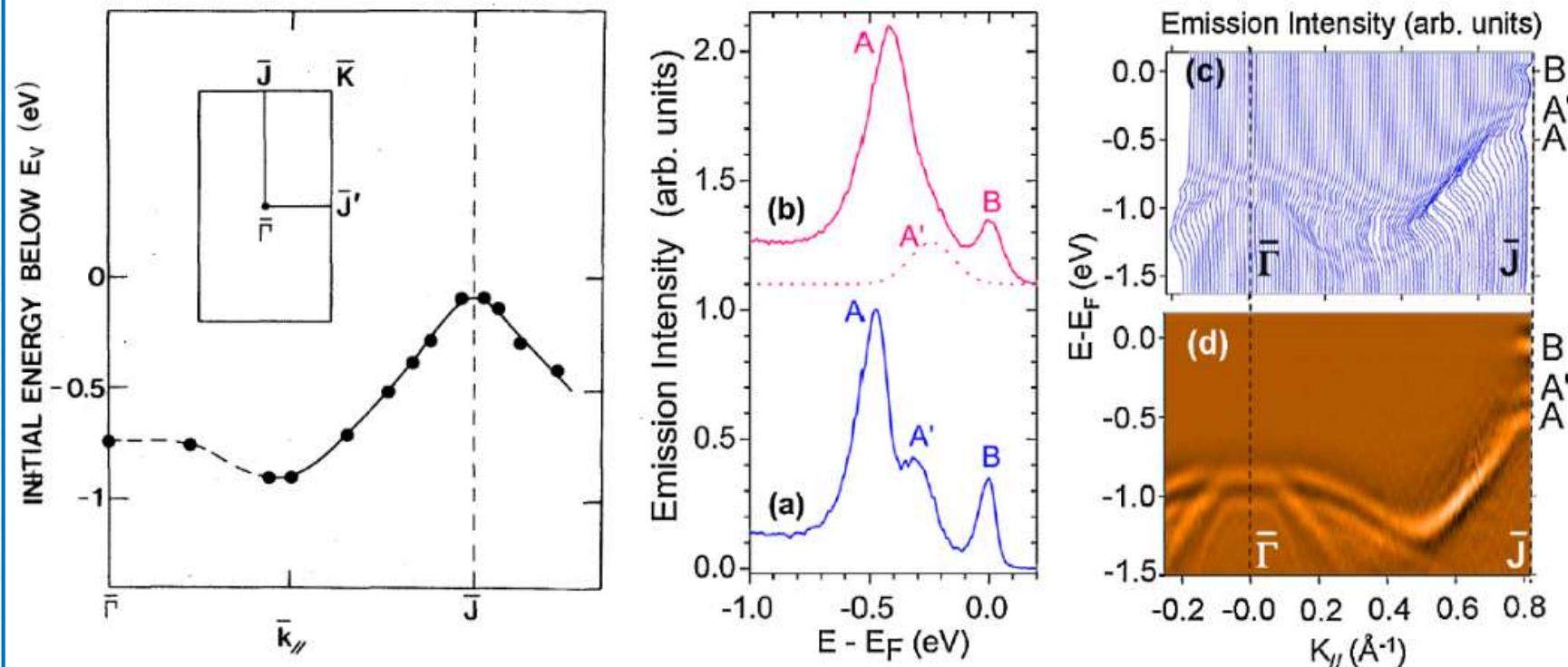
# electronic surface states at Cu(111)



quasi-free electron surface state on Cu(111),  
Schockley state, *s-like*

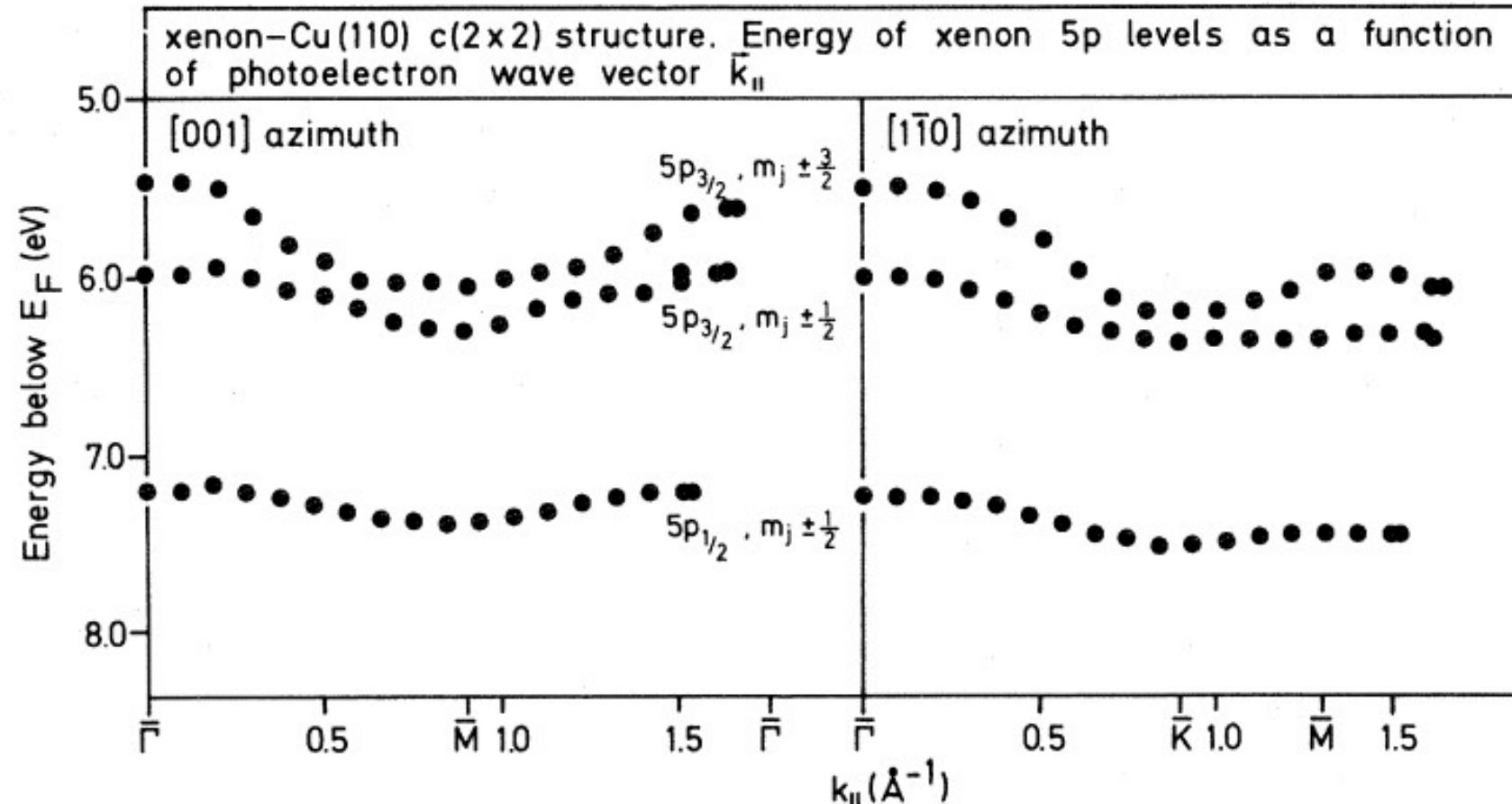
S.D. Kevan, Phys. Rev. Lett. 50, 526 (1983).

# Dangling bonds Si(111)-(2x1)



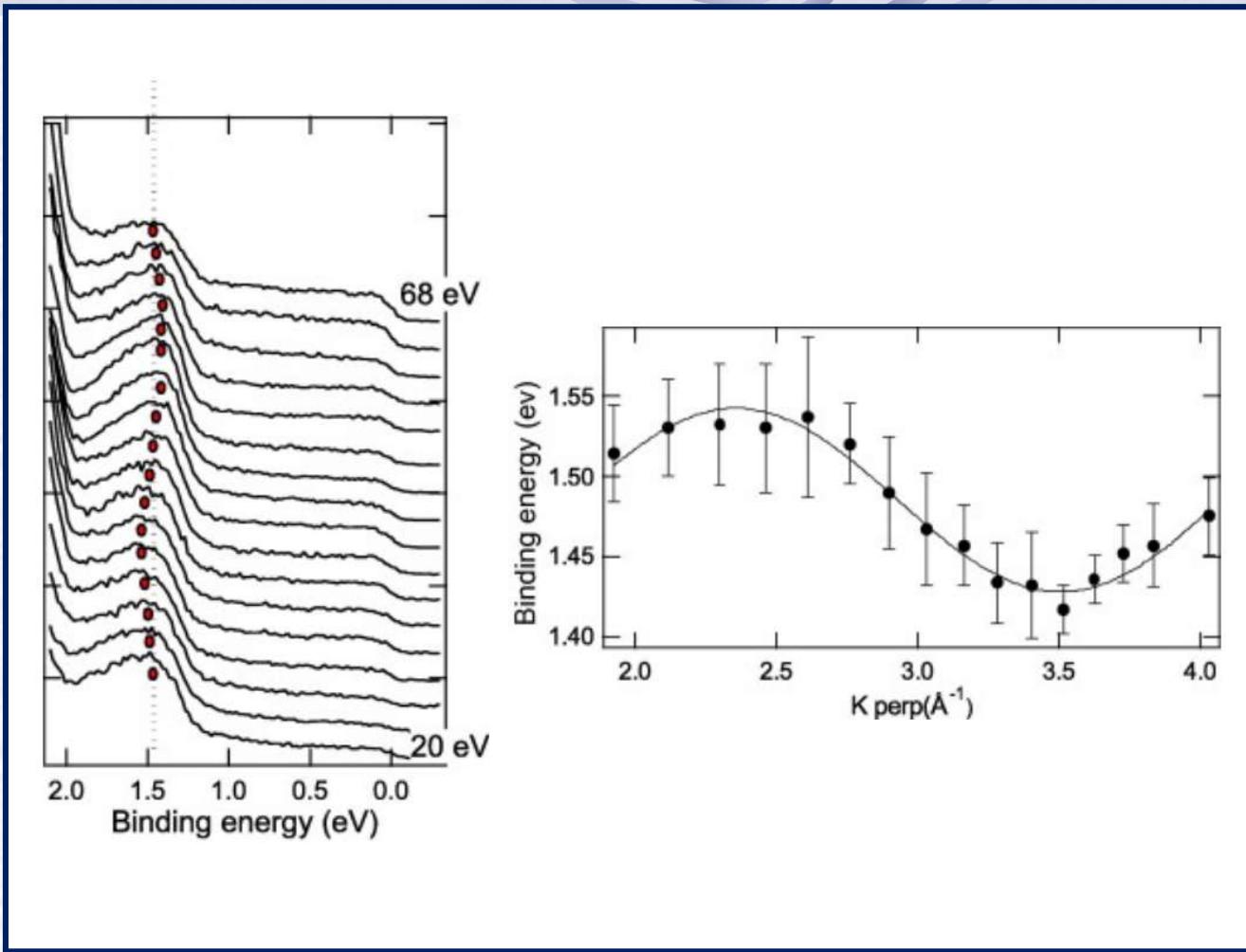
Dangling-bond surface state dispersion at the Si(111)-(2x1) reconstructed surface along the  $\bar{\Gamma}\bar{J}$  direction of the Surface Brillouin Zone (SBZ). One of the first experimental ARPES dangling-bond dispersion (left panel); recent high-resolution ARPES dangling-bond dispersion.

# Physisorbed Xe c(2x2)/Cu(110)

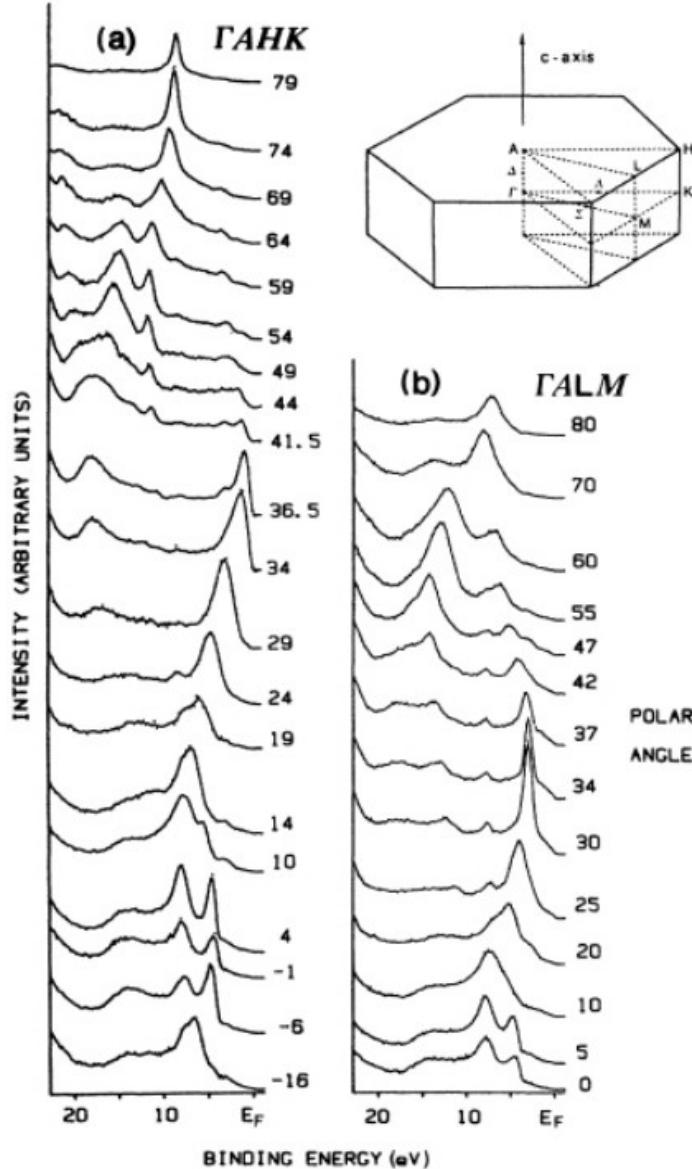


Experimental band structure of the 5p levels of Xe physisorbed in an ordered c(2x2) structure onto the Cu(110) surface. ARPES bands

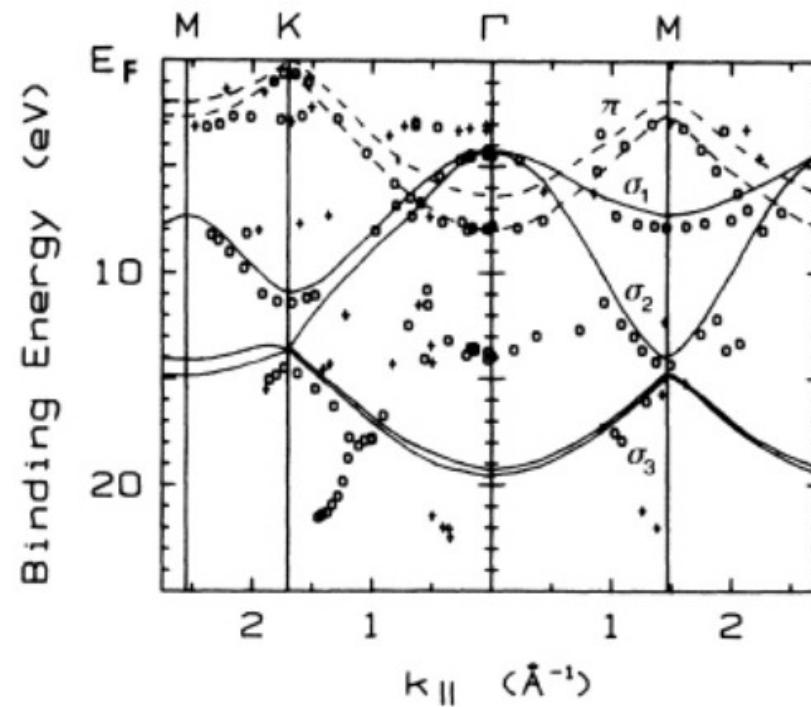
# Pentacene HOMO on Cu(119)



2-nm thick **pentacene** film grown on Cu(119). ARPES selection of spectra taken at **normal emission** and varying the photon energy (left); highest-occupied molecular-orbital (**HOMO**) band dispersion along  $k_{\perp}$  (right).

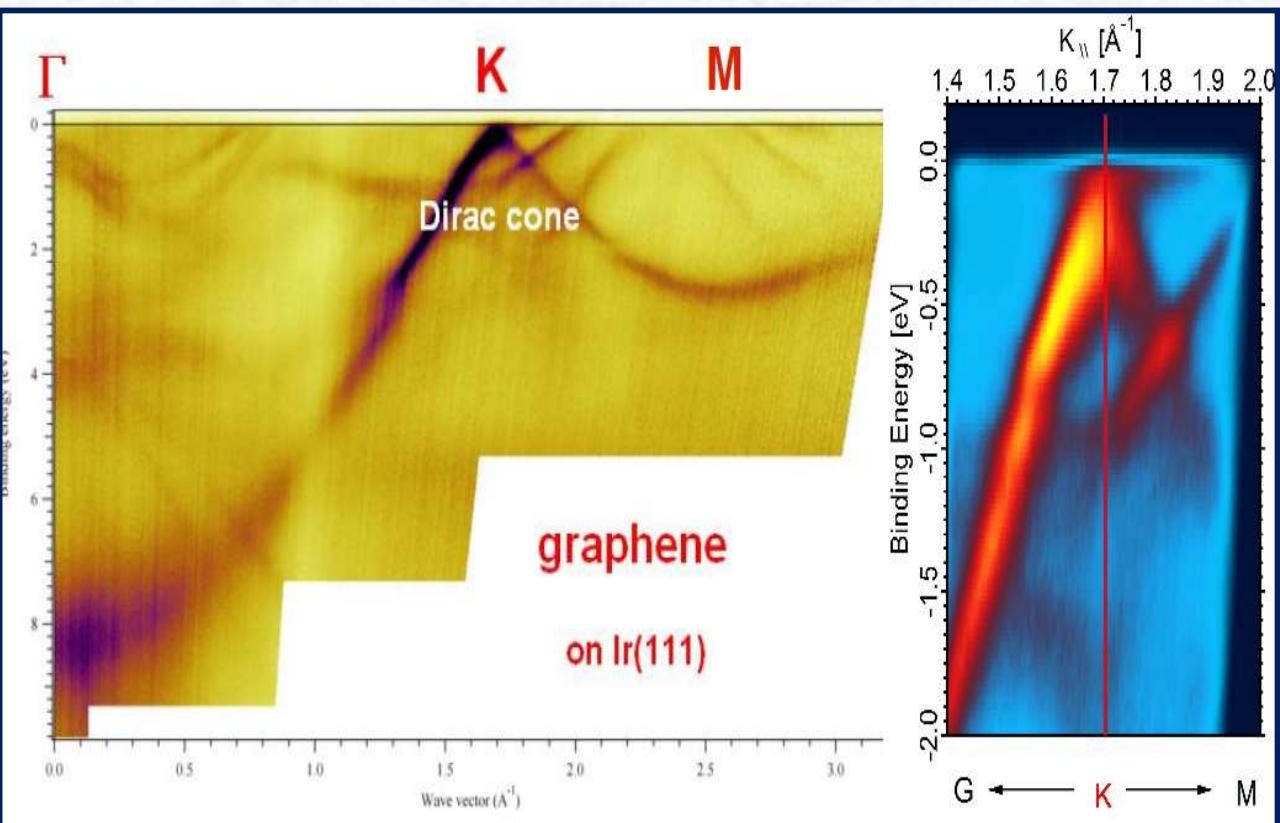
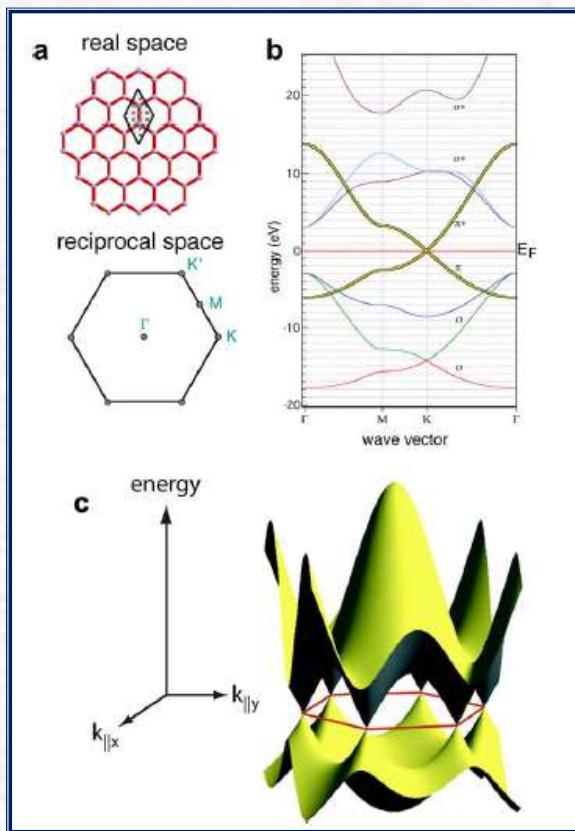


## ARPES graphite (HOPG)



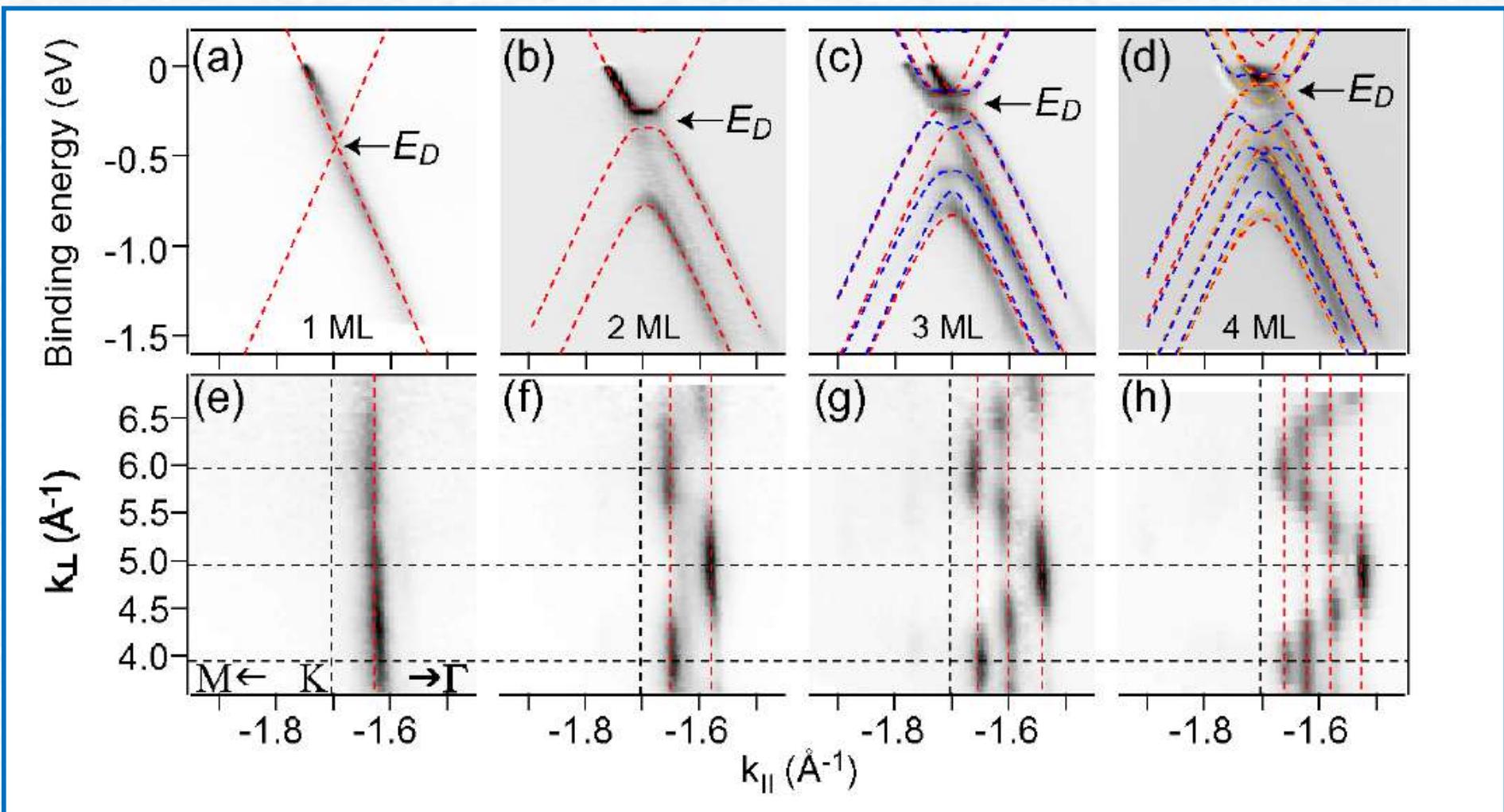
Valence band of graphite (**HOPG**), stacking of the ARPES spectra as a function of **polar angle** (left) and experimental band structure (right).

# Graphene band structure



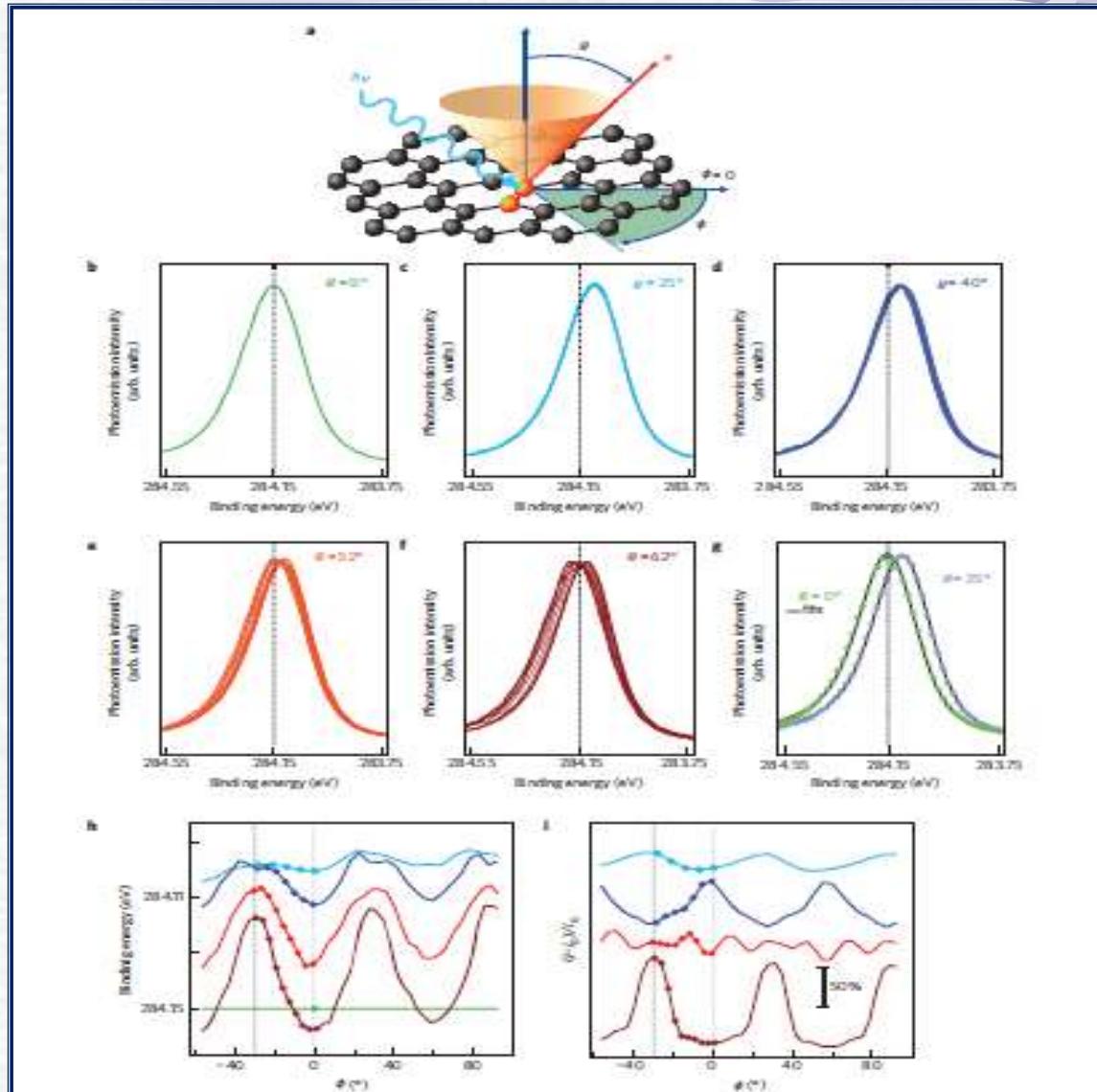
Graphene band structure along  $\Gamma$ KM and zoom of the Dirac cone around the K point of the SBZ. ARPES data taken with high-resolution ARPES and a He discharge source

# Band formation in graphene multilayers



Formation of an electronic band, stepwise: from 1-layer (extreme left) to 4-layer (extreme right) graphene band structure along across the Dirac point.

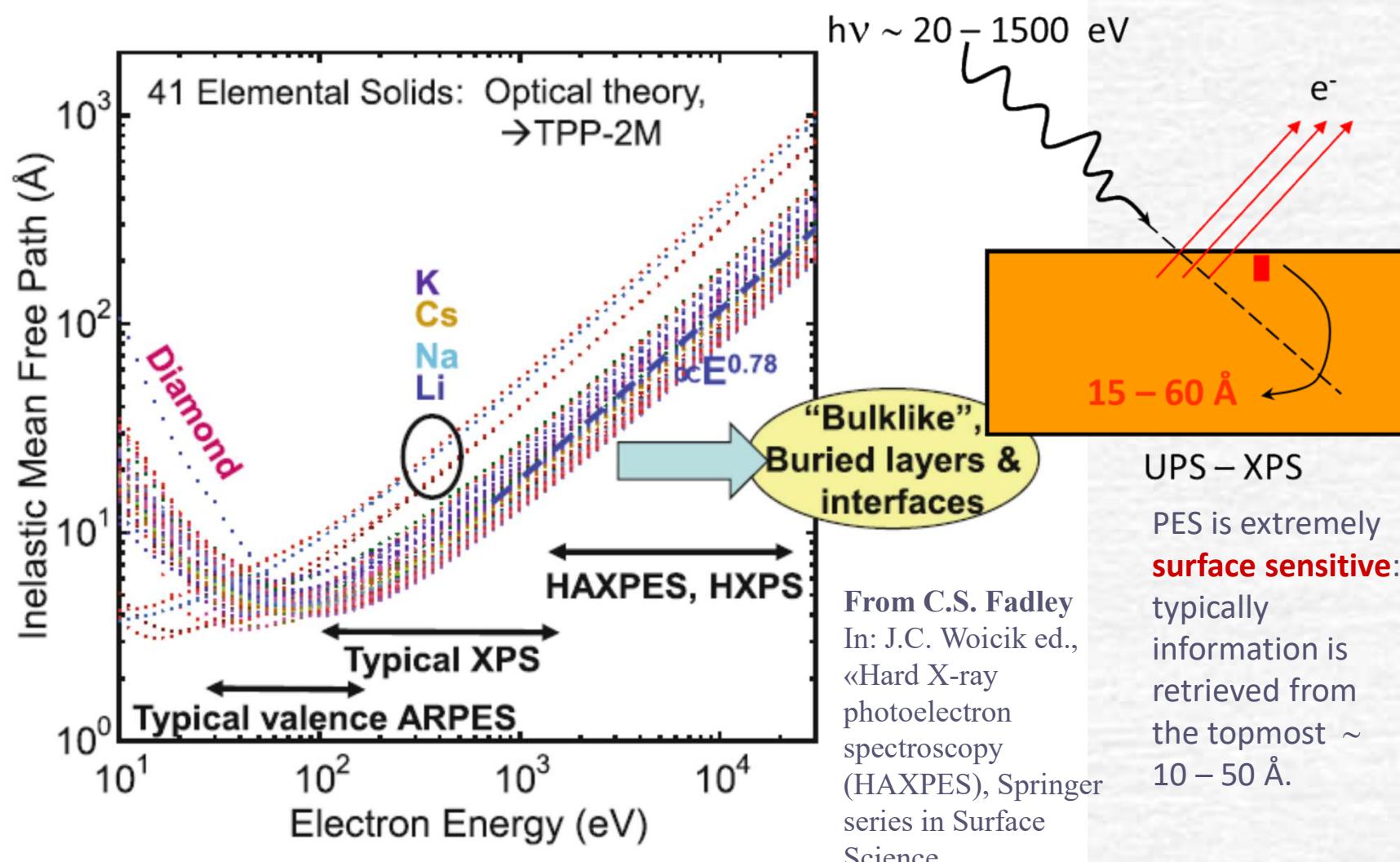
# K shell dispersion in graphene

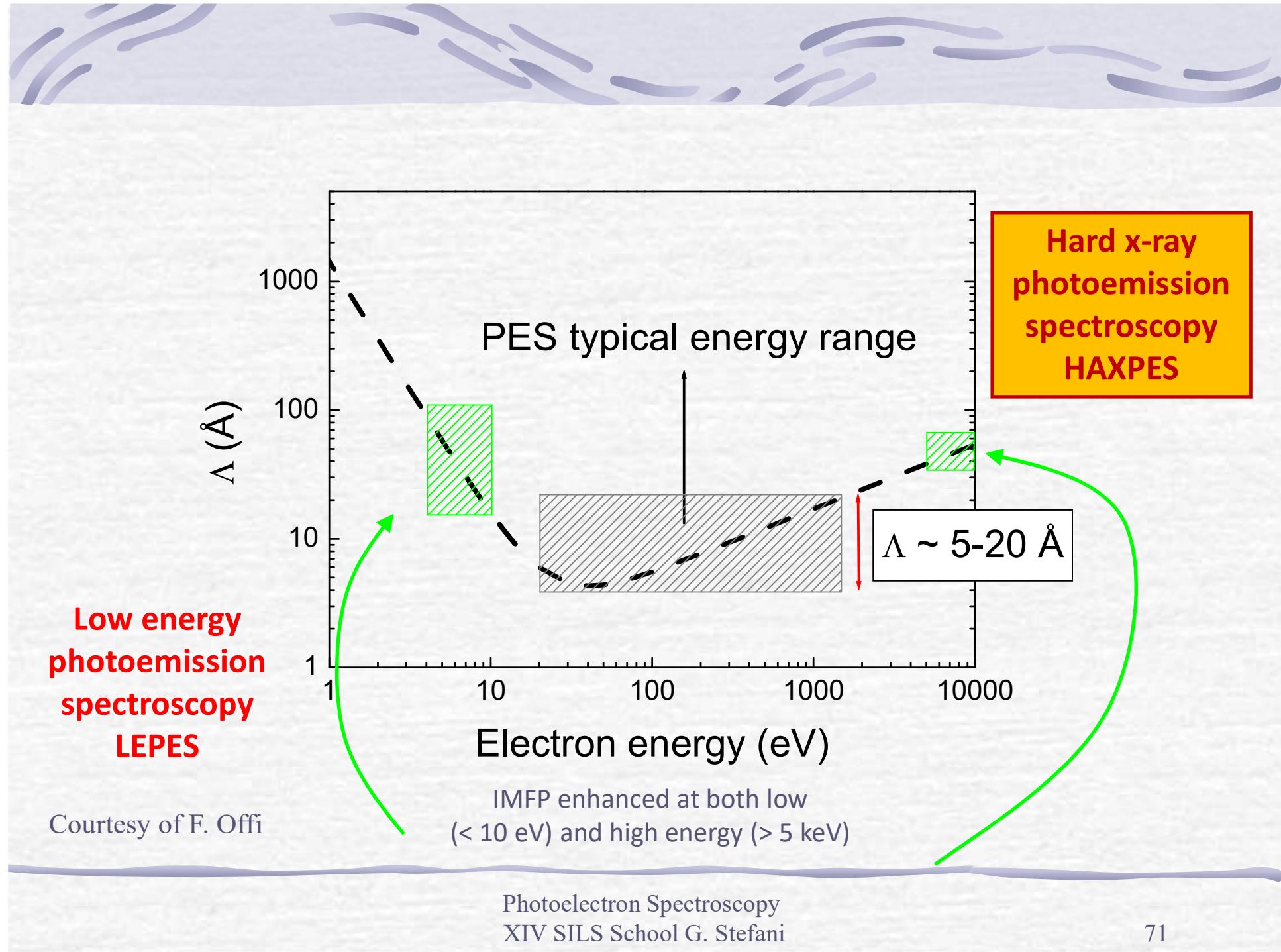


Spectral function of the C 1s core-level in graphene as a function of the emission polar angle

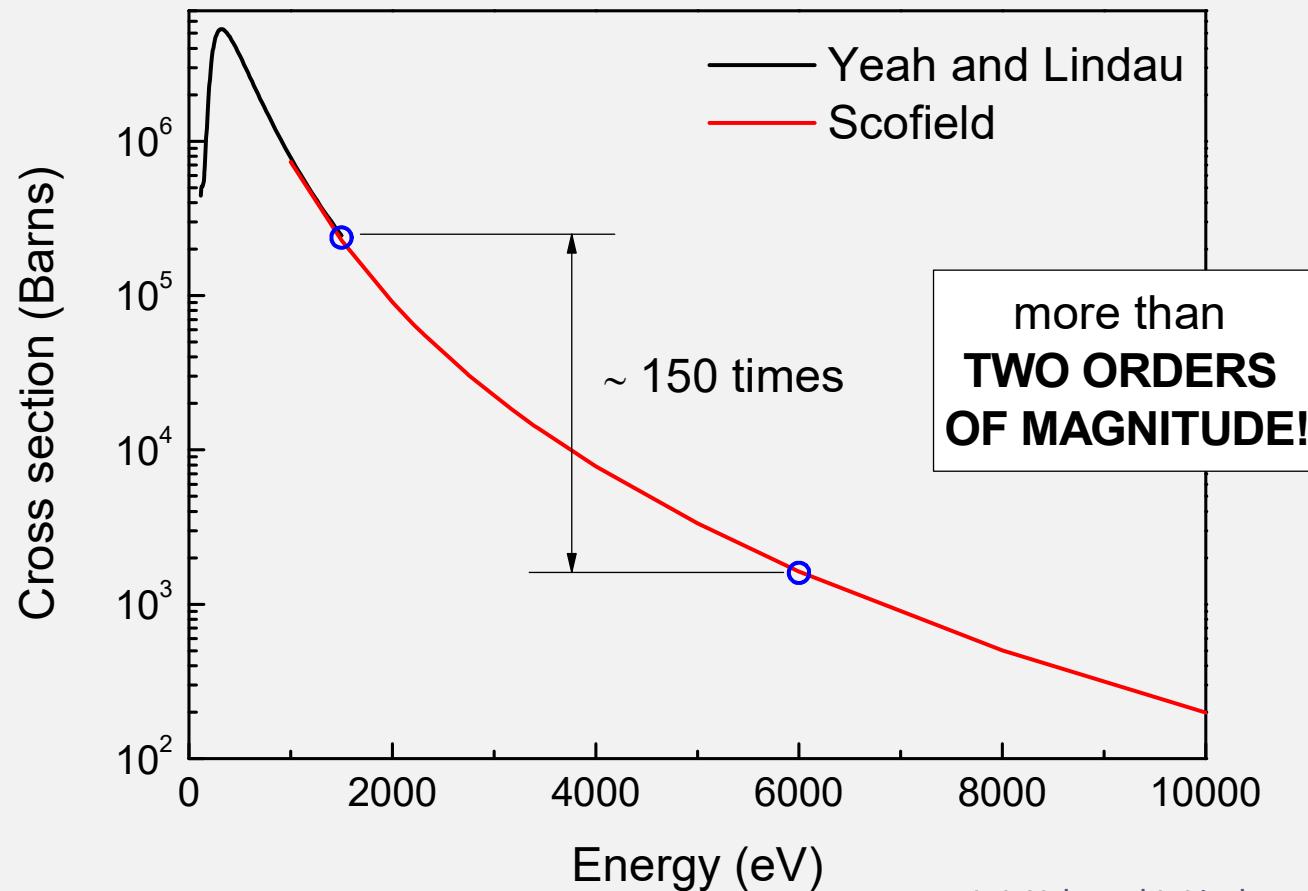
Silvano Lizzit, et al.: Nature Physics **6**, 345-349 (2010)

# Why Hard X-ray PES (HAXPES) ?





# Main difficulty with HAXPES



J. J. Yeh and I. Lindau, At. Data Nucl. Data Tables **32**, 1 (1985)

J. H. Scofield, LLNL Report, UCRL-51326 (1973)

## Early attempts

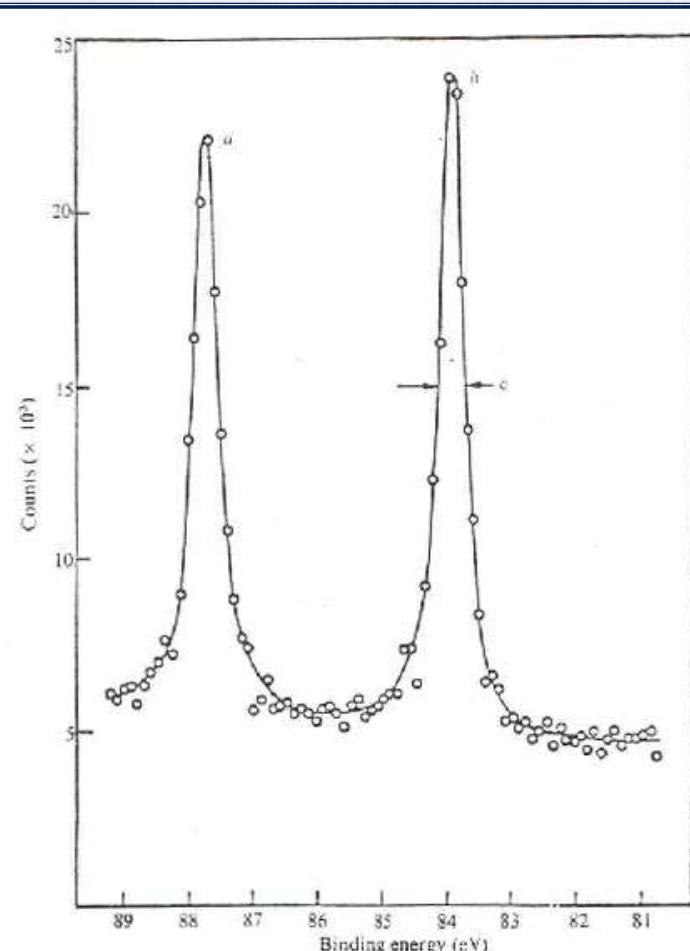


Fig. 2 Electron distribution curve for the 4f doublet of gold. The Fermi level is chosen as the reference level for the binding energy. *a*,  $4f_{5/2}$  peak; *b*,  $4f_{7/2}$  peak; *c*, 0.42 eV (FWHM).

Courtesy of F. Offi

Au 4f core level

**25 counts/s**

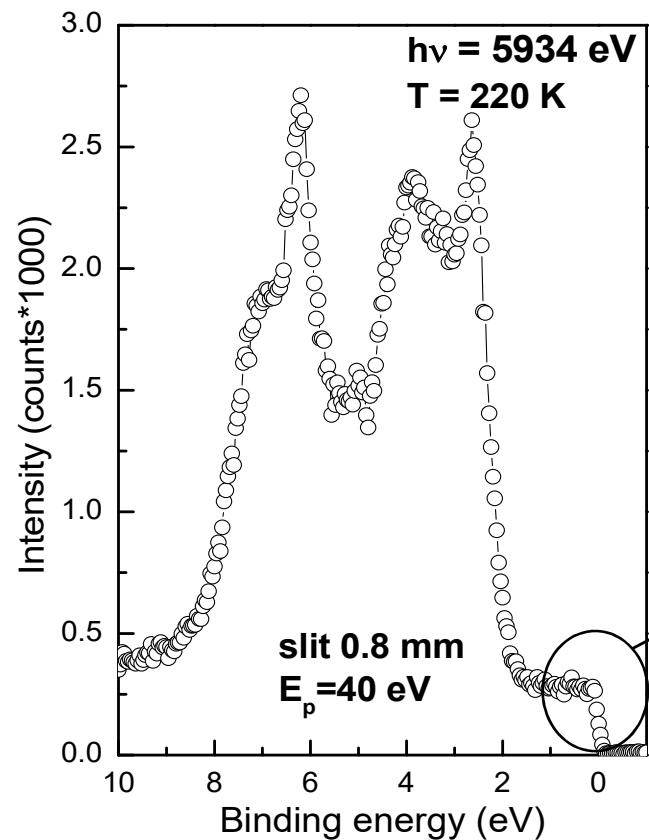
~8 keV Kinetic Energy  
Total Energy Res. 420 meV

**NOT POSSIBLE TO  
MEASURE VALENCE BAND**

I. Lindau, P. Pianetta, S. Doniach, W.E. Spicer  
Nature **250**, 214 (1974)

# RECENT RESULTS ON VALENCE BAND

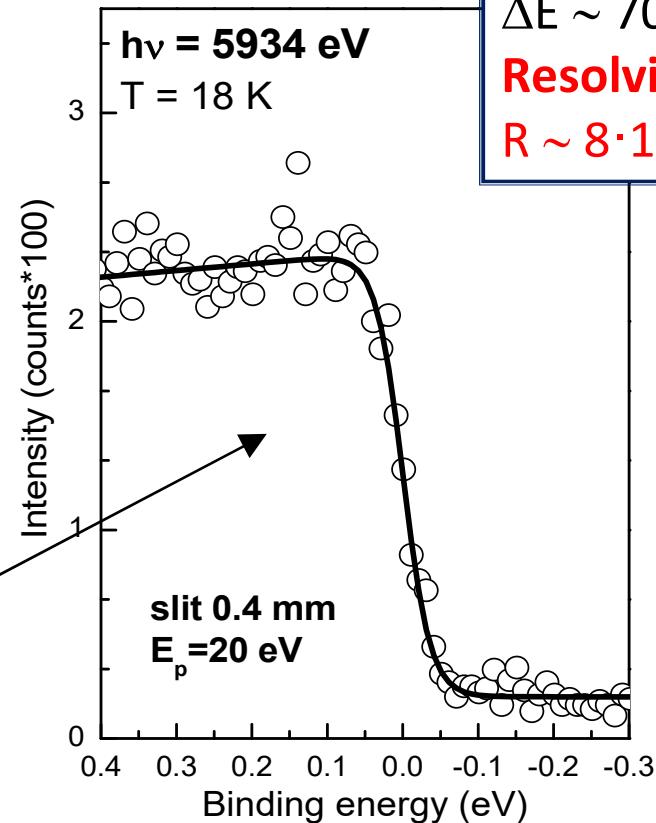
Au polycrystalline



Energy resolution

$\Delta E \sim 70 \text{ meV}$

Resolving power  
 $R \sim 8 \cdot 10^4$



G. Paolicelli *et al.*, Journal of Electron Spectroscopy and Related Phenomena **144-147**, 963 (2005)

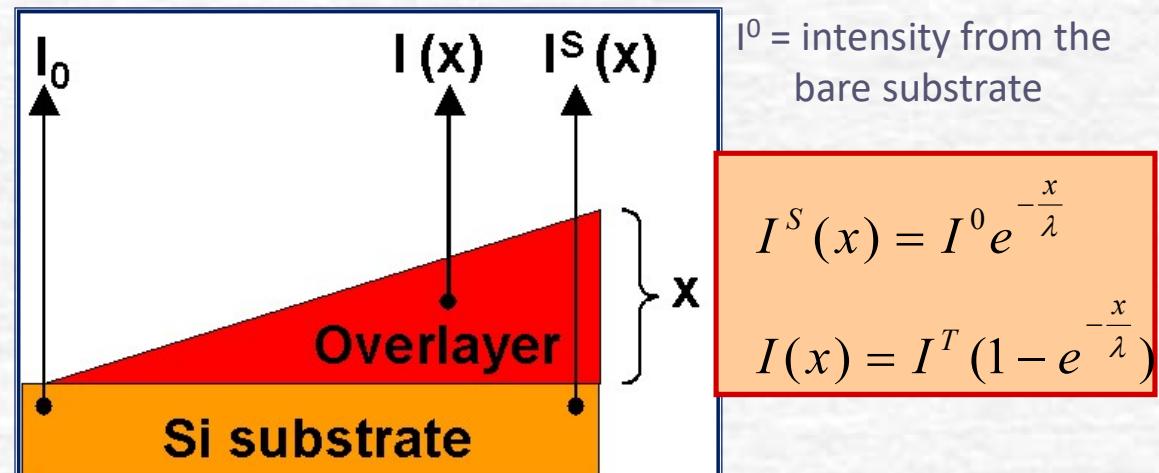
Courtesy of F. Offi

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# Estimation of the bulk sensitivity

Bulk sensitivity in PES is typically estimated via measurements of electron effective attenuation length ( $\lambda$ ), the thickness of the overlayer that reduces to  $1/e$  the intensity  $I^S$  of a core level emission from the substrate.

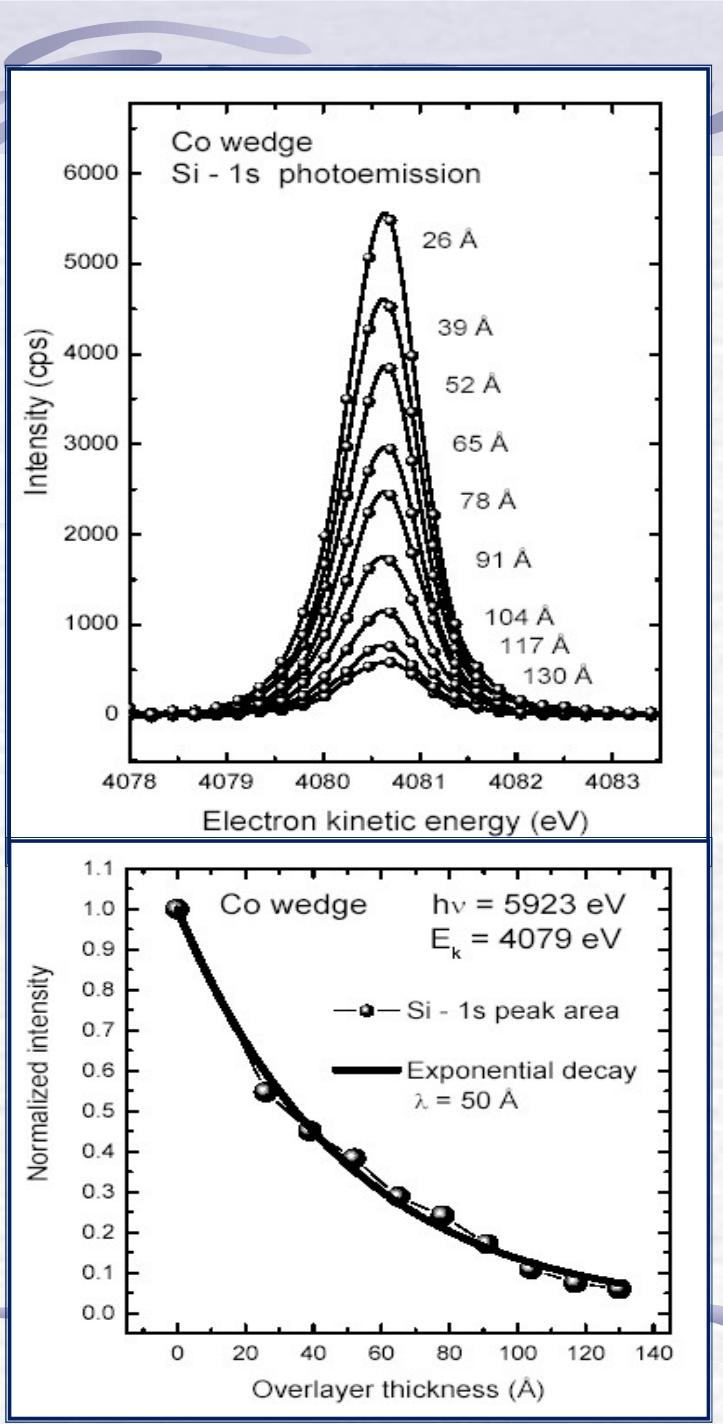
The “overlayer” experiment:



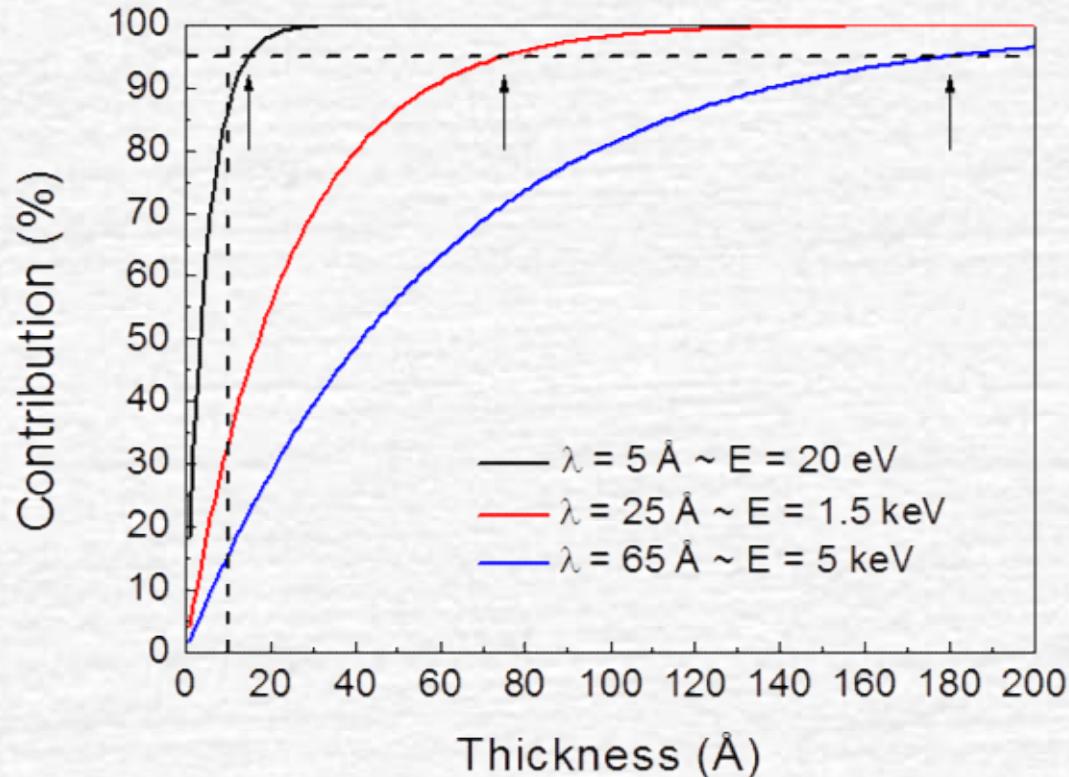
M. Sacchi *et al.*, Phys. Rev. B **71**, 155177 (2005)

Courtesy of F. Offi

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# Contribution of the surface layer



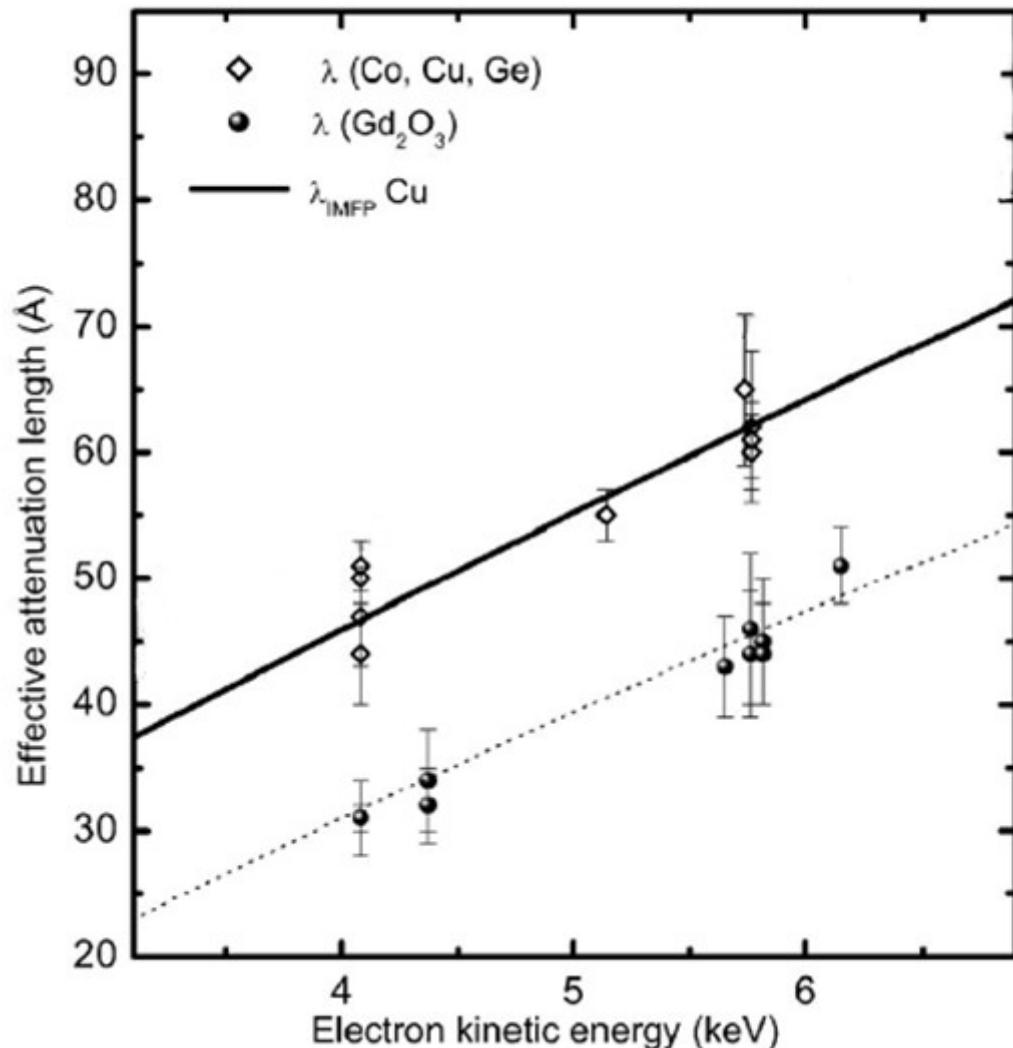
Courtesy of F. Offi

$$I(x) = e^{-\frac{x}{\lambda}}$$

**Information depth**  
layer thickness from  
which 95% of the total  
signal is produced

| E (eV) | EAL (Å) | Inf. Depth (Å) |
|--------|---------|----------------|
| 20     | 5 Å     | 15 Å           |
| 1500   | 25 Å    | 75 Å           |
| 6000   | 65 Å    | 195 Å          |

# EAL in different materials

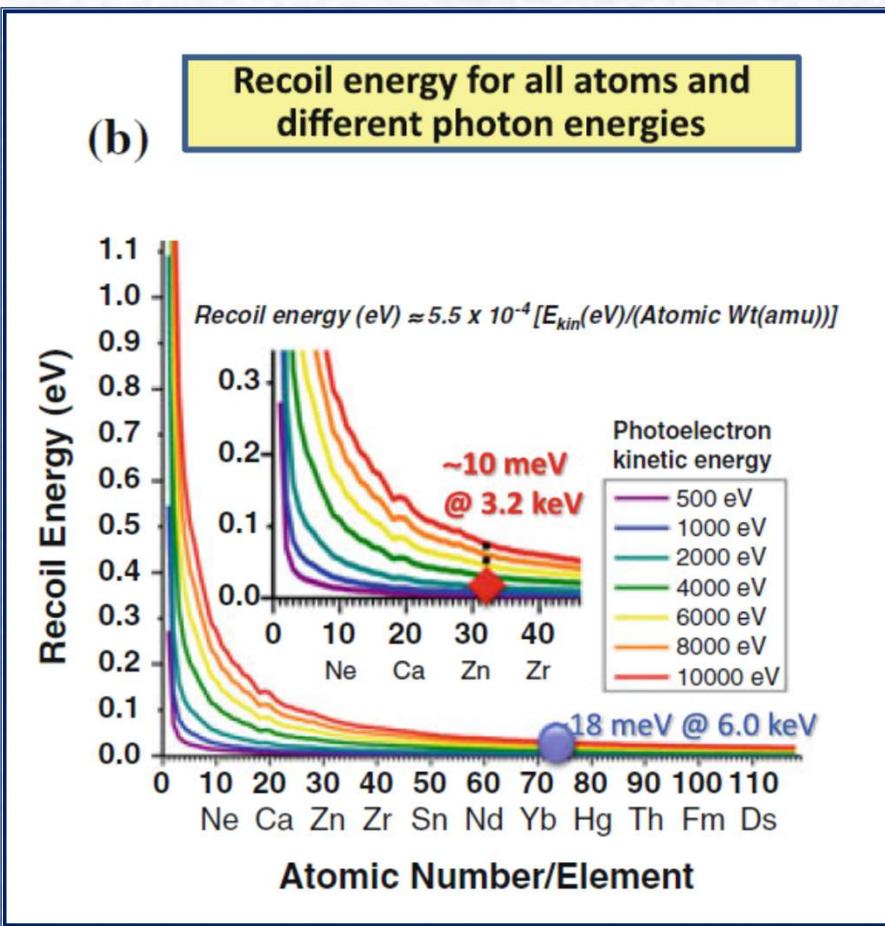


Courtesy of F. Offi

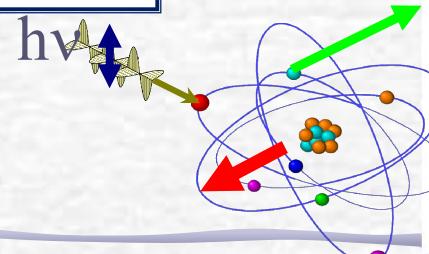
For most materials the experimental values of EAL are quite **higher than what obtained at lower photon energies** and quite in **agreement with theoretical models**

M. Sacchi *et al.*, Phys. Rev. B **71**, 155177 (2005)

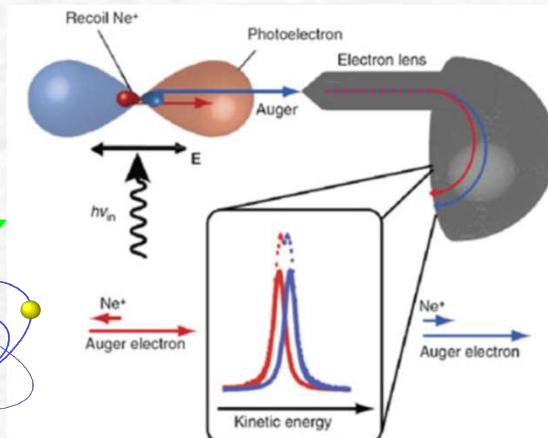
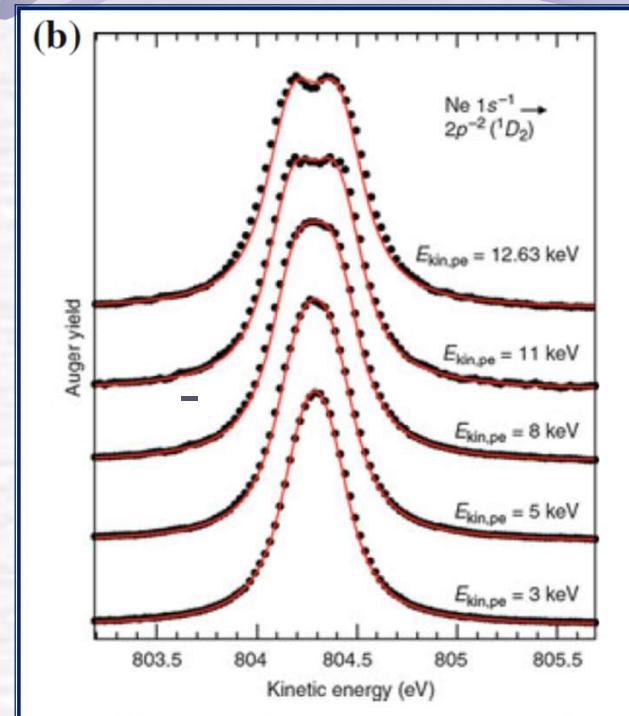
# The non negligible recoiling nucleus



From C.S. Fadley  
In: J.C. Woicik ed.,  
«Hard X-ray  
photoelectron  
spectroscopy  
(HAXPES), Springer  
series in Surface  
Science



Photoelectron Spectroscopy  
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# The End

# References

- ❖ C.S. Fadley “Basic Concepts of X-ray Photoelectron Spectroscopy”, in Electron Spectroscopy, theory, techniques and applications, Brundle and Baker Eds. (Pergamon Press, 1978) Vol. 11, ch.1  
available at: <HTTP://WWW.PHYSICS.UCDAVIS.EDU/FADLEYGROUP>
- ❖ S. Hufner “Photoelectron Spectroscopy, principle and applications” (Berlin Springer 2003) 3<sup>rd</sup> Edition
- ❖ V. Schmidt “Photoionization of atoms using synchrotron radiation” Report on Progress in Physics 55(1992)1482
- ❖ C.M. Bertoni in “Synchrotron Radiation Basics, Methods and Applications (Springer Verlag Berlin Heidelberg 2015, pg. 145)
- ❖ C. Mariani and G. Stefani in “Synchrotron Radiation Basics, Methods and Applications (Springer Verlag Berlin Heidelberg 2015, pg. 275)
- ❖ J.C. Woicik ed., «Hard X-ray photoelectron spectroscopy (HAXPES), Springer series in Surface Science